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MATERIAL DEVELOPMENT STUDY FOR A HAZARDOUS CHEMICAL PROTECTIVE CLOTHING OUTFIT

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TABLE OF CONTENTS

		Page No
	EXECUTIVE SUMMARY	1
1.0	INTRODUCTION	3
2.0	STUDY SUMMARY	5
	2.1 Levels of Protection - Task I 2.2 Requirements - Task II 2.3 Testing Program (Plan) - Task III 2.4 Interim Report - Task IV 2.5 Material Compatibility Testing - Task V	5 17 25 33 33
3.0	EXPERIMENTAL	41
	3.1 Chemical Sources 3.2 Swatch Test 3.3 Penetration Test 3.4 Single Decon Tests 3.5 Multi-Decon Test 3.6 Concentration vs Time Tests 3.7 Seam Test 3.8 Summary of Analytical Methods	41 41 44 49 50 51 51
4.0	DISCUSSION OF RESULTS	55
5.0	RECOMMENDATIONS	61
	5.1 General Comments 5.2 Future R&D Effort (Task VI)	61 61
	BIBLIOGRAPHY	67
	APPENDIX A - LEVELS OF PROTECTION	A-1
	APPENDIX B - MANUFACTURERS AND SUPPLIERS OF PROTECTIVE EQUIPMENT	B-1
	APPENDIY C - MATERIAL COMPATIBILITY TEST DATA (Task V)	C − 1
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LIST OF ILLUSTRATIONS

Figure	No.	Page No.
1	Butyl Rubber Performance Rating Evaluation Plan	26
2	Polycarbonate Performance Rating Evaluation Plan	27
3	Material Compatibility Testing Sequence	31
4	Swatch Test Data Breakdown	34
5	Breakdown of Penetration/Decon Tests- Butyl Rubber	36
6	Swatch Test Specimens	43
7	Penetration Cell	45
8	Penetration Test Panel Schematic	46
9	Modifications of Penetration Cell for Gases and Contaminant Free Service	48
	LIST OF TABLES	
able 1	No.	Page No.
1	Guidelines for Selection of Personal Protection Equipment	8
2	Selection of Respirators for Emergency or Short-Term Use on the Basis of Hazard and Expected Concentration(Particulates)	9
3	Protection Factor Formula	10
4	Respiratory Protection Factors	11
5	Selection of Respirators for Emergency or Short-Term Use on the Basis of Hazard and Expected Concentration (Gases and Vapors)	14
6	Pafarral List (Protective Clothing)	16

LIST OF TABLES (continued)

Table No.		Page No
7	Chemicals with Unassigned Levels of Protection	18
8	Pyrophoric Chemical List (Fire Suit)	19
9	Material Requirement Summary	20
10	Decon Cycle Test Results for Butyl Rubber	37
. 11	Chemicals Requiring Alternate Decon Investigation	38
12	Butyl Seam Test Results	40
13	Summary of Analytical Methods	53
14	Chemical Resistance by Class	59
15	Potential Alternative Materials for In- vestigation as per CHRIS Chemical Reactivity Grouping	63
16	Sealed Suit Chemicals for Future	64

EXECUTIVE SUMMARY

The U.S. Coast Guard has a mandate to respond to discharges of hazardous chemicals into or near the waters of the United States. Appropriate protective gear is required for the Strike Team personnel in order to cope with the toxicity and other hazardous properties of these discharges.

The Chemical Hazards Response Information System Manual (CHRIS) lists 985 chemicals which might be discharged into inland or coastal waterways. The manual provides the On-Scene Coordinator or other officials with some of the information necessary to properly respond to discharges of hazardous chemicals. The primary objectives of this program were to determine the levels of protection required for each of the chemicals and to develop guidelines for the personal protective equipment which might be required for Strike Team personnel.

It was obvious, at the inception of the program, that levels of protection could not be assigned for many of the chemicals due to lack of toxicological information. Further, it was determined that compatibility data for materials commonly used in protective ensembles, specifically butyl rubber and polycarbonate, did not exist.

Levels of protection were established by reviewing each of the 985 chemicals on the CHRIS list, referring to current information. Clothing requirements and respiratory protection were determined for each case where possible. Many of these required judgments based or good practice, chemical toxicity, expected exposure conquions, etc. Toxicity data was not available for 45 chemicals. For these, a level of protection was not established and they were placed into a future R&D category.

Protective clothing ensembles are available commercially, and the Coast Guard has acquired a Hazardous Chemical Protective Clothing Outfit (HCPCO). However, the levels of protection required for specific chemicals given in the CHRIS list have not been addressed in a correlated fashion. The ideal material which is totally compatible with all of the chemicals is not available - none of the materials used in manufacture of protective equipment has such compatibility. Requirements for use of a fully sealed

suit include: maximum wear time of two hours; 20 hours resistance to chemical damage (represents 10 wear cycles and decontaminaton); 10 year shelf life; ambient temperature of -30°C to 40°C; no UV or ozone degradation; be impermeable to 100% chemical contact. Materials of suit construction should also be relatively inexpensive and easy to fabricate. Decontamination should not degrade future performance.

Studies of information on suit materials compatibility of the 985 chemicals on the CHRIS list were confined to butyl rubber and polycarbonate, since these are the materials in the existing HCPCO ensemble. Of these chemicals, 403 were considered to require a fully sealed suit: 45 chemicals showed insufficient data for a decision and 15 chemicals are pyrophoric. Of the 403 chemicals requiring a sealed suit, it was found that sufficient data was not available on 113 chemicals for the polycarbonate chemical resistance, and 377 for the butyl rubber.

A materials compatibility program to demonstrate physical damage and penetration of materials by chemicals in question was performed. Decontamination effectiveness and degradation of protection by repeated decontamination were studied. In these tests, any detectable permeation of the test chemical through the butyl rubber on a continuous 20 hour penetration test was cause for failure. Additionally, failure during 10 contamination/decontamination cycles was cause for rejection. Seams were also studied and found to be difficult to decontaminate, but were basically as resistant as the parent butyl suit material.

As a result of this program and the best information available in the literature, polycarbonate was found to be compatible with 241 chemicals and incompatible with 119 chemicals; no assignment could be made for the remaining 43 chemicals. Butyl rubber was found to be compatible with 155 chemicals and incompatible with 187 chemicals; no assignment could be made for the remaining 61 chemicals.

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MATERIAL DEVELOPMENT STUDY FOR A HAZARDOUS CHEMICAL PROTECTIVE CLOTHING OUTFIT

1.0 INTRODUCTION

This report summarizes a study performed for the U.S. Coast Guard to broaden the data base on protective clothing required for Strike Team personnel. These personnel must deal with accidental discharges of hazardous chemicals into or near waters of the United States and various degrees of protection must be provided depending upon the specific chemical spill.

Work centered around the evaluation of butyl rubber and polycarbonate materials used in the Hazardous Chemical Protective Clothing Outfit (HCPCO), a sealed suit ensemble. Although some chemical compatibility data for butyl rubber and polycarbonate exists in the literature, data is limited on 985 chemicals of concern in regard to chemical compatibility and the degree of protection afforded. Chemical compatibility data in the literature applied primarily to engineering systems information, e.g., gasketing material, O-rings and so on, and translating it in the context of personnel protection is a different matter. This is because data on permeation and ability to decontaminate are even less known.

The project proceeded by completing succeeding task assignments. These were as follows:

Task I - Levels of Protection. The objective of this task was to provide information on the type of personnel protection that will be necessary for each of the 985 hazardous chemicals in CHRIS. Potential sources for personnel protection outfits were to be provided. The output of this task provided input into Task V (Material Compatibility Testing) and Task VI (Future R&D Effort). An Interim Report was prepared on Task I results.

Task II - Define Requirements. The objective of this task was to establish parameters which must be met for candidate materials so that they are suitable for use in a hazardous personnel protective outfit. Part of the task was to provide candidate materials and selection rationale. This task served as a basis for evaluation of candidate materials and, in this respect, was supplementary to the development of the test program plan.

Task III - Testing Program. The objective of this task was to devise a test plan based upon information generated in Tasks I

- and II. Task V, Material Compatibility of Testing, was conducted according to the Task III test program plan.
- Task IV Interim Report. This task summarized the results of Task II (Define Requirements) and Task III (Testing Program). Approval of the Task IV Interim Report was required before proceeding with Task V, Material Compatibility Testing.
- Task V Material Compatibility Testing. The preceding task provided the guidelines for conducting Task V. The objective of Task V was to collect quantitative data on chemical resistance of butyl rubber and polycarbonate used in the HCPCO suit. Output of this task is factored into Task VI, Future R&D Efforts.
- Task VI Future R&D Efforts. The objective this task is to recommend alternatives to butyl rubber and polycarbonate used in the HCPCO which do not provide suitable protection with some chemicals. Additional studies needed to further define levels of protection and to recommend protective garments are suggested as part of this task.
- Task VII Final Report. This task summarizes the work conducted in Task I through Task VI.

2.0 STUDY SUMMARY

The results of Task I through Task IV have been presented in Interim Reports. Task I, Levels of Protection, was given in Interim Report MSAR 79-36. Task II, Define Requirements, was presented in Interim Report MSAR 79-45. Task III, Testing Program, was given in Interim Report MSAR 79-52. Task IV was a report summarizing Tasks II and III. Task VII, this report, constitutes primarily work conducted in Task V, Material Compatibility Testing, and Task VI, Future R&D Efforts, as well as summarizing work performed in the other tasks.

2.1 Levels of Protection - Task I

The Coast Guard is responsible for dealing with spills of 985 hazardous chemicals. Safety equipment and levels of protection have been described to some extent in CHRIS Volumes I, II, (1978). The basic objective of this task was to determine the degree of protection required as a function of the physical, chemical and toxicological characteristics of each of the 985 chemicals. Levels of protection along with other available pertinent information are given in Appendix A.

In establishing levels of protection, it was assumed that a worst case condition would exist at the spill site. Members of the strike cleanup team could be exposed to high and unknown air concentrations of the spilled chemical. Leaking rail cars, trucks or barges could result in direct contamination of the skin with large quantities of the chemical. Members of the crew could be required to enter contaminated waterways to erect dikes or dams and thus be exposed directly to high levels of the contaminant. Since exposure levels are incident dependent, and cannot be predicted, maximum levels of protection are recommended for the initial entry to the area by the members of the strike team. Once the severity and associated hazards of the spill have been identified, a lower level of protection could be used.

Basic premises made to evaluate potential hazards posed by hazardous chemical spills to recommend protective clothing and respiratory equipment were:

- Guidelines used to evaluate the hazard should cover all modes of entry into the body and should be based upon acute exposures.
- 2. The recommended protective equipment should be classified into three basic categories to minimize stocking, storing and logistic problems.

In this task, the basic source of information was Sax's ranking of hazard levels ranging from 3, a severe hazard to 0, the least hazard. These values are based on acute exposures and not on chronic exposures. Effects of exposure relate to inhalation, percutaneous (skin absorption) and skin irritation hazards.

An inhalation hazard of 3 and a skin irritant of 3 constituted rationale for use of a sealed suit for any chemical whether liquid, solid, or gaseous state. Where a 3 appears under respiratory hazard, the use of an air line supply or complete respiratory support without recourse to ambient atmosphere at the point of the spill is dictated.

This leaves the middle ground of a 2 categorization where some judgment is required. A 1 category in either contact, percutaneous or respiratory effects can be handled by individual items of protection other than the fully encapsulating sealed suit.

2.1.1 Rationale for Establishing Levels of Protection

There are a number of sources which can provide guidelines for selection of protection levels. References 1-9 given in the Bibliography were the primary references used.

Threshold Limit Values, Short Term Exposure Limits and Excursion Limits have been developed to provide a safe environment for workers who may be exposed to toxic contaminants during a normal routine work cycle. These values are useful in maintaining healthful work environments, and in some cases provide relative toxicity data among various classes of contaminants. Concentrations at or below the recommended TLV and STEL require no protective equipment; in the case of massive hazardous chemical spills, it must be assumed that the ambient air concentration of the contaminant will exceed these values. Thus, TLV, STEL and EL values offer limited information on required levels of protection for concentrations exceeding these values.

Threshold Limit Values define chronic exposure lavels where the worker is exposed to low concentrations for long periods of time. In the case of a hazardous chemical spill, the members of a Strike Team would likely suffer acute exposures of exceptionally high concentrations of hazard chemicals. In addition, the TLV's indicate those chemicals which may find their way into the body through the percutaneous route but they do not necessarily identify those that may be tissue irritants.

These limitations dictated the need for an alternate procedure for establishing levels of protection. Sax's <u>Dangerous Properties of Industrial Materials</u> provides a Toxic

Hazard Rating Code with information on both acute and chronic effects and assigns ratings based on the mode of action or entry, i.e., inhalation, ingestion, skin absorption and irritation.

The rationale used for establishing required levels of protection is based on the hazard rating code of Sax:

O NONE:

(a) No harm under any conditions;

(b) Harmful only under unusual conditions or overwhelming

dosage.

1 SLIGHT:

Causes readily reversible changes which disappear after end of

exposure.

2 MODERATE:

May involve both irreversible and reversible changes not severe enough to cause death or permanent

injury.

3 HIGH:

May cause death or permanent injury after very short exposure to small

quantities.

U UNKNOWN:

No information on humans considered

valid by authors.

These are applied to the routes of entry to the body, and the resultant effect such as:

Acute Local

Acute Systemic

Irritant Ingestion Inhalation

Ingestion Inhalation

Skin Absorption

Table 1 lists the guidelines used in selecting protective equipment based upon toxicity rating, mode of entry, and physical form of the chemical. If a chemical is a carcinogen, a sealed suit was recommended. If a chemical has a O rating or is innocuous, it is recommended that strike team personnel wear standard safety equipment including hard hats, safety glasses, gloves, and steel-toed footwear in observance of good safety practices.

2.1.1.1 Respiratory Protection

The first concern is respiratory protection, since this is the most likely exposure route, and is generally the most critical. Guides to selection of type and degree of respiratory protection equipment developed are presented in Tables 2,3 and 4.

TABLE 1 - GUIDELINES FOR SELECTION OF PERSONAL PROTECTION EQUIPMENT

	SÞ	Clothing	None required "	Nonsealed Nonsealed	Nonsealed	Nonsealed FES ¹ FES ¹
	nse Solids	Respiratory	None required "	Particulate 	Particulate & chemical filter	Air supply ³
•	Response iquids	Clothing	None required "	 Nonsealed	Nonsealed	Nonsealed FES FES
	Gas & Liquids	Respiratory	None required	Chemical filter	Chemical filter or air supply	Air supply ³
	~	Entry	Inhalation Skin Absorption Irritation	Inhalation Skin Absorption Irritation	Inhalation Skin Absorption Irritation	Inhalation Skin Absorption Irritation
	Toxicity	200	0	-	2	٣
					^	

l Fully encapsulating sealed suit indicates need for air supply 2Contaminants in gaseous form at STP indicates need for air supply 3 Carcinogens indicate need for air supply

TABLE 2 - SELECTION OF RESPIRATORS FOR EMERGENCY OR SHORT-TERM USE ON THE BASIS OF HAZARD AND EXPECTED CONCENTRATION (Particulates)

Toxicity	Expe	ected Concentrati (Dusts, Fum	ons of Particulate es and Mists)	e Matter
	Two to five times TLV	Five to twenty times TLV	Above twenty times TLV	Oxygen defi- cient, emer- gency, highly corrosive
Low	Respirator not usually needed	Filter	Filter or air- line respirator	Where expo- sure is to extremely
Moderate or High (tox- icity no greater than lead)	Filter	Filter or air- line respirator	Air-line or self-contained air or oxygen	corrosive dusts or to dusts in an oxygen defi- cient atmos- phere, a self-
Extremely High (toxicity greater than lead)	Filter or air- line respirator	Air-line res- pirator	Self-contained air or oxygen	contained air or oxygen respirator must be used.

NOTES:

- (1) TLV refers to the Threshold Limit Values for a number of substances published by the American Conference of Governmental Industrial Hygienists (Sections 1, 12).
- (2) See Sections 1 and 2 for a discussion of toxicity ratings and their relation to TLV's.(3) Expected concentrations of particulate matter have been shown only as multiples of the threshold limit values. Where these values are not available, the following concentrations may be used as a guide:

	Mineral Dusts	Other Dusts, Furnes, and Mists
2 to 5 (TLV)	up to 50 mppcf*	Up to 0.5 milligrams per cubic meter
5 to 20 (TLV)	50 to 1000 mppcf*	0.5 to 10 millig ams per cubic meter
Above 20 (TLV)	above 1000 mppcf*	Above 10 milligrams per cubic meter

- (4) When unavoidable conditions necessitate using respirators for longer periods (above 1 hour), use equipment in a higher protective category than shown above.
- (5) Subject to limitations (Table 2.1), hose-type respirators may be used in place of air line.

From Sax, N.I., <u>Dangerous Properties of Industrial Materials</u>, Fifth Edition, p. 47, Van Nostrand Reinhold Company, New York (1979).

^{*} Mppcf = millions of particles per cubic foot.

TABLE 3 - PROTECTION FACTOR FORMULA (12)

 $protection factor (PF) = \frac{ambient air concentration}{concentration inside facepiece or enclosure}$

Protection Factor	Respirator Efficiency (%)	Facepiece Penetration (%)	Selection Guide for Maximum Use Concentration (× TLV)
5	80	20	5
10	90	10	10
20	95	5	20
50	98	2	50
100	99	1	100
200	99.5	0.5	200
500	99.8	0.2	500
1,000	99.9	0.1	1,000
2,000	99.95	0.05	2,000
5,000	99.98	0.02	5,000
10,000	99.99	0.01	10,000

From Patty's <u>Industrial Hygiene and Toxicology</u>, Vol. 1, 3rd Revised Edition, p. 1035, John Wiley & Sons, New York (1978)

TABLE 4 - RESPIRATORY PROTECTION FACTORS (12)

уре	Respirator	Facepiece ^a Pressure	Protection Factor
1.	Air purifying		
	A. Particulate ^c removing		
	Single-use, dust	-	5
	Quarter mask, dust/		5
	Half mask, dust	-	10
	Half or quarter mask, high efficiency		10
	Half or quarter mask, fume*	_	10
	Full facepiece, high efficiency	-	50
	Powered, high efficiency, all enclosures	+	1,000
	Powered, dust or fume, all enclosures	+	- '
	B. Gas and vapor removing		10
	Half mask	-	50
	Full facepiece	_	
11.	Atmosphere supplying		
	A. Supplied air		
	Demand, half mask	-	10
	Demand, full facepiece	-	50
	Hose mask without blower, full facepiece	-	. 50
	Pressure demand, half mask*	+	1,000
	Pressure demand, full facepiece!	+	2,000
	Hose mask with blower, full facepiece	-	, 50
	Continuous flow, half mask*	+	1,000
	Continuous flow, full facepiece!	+	2,000
	Continuous flow, hood, helmet, or suit*	+	2,000
	B. Self contained breathing apparatus (SCBA)		
	Open circuit, demand, full facepiece	-	50"
	Open circuit, pressure demand full facepiece	+	10,000
	Closed circuit, oxygen tank-type, full facepiece	-	50
11.	Combination respirator		
	A. Any combination of air-purifying and atmosphere- supplying respirator	Use minimum p	
	B. Any combination of supplied-air respirator and an SCBA	listed above for operation.	type of mode o

Exception: Combination supplied-air respirators, in pressure demand or other positive pressure mode, with an auxiliary self-contained air supply, and a full facepiece, should use the PF for pressure demand SCBA.

From Patty's Industrial Hygiene and Toxicology, Vol. 1, 3rd Revised Edition, p. 1036, John Wiley & Sons, New Yor, (1978)

[&]quot;The overall protection afforded by a given respirator design (and mode of operation) may be defined in terms of its protection factor (PF). The PF is a measure of the degree of protection afforded by a respirator, defined as the ratio of the concentration of contaminant in the ambient atmosphere to that inside the enclosure (usually inside the facepiece) under conditions of use. Respirators should be selected so that the concentration inhaled by the wearer will not exceed the

appropriate limit. The recommended respirator PFs are selection and use guides, and should only be used when the employer has established a minimal acceptable respirator program as defined in Section 3 of the ANSI Z88.2-, 369 Standard.

- *In addition to facepieces, this includes any type of enclosure or covering of the wearer's breathing zone, such as supplied-air hoods, helmets, or suits.
- *Includes dusts, mists, and fumes only. Does not apply when gases or vapors are absorbed on particulates and may be volatilized or for particulates volatile at room temperature. Example: Coke oven emissions.
- ⁴ Any single-use dust respirator (with or without valve) not specifically tested against a specified contaminant.
- Single-use dust respirators have been tested against asbestos and cotton dust and could be assigned a PF of 10 for these particulates.
- Dust filter refers to a dust respirator approved by the silica dust test and includes media of all types, that is, both nondegradable mechanical-type media and degradable resimmpregnated wool felt or combination wool-synthetic felt media.
- *Fume filter refers to a fume respirator approved by the lead fume test. All types of media are included.
- * High efficiency filter refers to a high efficiency particulate respirator. The filter must be at least 99.97 percent efficient against 0.3 µm DOP to be approved.
- 'To be assigned, based on dust or fume filter efficiency for specific contaminant.
- ¹ For gases and vapors, a PF should be assigned only when published test data indicate that the cartridge or canister has adequate sorbent efficienty and service life for a specific gas or vapor. In addition, the PF should not be applied in gas or vapor concentrations that are (1) immediately dangerous to life, (2) above the lower explosive limit, and (3) cause eye irritation when using a half mask.
- *A positive pressure supplied-air respirator equipped with a half-mask facepiece may not be as stable on the face as a full facepiece. Therefore, the PF recommended is half that for a similar device equipped with a full facepiece.
- ¹A positive pressure supplied-air respirator equipped with a full facepiece provides eye protection but is not approved for use in atmospheres immediately dangerous to life. It is recognized that the facepiece leakage, when a positive pressure is maintained, should be the same as an SCBA operated in the positive pressure mode. However, to emphasize that it basically is not for emergency use, the PF is limited to 2000.
- The design of the supplied-air hood, suit, or helmet (with a minimum of 170 liters/min of air) may determine its overall efficiency and protection. For example, when working with the arms over the head, some hoods draw the contaminant into the hood breathing at the This may be overcome by wearing a short hood under a coat or overalls. Other limitations specified by the approval agency must be considered before using in certain types of atmospheres.
- *The SCBA operated in the positive pressure mode has been tested on a selected 31-man panel and the facepiece leakage recorded as <0.01 percent penetration. Therefore, a PF of 10,000+ is recommended. At this time, the lower limit of detection 0.01 percent does not warrant listing a higher number. A positive pressure SCBA for an unknown concentration is recommended. This is consistent with the 10,000+ that is listed. It is essential to have an emergency device for use in unknown concentrations. A combination supplied-air respirator in pressure demand or other positive pressure mode, with auxiliary self-contained air supply, is also recommended for use in unknown concentrations of contaminants immediately dangerous to life. Other limitations, such as skin absorption of HCN or tritium, must be considered.

Guidelines for selection of respirators for emergency or shortterm use have been based on relative toxicity factors and expected concentrations (Table 5). Protection factors have been developed based on respiratory cartridge efficiency, ambient air concentration and facepiece characteristics (Tables 3-4).

Numerous problems exist in formulating recommendations for Strike Team personnel exposed to gaseous and liquid chemical spills. These include:

- 1. The logistics problems involved in storing and maintaining a variety of respiratory canisters and facepiece ensembles.
- The unknown and perhaps variable air concentrations existing in the area of the spill.
- 3. Lack of information on specific canisters which can be used for many of the 985 chemicals.
- Variable canister lifetimes depending upon the type and concentration of chemical.

Respiratory protection against solid, nonv atile chemical spills is more easily defined and requires basically a high efficiency particulate filter cartridge; solids with moderate vapor pressures could require a combination particulate-chemical canister.

2.1.1.2 Protective Clothing

Types of protective clothing classifications can be considered to be:

- Whole body fully encapsulating sealed suits
- 2. Whole body nonsealed suits
- 3. Partial body clothing items

An example of a whole body encapsulating sealed suit is the Hazardous Chemical Protective Clothing Outfit, HCPCO (POTMC Army designation). This suit is made of butyl rubber with a polycarbonate fishbowl helmet. This suit provides full body protection for dealing with those chemicals which require whole body protection. Because it is fully sealed, an environmental control unit is used with the suit to provide the wearer with respiratory air and cooling.

TABLE 5 - SELECTION OF RESPIRATORS FOR EMERGENCY OR SHORT-TERM USE ON THE BASIS OF HAZARD AND EXPECTED CONCENTRATION (Gases and Vapors)

Toxicity	Expec	ted Concentration	ons of Gases or V	apors
	Two to five times TLV or up to 1000 ppm	Five to ten times TLY or 1000-5000 ppm	Above ten times TLV or 5000-20,000 ppm	Oxygen defi- ciency, emer- gency or above 20,000 ppm
Low	No respirator, or chemical cartridge needed	Canister gas mask	Canister gas mask or air- line respirator	Self-contained air or oxygen
Moderate	Chemical car- tridge	Canister gas mask or air- line respirator	Air-line or self-contained air or oxygen	Self-contained air or oxygen
High	Canister gas mask	Air-line res- pirator	Self-contained air or oxygen	Self-contained air or oxygen

NOTES

- (1) TLV refers to the Threshold Limit Values for a number of substances published by the American Conference of Governmental Industrial Hygienists (see Section 1 and section 12).
- (2) See Sections 1 and 2 for a discussion of toxicity ratings and their relation to TLV.
- (3) When unavoidable conditions necessitate using respirators for longer periods (above 1 hour), use equipment in a higher protective category than shown above.

(4) Subject to limitiations (Table 2.1), hose-type respirators may be used in place of air line.

From Sax, N.I., <u>Dangerous Properties of Industrial Materials</u>, Fifth Edition, p. 47, Van Nostrand Reinhold Company, New York (1979).

Part of this task effort was to list those chemicals which require a whole body encapsulating sealed suit (HCPCO). A majority of the chemicals will require protective clothing less sophisticated than the HCPCO type suit and in some cases no special protective clothing at all.

Advisement of types of protective clothing available other than whole body encapsulating sealed suits was a part of Task I work goals.

Protective clothing manufacturers were contacted to obtain clothing and chemical compatibility data. A list of organizations surveyed is given in Table 6. Appendix B lists the specific types of protective devices available from suppliers.

Some chemical compatibility information is available in protective clothing equipment brochures but is limited in the number of chemicals. Chemical compatibility data for clothing materials is presented only as a guideline and the manufacturer does not guarantee any specific chemical compatibility/clothing performance. Protective clothing manufacturers will provide sample swatches and advise that those interested in their protective clothing products conduct testing and evaluation themselves. Complete compatibility data on all 985 chemicals listed in CHRIS was not obtainable from manufacturers data sources and therefore testing was required in many cases.

Types of protective clothing items include gloves, boots, aprons, hoods, coveralls, pants, jackets, coats, one and two piece suit combinations as well as whole body encapsulating sealed suits. The majority of protective clothing is polyvinyl chloride coated on nylon or cotton fabric material. Rubber and urethane fatric materials are also available. TYVEK (duPont trademark) is a lightweight polyethylene fabric material which is low cost and generally used for throw-away type protective apparel. Butyl rubber protective clothing garments are top of line in cost. Unfavorable cost and manufacturing problems have detracted from Teflon as a protective clothing material although it would be the most chemical resistant of all materials.

2.1.2 Summary and onclusions

The 985 chemicals were reviewed and levels of protection were established based on Sax's Toxic Hazard Rating. It had been estimated initially that perhaps 100 of the chemicals would require a full encapsulating garment. The rationale used in this study indicated that 403 chemicals may require a fully encapsulating sealed suit with an air supply.

Information is available on the compatibility of many of the chemicals with the standard butyl rubber-polycarbonate

TABLE 6 - REFERRAL LIST

(Protective Clothing)

ORGANIZATION

- 1. Clever Maid Uniform Company
- 2. Singer Safety Products Company
- 3. Boss Manufacturing Company, Ind. Division
- 4. Clark, David, Company, Inc.
- 5. Advance Glove Manufacturing Company
- 6. Star Glove Company
- 7. Shaffer Products
- 8. Texier, H. Glove Company, Inc.
- 9. Colonial Uniform Manufacturing Corporation
- 10. Loveline Industries, Inc.
- 11. Fyrepel Products, Inc.
- 12. Fabohic Inc.
- 13. Plastimayd Corporation
- 14. Wheeler Protective Apparel, Inc.
- 15. Titan Abrasive Systems, Inc.
- 16. Guardian Safety Equipment Company
- 17. Norton Company Safety Products Division
- 18. Encon Manufacturing Company
- 19. Mine Safety Appliances Company
- 20. East Wind Industries
- 21. Army Research Center
- 22. DuPont de Nemours & Company
- 23. Arrowhead Products, Division Federal Mogul
- 24. International Latex Corporation
- 25. Rich Industries (Snyder Manufacturing Company)
- 26. Standard Safety and Equipment Company
- 27. Edmont-Wilson Division, Bechton Dickinson & Company
- 28. JOMAC Products, Inc.
- 29. Plymouth Rubber Company

encapsulating suit. However, the information is not complete and compatibility tests were required for those cases where compatibility was not known.

In summary, of the 985 chemicals reviewed, it was concluded that 403 chemicals would require a fully sealed suit. There were 44 chemicals in which a decision could not be made and these are given in Table 7. Fifteen chemicals are pyrophoric requiring a fire suit; these are listed in Table 8.

Appendix C lists the 403 chemicals in the test program which are discussed later in this report.

2.2 Requirements - Task II

Requirements given herein apply to clothing material sought to provide head and body protection to personnel who will encounter chemicals of all types of reactivity groups.

The list of requirements for candidate materials is given in Table 9.

2.2.1 Service Time

Time period within a fully sealed suit shall be a maximum of two (2) hours based upon the physiological impact of wearing the suit.

First level requirement for candidate material shall be twenty (20) hours chemical resistance time. This requires that the candidate material be exposed for a twenty (20) hour contact time without damage and represents ten (10) wear cycles and associated decontamination.

A secondary protection level requirement for the candidate material shall have a chemical resistance service time of two (2) hours. This represents the absolute minimum wear time for a single use duty during which no catastrophic degradation of material is permissible to jeopardize personnel welfare.

Because of associated cost, a multiple wear cycle material is desired over a single wear cycle material. However, a single wear cycle material should be given consideration as an acceptable alternative.

2.2.2 Shelf Life

randidate material shall have a shelf life period of ten (10) years when properly stored to prevent premature aging. Chemical resistance and durability properties during its shelf life period shall not be affected.

TABLE 7 - CHEMICALS WITH UNASSIGNED LEVELS OF PROTECTION

Code Chemical Name Sta	
AEA n-Aminoethylethanolamine ALS Ammonium Lauryl Sulfate AMT Ammonium Thiocyanate AMY n-Amyl Chloride L	
BZP Benzophenone S BBP Benzyl n-Butyl Phthalate L BZO Benzyl Dimethyl-Octa-Decyl Ammonium Chloride L BMA Benzyltrimethyl Ammonium Chloride L	·
BDO 1,3-Butanediol L BUD 1,4-Butenediol L CPH Camphene S CLS Caprolactum Solution	
CAR Carene CRN p-Chlorotoluene COS Cobalt Sulfamate(ous) CGE Cresyl Glycidyl Ether S	
DES 2,4-D Esters DBZ n-Decylbenzene DIC Dicamba DSD Dodecylsulfate, Diethanolamine Salt	
DSM Dodecylsulfate, Magnesium Salt DDS Dodecylsulfate, Sodium Salt DST Dodecylsulfate, Triethanolamine Salt DAI Dodecyclbenzene Sulfonic Acid, Isopropanolamine Salt S	
EHP Ethoxydihydropyran L EBR Ethylbutyrate L ETN Ethylnitrite L/ IPC Isopropyl Percarbonate S	G
MCP Methylcyclopentane L MPA Monoisopropanolamine L OET Octyl Epoxy Tallate L ORS Oil Misc., Resin L	
ORD Oil Misc., Road ORN Oil Misc., Rosin OSP Oil Misc., Sperm OCF Oil Misc., Clarified	•
PPB Phosphorous, Black SAS Silver Aklyl Sulfates SFL Sulfolane TRN Thorium Nitrate S	
VEE Vinyl Ethyl Ether	L S / G L

TABLE 8 - PYROPHORIC CHEMICAL LIST (Fire Suit)

CHRIS Code	Chemical Name	C 4 - 4
0000	Chemical Name	State
CRE	Calcium Resinate	S
DBR	Decaborane	S
DEZ	Diethyl Zinc	L
DMZ	Dimethyl Zınc	L
EAD	Ethylaluminum Dichloride	L
EAS	Ethylaluminum Sesquichloride	L
NKC	Nickel Carbonyl	L
PTB	Pentaborane .	L
PPR	Phosphorous, Red	S
PPW	Phosphorous, White	S
PTM	Potassium, Metallic	S .
SDU	Sodium, Metallic	S
SZA	Sodium Azide	S
TIA	Triisobutylaluminum	Ĺ
TAL	Triethylaluminum	. <u>L</u>

TABLE 9 - MATERIAL REQUIREMENT SUMMARY

Para- graph	Item	Requirement
2.2.1	Service Time Wear Time Chemical Service Time Primary level No. wear cycles Secondary level No. wear cycles	2 hours 20 hours 10 2 hours 1
2.2.2	Shelf Life	10 years
2.2.3	Ambient Work Conditions Pressure	<pre>1 Atmosphere (Air with no contaminant to 100% contaminant</pre>
	Temperature Humidity Sunlight Fungus	-30°C (-22°F) to 40°C (104°F) Up to 100% and water contact No UV or 03 degradation Will not support mold growth
2.2.4	Temperature Characteristics Operational and storage temperatures, -30°C (-22°F) to 70°C (158°F) range	Noncracking, tackiness, stiffening, flaking or separation
2.2.5	Durability	Meet physical properties per MIL-C-12189E, Table 1
2.2.6	<u>Odor</u>	No offensive odors
2.2.7	Toxicity	No inherent toxicity hazard
2.2.8	Fabrication Methods	Sewed, glued, heat bonded impulse heated, sonic bonded, radio frequency bonded or other method to provide leak tight joints Capable of bonding to other acceptable materials Field and factory repairable

TABLE 9 - MATERIAL REQUIREMENT SUMMARY (cont.)

Para- graph	Item	Requirement
2.2.9	Workmanship	Material conforms to quality and grade per these requirements
2.2.10	Support and Maintenance	Provide data and procedure to sanitize, leak check, seal, repair, inspection, optical quality, storage record keeping
2.2.11	Decontamination	Material does not degrade from decontamination procedures
2.2.12	Material and Production Cost	
	Material Cost Range	\$10-\$50 per square yard Variable depending upon garment material
2.2.13	Chemical Compatibility	
	(985 CHRIS List Chemicals)	100% compatibility desired
2.2.14	Other Requirements	Impermeable Fire retardant Static electricity free
2.2.15	Intended Use	Protective garment

2.2.3 Ambient Work Conditions

Candidate materials shall be capable of exposure to contaminated air with contaminant concentrations ranging from 0 to 100%. Candidate materials must be able to withstand a temperature range of -30°C (-22°F) to 40°C (104°F), a humidity of 100% and water content, have no ultraviolet light or ozone degradation and be fungus resistant.

2.2.4 Temperature Characteristics

Over a temperature range of -30°C (-22°F) to 40°C (104°F) ambient work conditions and storage conditions of -30°C (-22°F) to 70°C (158°F), material shall not exhibit cracking, become tacky, exhibit stiffening, flaking or separation from these temperature extremes.

2.2.5 Durability

Candidate material shall be resistant to abrasion, scuffing and tearing. Physical property data for butyl rubber per MIL-C-12189F, Table 9 shall apply as a material durability requirement.

2.2.6 Odor

Candidate material shall not have an offensive odor.

2.2.7 Toxicity

Candidate material shall not in itself present a toxicity hazard.

2.2.8 Fabrication Methods

Candidate materials shall possess suitable fabrication characteristics. Material for body protective garments shall be sufficiently flexible so as not to impede body movement.

Candidate material shall be capable of being fabricated to produce leak-free seams by sealing, sewing, gluing, heat bonding, impulse heating, sonic bonding, radio-frequency bonding, and other methods. Bonding to other acceptable materials as well as to itself is also desirable.

It is required that material be capable of repair in the field for minor tears or at the fabricator shop for more major tears to salvage the protective ensemble.

2.2.9 Workmanship

Finished material shall conform to the quality and grade of product established by these requirements.

2.2.10 Support and Maintenance

Support and maintenance procedures shall be required for proper care of protective clothing materials. Information shall be provided on:

- 1. sanitizing after use
- 2. leak checking
- 3. leak sealing
- 4. major and minor repairs
- 5. inspection for chemical attack
- 6. checking optical quality of transparent materials
- 7. storage
- 8. record keeping

Material requirements are based upon appropriate care: storage in uncreased condition, prompt decontamination and sanitization, proper inspection and repairs.

2.2.11 Decontamination

Candidate material shall be capable of being decontaminated to recondition for additional service. Decontamination procedures might include, in addition to others, such procedures as:

- 1. Water wash or flush.
- 2. Water wash with detergent.
- 3. Wash with neutralizing agent.
- 4. Wash or wipe with appropriate solvents.
- 5. Brushing or other means of removing contaminants.

Decontamination procedures shall be based upon consideration that contaminants might be solids or liquid residual. Decontamination shall be considered to apply to both external and internal surfaces. Decontamination shall not deteriorate material for additional wear cycles.

2.2.12 Material and Production Costs

Material unit cost and fabrication cost data shall be supplied. Cost level shall be considered for making analysis of

the material's suitability in relation to the Coast Guard's ourposes and use. Material costs in the range of \$3.50 per square yard are typical of protective clothing materials; cost for special runs may be in the range of \$10 per square yard. Material costs generally are minor costs of the total cost; fabrication and quality control costs are the major cost. Material scrap shall be in the range of 7-10% so that material costs do not become a greater portion of total cost. A material cost of \$20 per square yard shall not be considered excessive if service life compared to \$10 per square yard is more than doubled. Based upon material unit cost range of \$3.50 to \$20 per square yard, material costs per ensemble shall be in the range of \$20 to \$120.

It is recognized that fabrication costs will depend primarily upon specific requirements for the suit. The magnitude of the costs will depend upon any special research, development and engineering time to develop a protective suit. Furthermore, protective clothing ensembles manufactured to applicable MIL specifications and to specific design features entail greater cost. Cost targets cannot be established until a final selection is made of garment requirements for each of the chemicals.

2.2.13 Material Chemical Compatibility Data

Data shall be provided on the number and types of chemicals which the candidate material is compatible with so assessment of its applicability can be made. Preliminary evaluation of the candidate material may be judged on basis of chemical reactivity groups, however some exceptions are likely to be found within a reactivity group. Comparison of each of the 985 chemicals given in the CHRIS list against the candidate material shall be made. Data shall consist of the percent compatible with the chemicals and a listing of those which are not and a list for those for which information is not available.

As an example, a data search indicated butyl rubber was compatible with about 70% of the chemicals:

No.	compatible	685
No.	incompatible	119
No.	unknown	111
TOTAL		985

For polycarbonate, the compatibility was approximately 56%:

No. compatible	553
No. incompatible	153
No. unknown	<u>279</u>
TOTAL	985

2.2.14 Other Requirements

Candidate materials shall be impermeable, fire retardant, and not exhibit static electricity properties.

2.2.15 Intended Use

Candidate materials shall be fabricated for assembly of a protective garment. This protective garment shall be used by Coast Guard Strike Team personnel in cleanup operations that include 985 CHRIS list chemicals.

2.3 Testing Program (Plan) - Task III

Levels of protection investigated during Task I indicated 403 chemicals required a sealed suit out of 985 chemicals involved. An additional 15 chemicals were pyrophoric and were assigned to a fire suit. Another 44 chemicals were indeterminate and a level of protection was unassigned. The remaining 523 chemicals were assigned levels of protection of an unsealed suit or use of standard safety equipment; i.e., gloves, goggles, hard hats and steel-toed footwear.

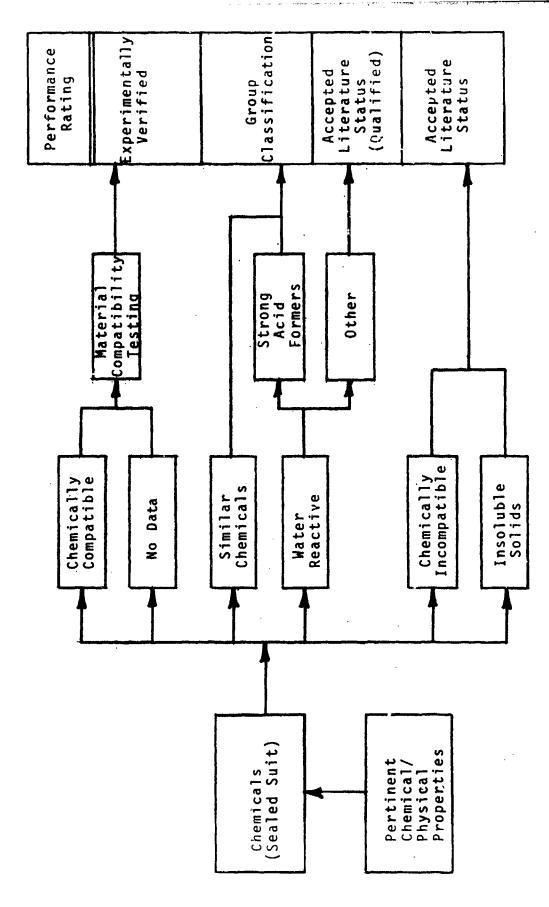
Of the 403 chemicals requiring sealed suits, 327 were specified by use of Sax's Toxic Hazard Rating and the remaining 76 chemicals were determined by other standards and best judgments. A total of 403 chemicals were candidates for the Material Compatibility Testing, Task V.

A program plan was devised with the objective to assign a performance rating for each of the 403 chemicals of interest.

In general, a Performance Rating Guide for chemicals requiring a sealed suit would evolve from the evaluation of available literature compatibility and the outcome of extensive tests conducted at MSAR. There were, however, several overriding factors which prevented the straightforward application of published data. The developmental process used to generate performance ratings is detailed below and is summarized in Figures 1 and 2.

2.3.1 Applicability of Literature Data

Numerous materials selection guides have been published detailing the compatibility of certain plastics and rubbers with various chemicals. Unfortunately, the basis for this information is unclear or not revealed. Realistically, it would have been impractical to ignore this extensive data source in its entirety. As a result, guidelines were developed to distinguish the type data which could be accepted at face value. The following rationale concerning published literature compatibility data was used.



- BUTYL RUBBER PERFORMANCE RATING EVALUATION PLAN FIGURE 1

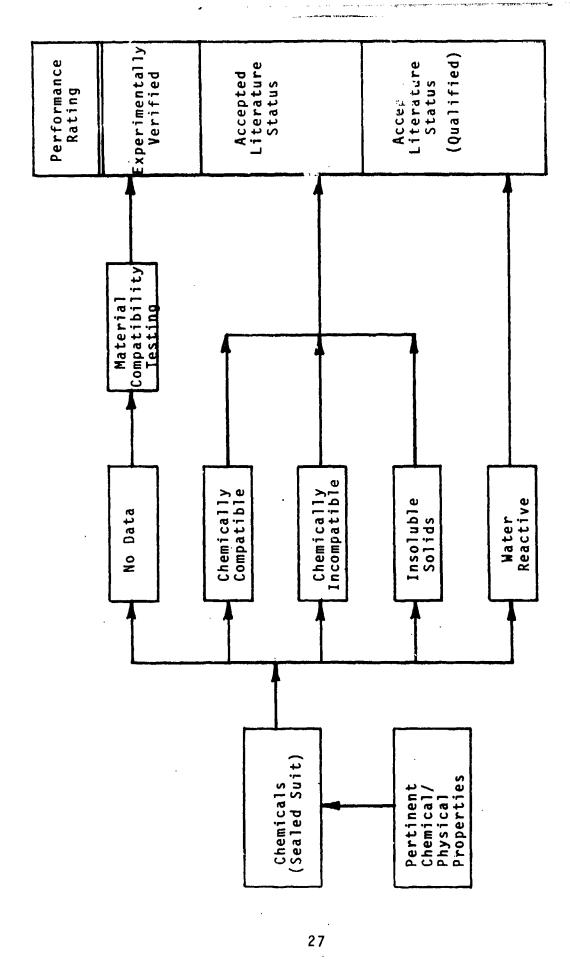


FIGURE 2 - POLYCARBONATE PERFORMANCE RATING EVALUATION PLAN

2.3.1.1 Polycarbonate

The literature findings (compatible or noncompatible) were used without further consideration. It was concluded that valid and applicable type testing have been performed. Key test factors such as resistance to fogging, softening, discoloration, etc., would undoubtedly have been included in these tests. Testing at MSAR was limited therefore only to those chemicals for which no data could be uncovered. Over 100 chemicals fell into "no data" classification and were subsequently tested at MSAR.

2.3.1.2 Butyl Rubber

A literature finding of "incompatible" or "unsatisfactory" of the chemical with butyl rubber was accepted as an unacceptable rating. It was confidently anticipated that the application of more extensive and rigorous testing would only confirm this data. On the other hand, a literature rating of "compatible" was not accepted because of the uncertainty of the test methods and the increased requirements demanded of a protective sealed suit, e.g., penetration data.

In addition to the shortcomings of the published data, there were other factors which were inherent in the chemicals themselves which influences the methods of generating a final rating guide. Three distinct classes of chemicals were singled out and treated separately, i.e., water reactive, insoluble solids and similar group classifications.

2.3.2 Water Reactive Chemicals

It was concluded that literature information concerning chemicals that could react with water applied only to the unreacted chemical, whereas mixtures of uncertain composition would exist in the field. Detailed laboratory studies of these chemicals were not undertaken partly because the field composition would be situation dependent. Performance Rating Guides for water reactive chemicals were generated in the following manner.

2.3.2.1 Polycarbonate

The accepted chemical compatibility literature findings were retained, however the rating would be qualified by noting that it applied only to the unreacted chemical.

2.3.2.2 Butyl Rubber

Specialized laboratory tests eventually showed that reactive chemicals which yielded strong acid decomposition products would eventually prove harmful to butyl rubber. This class of reactive chemicals was categorically given a "Fail" rating.

The literature data was retained for chemicals yielding lesser reactive materials. In this latter case, the rating was qualified and limited to the unreacted chemical.

The subject of water reactive chemicals and their effects on butyl rubber is detailed in a later section.

2.3.3 Insoluble Solids

No tests were scheduled on solid chemicals which were found to be insoluble in water. It was concluded that this general class of chemical could only penetrate the protective ensemble via mechanical defects such as holes, tears, faulty seams, etc. However, the possibility persisted that these chemicals could still chemically attack butyl and polycarbonate. The literature findings were subsequently adopted with no qualifications.

2.3.4 Similar/Group Classification

The overall plan provided for the classification of certain chemicals based on their close chemical relationship. In the event that chemicals could not be evaluated either experimentally or on the basis of literature data, such chemicals should be assigned a rating based on chemical similarity. The number of chemicals assigned a rating on this basis would be held to a minimum and the rating would only be assigned where proven or obvious trends existed.

2.3.5 Material Test Sequence

The overall evaluation plans shown in Figure 1 and 2 provide for a laboratory study to assess the compatibility of the chemical with butyl rubber and polycarbonate. The laboratory test program was limited to chemicals whose assigned literature rating was unknown and/or unaccepted. A Performance Rating Guide for these chemicals was generated by subjecting butyl rubber and polycarbonate to the chemicals in a series of selected and related tests. The laboratory test sequence is shown in Figure 3.

The selected time frames for these tests were 3 hours and 20 hours. A 3-hour time limit was intended to safely simulate a single mission in which the duration of exposure would be limited to about 2 hours. The 20-hour time frame for testing evolved from the consideration that the protective ensemble would be used for at least ten 2-hour periods. In this case, a 20-hour continuous exposure to the chemical was comparable or safely in excess of the demand posed by multi-short term use. The 20-hour test was not rigorously equated to 10 missions although this is implied. A brief description of the individual tests is described next. A detailed discussion of the experimental program is included in Section 3.0.

2.3.5.1 Swatch Test

All chemicals selected to be tested were routed initially through a basic compatibility test. The intended purpose was to immediately eliminate chemicals which grossly attacked the test specimens. The duration of the test was 20 hours in which the specimens of polycarbonate and butyl rubber were maintained completely submerged (liquid) or engulfed (gas) by the chemical.

The test specimens were examined after 3-hour contact and again after 20 hours. Quantitative data which were recorded were weight and dimensional changes and any visually apparent indications of attack; i.e., cracking, blistering, tackiness, discoloration, fogging, etc. Special physical property testing was performed on the butyl specimen following the 20-hour contact. These tests were limited to the measurement of tensile and tear strength (warp direction).

A greater than 10% change in weight and/or physical properties was selected as a fail criterion. Chemicals found to attack the test specimen within 3 hours were assigned a failed rating. Chemicals passing 3 hours but failing in 20 hours were retained in the test program, as shown in Figure 3 for concentration versus time testing.

2.3.5.2 Penetration Test (Butyl Rubber)

Chemicals advancing beyond the Swatch Test were subjected to penetration or permeation tests, which was the heart of the test program. The penetration test was limited to butyl rubber. Specimens of butyl rubber were contained in specially designed cells with the test chemical applied to the outer, thicker coated side. Clean air continuously purged the underside and suitable analytical techniques were used to measure the extent of penetration. Penetration by the nonvolatile inorganics was determined by swabbing the underside of the butyl specimen. The extent of penetration was measured at the end of 3 hours and/or at 20 hours.

The established fail criterion for the penetration test was the presence of positive signs of penetration. The criterion, which basically is dependent upon the sensitivity or detection limits of the analytical methods, was preferred over the general approach of establishing an allowable safe limit. The latter approach would have involved making uncertain toxiological decisions especially in instances where TLV type information was not available.

On the basis of the above, chemicals penetrating the butyl rubber within 3 hours were eliminated from further testing.

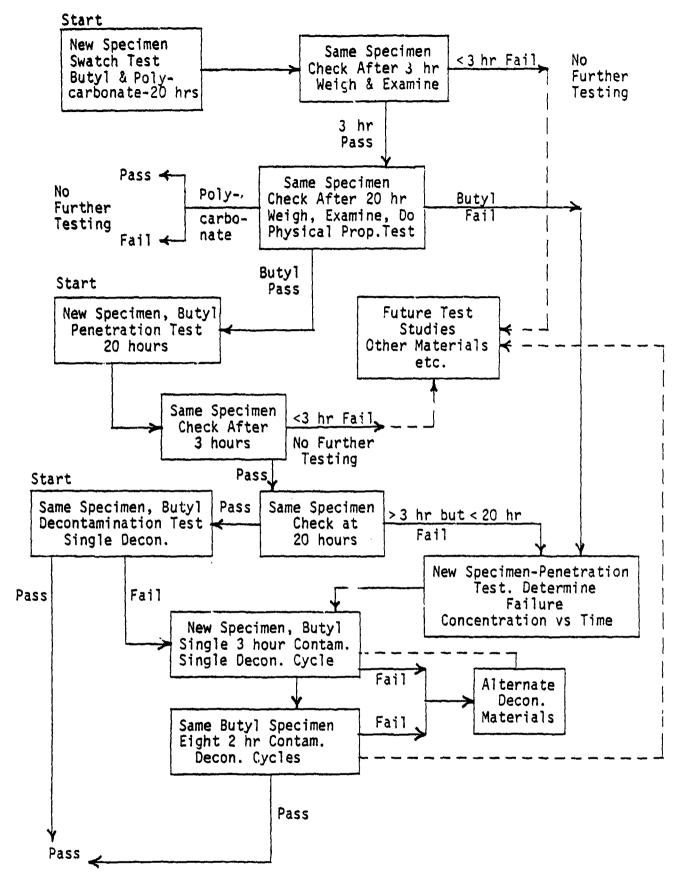


FIGURE 3 - MATERIAL COMPATIBILITY TESTING SEQUENCE

As is shown in Figure 3, chemicals breaking through the specimen between 3 and 20 hours were subjected to a concentration versus time test to determine service time. Chemicals passing the 20-hour test advanced to the Single Decon Test.

2.3.5.3 Single Decon Test

Buty! rubber specimens which passed the 20-hour penetration test were given a single decon test. A single decon test consisted of a cold water wash with 1% detergent. The specimen was then checked for decontaminating effectiveness. Any detection of the chemical (see paragraph 2.3.5.2 agove) on the butyl rubber represented a fail status. When buty! rubber specimens exposed for 20 hours failed the single decon test, a multi-decon test cycle was instituted.

Polycarbonate is impermeable, hence any degradation of the material is a surface phenomenon. During compatibility testing, the surface of the polycarbonate specimen was decontaminated with a cold water-1% detergent solution prior to post weighing of the specimen. The surface of the polycarbonate was adequately decontaminated by this procedure.

2.3.5.4 Multi-Decon Cycle Test

The multi-decon cycle test was limited to butyl rubber specimens. The objective of this test was to determine the ability to decontaminate when the butyl rubber was exposed to the chemical for shorter 2-3 hour time periods. Test sequence was planned upon initially exposing the butyl rubber for a 3-hour period followed by a decon with water and detergent as was similarly done in the single decon test. Any detection of the chemical represented a fail with no further testing conducted.

If no detection was made after the initial 3-hour exposure, eight 2-hour exposures with decontamination washes made between the 2-hour exposures was followed. After the last 2-hour exposure period, any detection of the chemical resulted in a fail and no further testing was done. Failure resulted in assignment into the category of future R&D studies with alternative decontaminating agents.

2.3.5.5 Concentration vs Time Tests

If a butyl specimen failed after 20-hour expsoure in the swatch or penetration test, it became scheduled for a concentration versus time test. The objective of this test was to determine at what period of time between three and 20 hours the butyl rubber failed. This test required monitoring between the three and 20-hour priod whereas monitoring in the penetration test was conducted at the end of the 3-hour period and the 20-hour

period only. Although it was scheduled that a new butyl rubber specimen be given a 3-hour exposure and decon after conducting the concentration versus time test, this test sequence was not conducted because it was deemed that the chemical had indeed penetrated to the degree that a decontamination problem existed.

2.4 Interim Report - Task IV

MSAR Reports 79-45 and 79-52 were issued in fulfill-ment of Task IV. Additional comments to MSAR 79-52 were prepared on lask III in a report issued 13 June 1979 and adopted as outlined in 2.3 above.

2.5 Material Compatibility Testing - Task V

Results of material compatibility testing and evaluation are given in Appendix C (test performance rating column). The sealed suit requirement was assigned to 403 chemicals as determined from Levels of Protection, Task I study. Using the Test Program Plan (Test III) as the testing guide, resulted in conducting tests with approximately 235 chemicals. Approximately 163 chemicals were not tested. On a mutually inclusive basis of having no data for either butyl rubber or polycarbonate, 39 of these chemicals were considered for future R&D study (see Section 5.0). Approximately 129 chemicals were not tested but given test performance rating. Rationale used was based upon chemical compatibility data and information pertaining to similar chemical reactivity, water reactivity, insolubility, properties and the like.

2.5.1 Swatch Test Results

Testing was conducted for butyl rubber and polycarbonate against those chemicals where no chemical compatibility data was available. Approximately 113 chemicals were classified for this test. The purpose of this test was to immediately screen out those chemicals which were obviously incompatible with butyl rubber. Figure 4 is a breakdown of the swatch test results. Failure rate was found to be about 60% for butyl rubber and 50% for polycarbonate for those selective chemicals scheduled in this test. Chemicals which passed (40% butyl rubber) went into the penetration test. No additional tests were conducted with polycarbonate.

2.5.2 Penetration Test Results

Penetration tests were conducted with butyl rubber only. This test applied to those chemicals which were classified as being compatible either from available information or from pussing the swatch test. The purpose of this test was to further establish resistance of butyl rubber to penetration since little or no data was available for the majority of chemicals, regarding penetration properties.

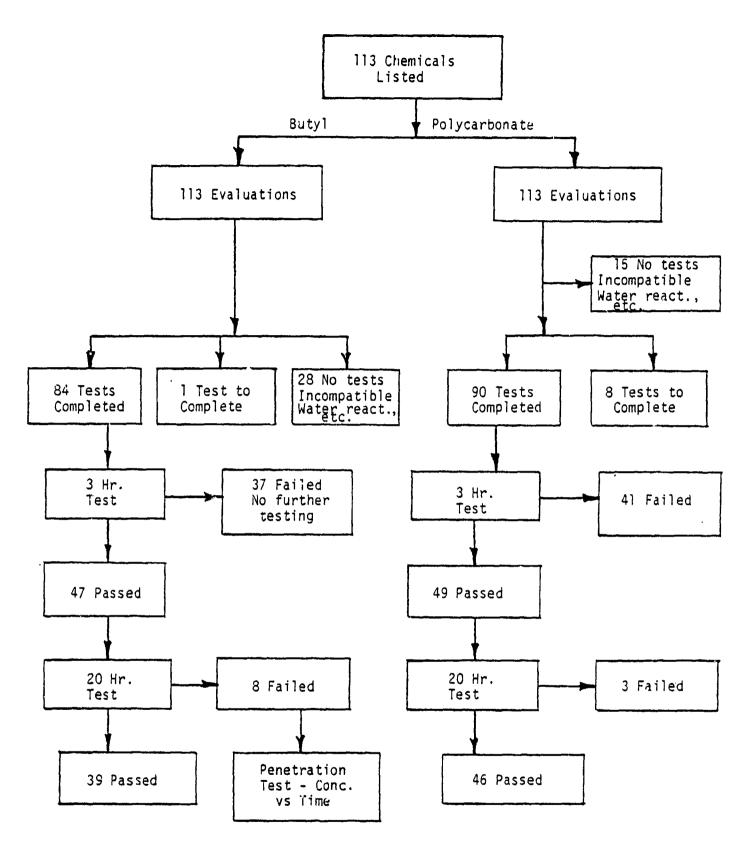


FIGURE 4 - SWATCH TEST DATA BREAKDOWN

Approximately 192 chemicals were used in conducting the penetration test. Basis for failing was any detection of the chemical. Figure 5 summarizes the penetration/decon test results. Approximately 60% passed the 20-hour penetration test. About 35% failed in 3 hours and about 5% failed after 20 hours. There were several cases where butyl rubber was found to be incompatible as evidenced by obvious attack. Chemicals which passed the penetration test were given a single decon test to further confirm test performance rating. Those which failed in 3 hours were considered basically not suitable. Those which failed between 3 and 20 hours were given the concentration versus time test.

2.5.3 Single Decon Test Results

If no penetration was detected after 20 hours of exposure to the chemical, a single decon test (cold water-1% detergent solution) was conducted. A breakdown of results from this test is shown in Figure 5. Approximately 66% passed the single decon test and were given a pass (acceptable) performance rating. The 34% which failed were given the multi-decon test.

2.5.4 Multi-Decon Test Results

The single decon test, which was performed after 20 hours of continuous exposure to the chemical, uncovered a number of chemicals in which decontamination could be a problem. These chemicals were subsequently examined in the multi-decon test which consisted of a series of short term contamination/decontamination cycles. Residual contamination was evaluated after an initial 3-hour exposure and again after a series of eight 2-hour exposures. Decontamination was performed after each exposure.

There were 26 chemicals investigated in the multi-decon test and Table 10 shows the breakdown of test results. Table 11 shows those chemicals which failed and were assigned to future R&D work for alternate decontamination studies. About 45% of the chemicals passed the first 3-hour exposure and 55% failed. Of the chemicals which entered the eight 2-hour portion of the test, a total of 30% passed and 70% failed.

2.5.5 Concentration vs Time Test Results

Swatch test and penetration test results determined the number of concentration versus time tests to conduct based upon failure between 3 and 20 hours. The purpose of this test was to determine the approximate time before breakthrough of the chemical. Appendix C (see table test performance rating column) lists service time/range and no special tabulation of this data is given since there was not a large number of tests conducted.

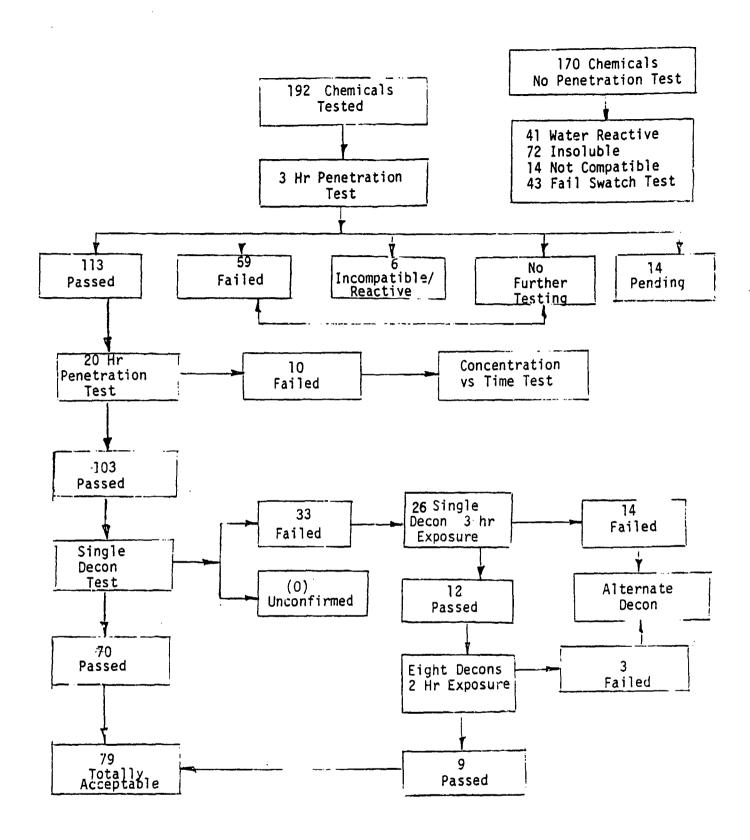


FIGURE 5 - BREAKDOWN OF PENETRATION/DECON TESTS - BUTYL RUBBER

TABLE 10 - DECON CYCLE TEST RESULTS FOR BUTYL RUBBER

Eight 2-hour Exposures & Decon Cycles*	Failed, 0.2 ppm Failed, 0.5 ppm Passed	Failed, 0.7 ppm Passed** Passed** (0.06 ppm)	Passed** Passed Passed**	Passed**	Passed**	Passed	etergent one minute
Single 3-hour Exposure & Decon	Failed, 1 ppm Passed Passed Failed, 0.52 ppm Passed	Failed, strong odor Passed Passed Failed, 0.8 ppm Passed	Passed Failed, 0.6 ppm Passed Failed, not detected but strong odor	Passed Failed, 0.4 ppm Failed, 0.6 ppm Failed, 0.8 ppm	Failed, 0.3 ppm Passed Failed, not detected	Passed Failed, not detected but strong odor	to chemical followed by a cold water detergent one minute cycle for residual contamination.
Chemical Name	Acrolein Acrylic Acid Aniline Benzonitrile Benzylamine	Creosote Dl-(Ethylhexyl)Phosphoric Acid 2,5 Dimethyl Hexane Dihydroperoxide, Wet l,1-Dimethyl Hydrazine Dimethyl Sulfate	Diquat Epichlorohydrin Glycidyl Metacrylate Isopropylamine Malathion	Methylucaine Methyi Amyl Alcohol n-Methylaniline Methyl Isobutyl Carbinol Nitrobenzene	2-Nitropropane Phosdrin Propyleneimine, Inhibited	Tetraethylenepentamine 2,4 Xylenol Hvdrogen Sulfide	tycle consists of 2 hr exposure then examined after the eighth todor.
CHRIS	ARL ACR ANL BZN BZM	CCT DEP DDW DMH DSF	DIQ EPC GCM IPP MLT	MTA MAA MIC NTB	NPP PHD PII	TTP XYL HDS	Each wash, Sligh

TABLE 11- CHEMICALS REQUIRING ALTERNATE DECON INVESTIGATION

CHRIS Code	Chemical Name
ARL ACR ANL	Acrolein Acrylic Acid Aniline
BZN	Benzonitrile
ССТ	Creosote, Coal Tar
DEP DMH	D1-(Ethyl Hexyl) Phosphoric Acid 1,1-Dimethyl Hydrazine
EPC	Epichlorohydrin
MLT MAA MAN MIC	Malathion Methyl Amyl Alcohol n~Methylaniline Methyl Isobutyl Carbinol
NTB NPP	Nitrobenzene 2-Nitropropane
PII	Propyleneimine, Inhibited
XYL	2,4-Xylenol

2.5.6 Seam Test Results

Purpose of the seam test was to provide additional information regarding seam design (Table 12). Chemicals tested were based on a selective method and most of the chemicals tested were those which passed the penetration/single decon tests. These tests pointed up that edge effects presented by the seam areas in fabrication of the HCPCO suit can hold up the chemical to a greater degree than smooth fabric presenting no seam edge effects. In regard to penetration, this test verified penetration results obtained with the fabric and also served to show that the seam design and construction was sound.

TABLE 12 - BUTYL SEAM TEST RESULTS

CHRIS Code	Chemical Name	Penet <u>3 hr</u>	ration 20 hr	Single <u>Decon</u>
AAC	Acetic Acid	Pass	Pass	Pass
ACY	Acetone Cyanohydrin	Pass	Pass	Fail (3.2 ppm)
AAM	Acrylamide	Pass	Pass	Pass
ANL	Aniline	Pass	Pass	Fail (8.2 ppm)
CHN	Cyclohexanol	Pass	Pass	Fail (1.4 ppm) Some blister ing on ad- nesive
NTT	p-Nitrotoluene	Pass	Pass	Fail (2.2 ppm)
PHN	Phenol	Pass	Pass	Pass
CBN	4-Chlorobutrylnitric	Pass	Pass	Fail (20 ppm) Some blister ing on ad- hesive
HDZ	Hydrazine	Pass	Pass	Pass

3.0 EXPERIMENTAL

The following sections detail the laboratory methods nd equipment used to evaluate those chemicals requiring sealed suits. The actual selection of the test methods was strongly influenced by the need to cope with an extremely large number of widely varying chemicals.

3.1 Chemical Sources

The bulk of the chemicals used in the test program were obtained from chemical supply houses. The purity of the chemicals was technical grade or better (>90%) and were used without further purification.

Specialty chemicals including formuations such as agricultural chemicals were obtained from a manufacturer listed in CHRIS for that specific chemical. In cases where the chemical was available in a number of different formulations, the manufacturer was requested to recommend the material most likely to be encountered "outside the fence". The most abundant type of chemicals belonging to the class are the agricultural chemicals; fungicides, herbicides, etc. In most cases, a single working ingredient was found to be available in a number of formulations; i.e., a water wettable powder, an emulsion, or as a solution in an organic solvent. The most common solvents or vehicles used in these preparations are kerosene, xylene, and toluene and all three are incompatible with butyl rubber. Toluene, for example, was examined briefly and was found to dissolve in butyl rubber to the extent of about 0.8 g/g (80% pickup).

CHRIS list chemicals requiring a fully scaled suit and marketed in solvents known to adversely affect butyl or polycarbonate were categorically assigned a "Fail" in the performance rating. These chemicals are identified in Appendix C by the insert "liquid formulation" adjacent to the chemical name.

3.2 Swatch Test

A swatch test was the initial test conducted when chemical resistated and available for polycarbonate and butyl subser. It consisted of exposing polycarbonate and butyl rubber specimens to the chemical and making inspection after a 3-hour and 20-hour period. Overall, one of the primary functions of the swatch test was to eliminate as early as possible the chemicals which grossly attacked the butyl and/or polycarbonate. Specimens were tested in separate containers so that the polycarbonate and butyl rubber specimens were observed individually with chemicals that were liquid or soluble solids dissolved in water. For gases, the specimens were placed in a cell through which the test gas was passed, as declibed in 3.3.1.

After three hours, the specimens were removed and examined for weight change, dimensional change, or obvious visual damage. This inspection was performed again after 20 hours; and in the case with butyl rubber, additional measurements were made to observe the change in tensile and tear properties in the warp direction of the fabric. A change of 10% in quantitative measurements, obvious visual damage or deterioration was used as fail criteria and no further testing was conducted. Only swatch tests were conducted with polycarbonate. Butyl ubber which passed the swatch test was then given subsequent penetration, single decon, etc. tests.

3.2.1 Polycarbonate Swatch Test Specimens

Polycarbonate used in the test was Lexan sheet Grade 9034 material. Size of the specimens was 1 1/4 inch wide x 3 1/2 inch long x 1/6 inch thick. The specimens weighed approximately 5 grams (see Figure 6).

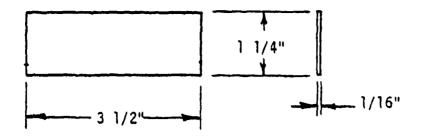
3.2.2 Butyl Rubber Swatch Test Specimens

Butyl rubber per MIL-C-12189E obtained from Plymouth Rubber Company was used in all tests in this study. This material is heavyw ight (11.0 to 13.5 ounces/square yard). Two specimens 3 inches wide x 6 inches long were cut for each test. After 20 hours of exposure to the chemical, one of the butyl specimens was given a 2-inch long slit and used to measure tear strength and the other one to measure the tensile strength in the warp direction (see Figure 6). The butyl rubber specimens weighed about 4.7 grams. Relative change in tensile and tear strength was obtained by comparing the exposed specimen to specimens which had not been exposed to the chemical.

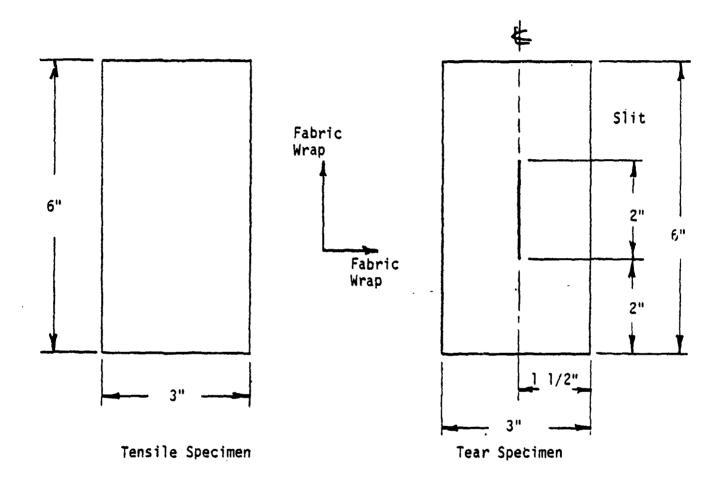
3.2.3 Swatch Test Data

Results of the swatch tests are given in Appendix C. Data collected was as follows:

- Wi = Initial weight polycarbonate and butyl rubber
- W₃ = Weight after 3 hours polycarbonate and butyl rubber
- W20 = Weight after 20 hours polycarbonate and butyl rubber
- Tensb = Tensile strength butyl rubber, unexposed blank
- Tens₂₀ = Tensile strength after 20 hours butyl rubber



Polycarbonate Test Specimen (Lexan Sheet Grade 9034)



Butyl Rubber (MIL-C-12189E) Test Specimens (Heavy Weight 11.0 to 13.5 oz/sq yd)

FIGURE 6 - SWATCH TEST SPECIMENS

Tearb = Tear strength - butyl rubber, unexposed blank

Tear $_{20}$ = Tear strength after 20 hours - butyl rubber Equations representing property changes were:

Weight change (3 hrs) = $\frac{W_3 - W_1}{W_1} \times 100 = \%$ weight change Weight change (20 hrs) = $\frac{W_20 - W_1}{W_1} \times 100 = \%$ weight change Tensile strength change (20 hrs) = $\frac{Tens_{20} - Tens_b}{Tens_b} \times 100 = \%$ Tensile Strength Change Tear strength change (20 hrs) = $\frac{Tear_{20} - Tear_b}{Tear_b} \times 100 = \%$ Tear Strength Change

3.2.4 Swatch Test Instrumentation

Instruments used for swatch test evaluations were:

- Weight Scale Mettler Model No. H31ARH
- 2. Strength Properties Tensile and Tear Instron Model 1135 Universal Testing Instrument [Set at Separation Speed of 31.7 cm/min (12.5 in./min) 225 kg (500 lb) load cell]

3.3 Penetration Test

The design of the penetration cell is shown in Figure 7. A number of modifications of this basic design were used in order to cope with the various chemicals. Figure 8 shows a manifold arrangement of cells which permitted four similar tests to be conducted simultaneously. In general, test chemical was applied to one side of the butyl specimen and the subsequent penetrant removed from the underside by an air sweep or by washing.

3.3.1 Cell Design and Modifications

The basic cell was constructed of stainless steel and with the exception of the fittings all components were obtained from a single manufacturer. The design was adopted predominantly because it was simple, easy to operate and was sufficiently flexible to accommodate the wide assortment of different chemicals.

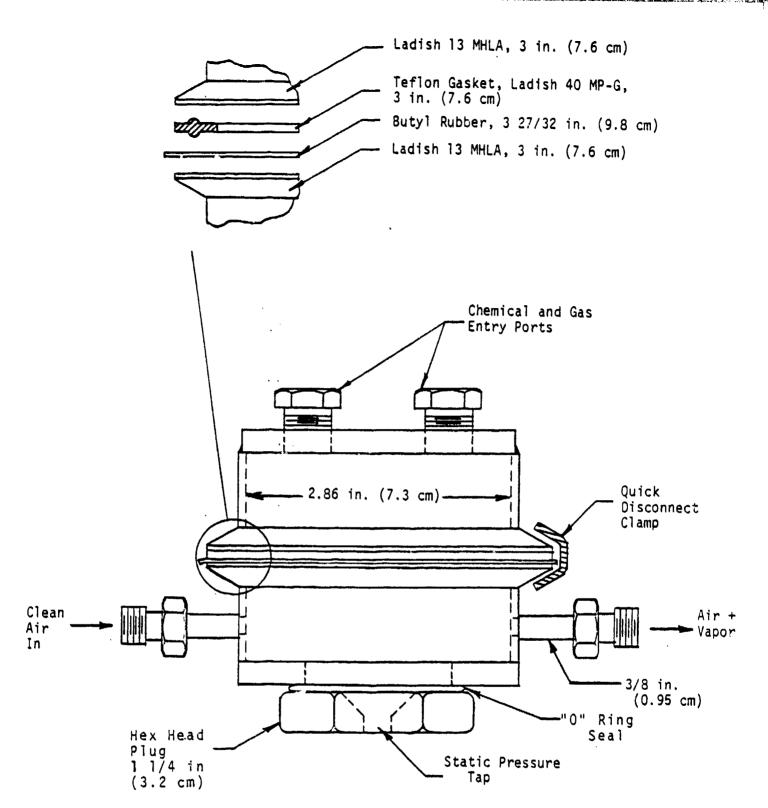


FIGURE 7 - PENETRATION CELL

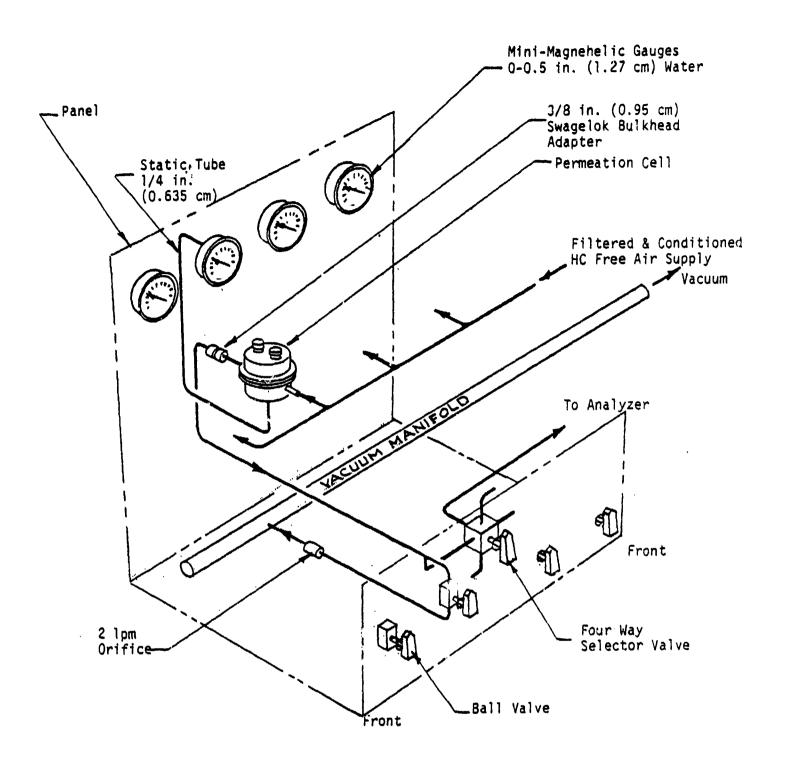


FIGURE 8 - PENETRATION TEST PANEL SCHEMATIC

As is shown in Figure 7, the cell consisted of two similar halves separated by a Teflon "0" ring.

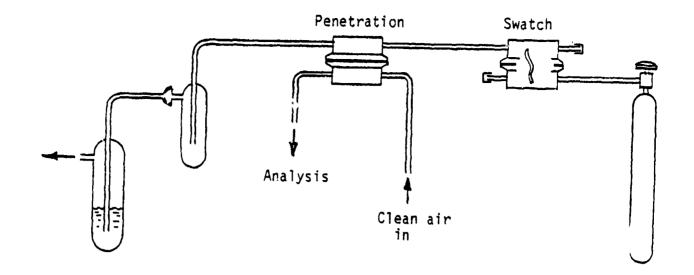
The slightly oversized butyl rubber test specimen was placed directly on the lower flanged cell unit with the "O" ring directly on top. The two halves of the cell were secured by a quick-disconnect clamp.

Leak tightness was assured by the action of the "O" ring which was compressed into corresponding slots cut into the flanges of each half end. The "O" ring when forced into the bottom slot stretched the butyl rubber slightly resulting in the production of a slight strain. When secured in place, the butyl rubber was pulled sufficiently tight to be wrinkle free. The strain on the butyl rubber was not measured but was consistent for all tests. The chemically exposed surface was 41.2 sq cm (6.42 sq in.).

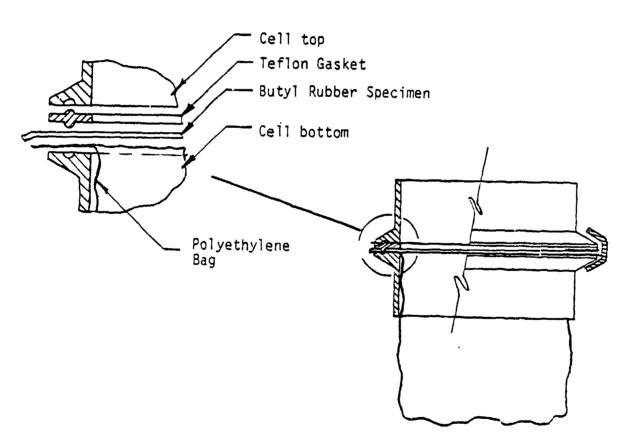
The cell used in studying the penetration from gases consisted of combining two of the lower half cell units shown in Figure 7. An overall gas test system is shown in Figure 9(a). The particular arrangement was employed in cases where only limited supply of the test gas was available. In this case, several tests were conducted at the same time in order to conserve challenge chemical. Tests with gases were conducted with a continuous flow of test gas passing above the butyl specimen. Flow of test gas was usually 1 to 5 cc/min.

Figure 9(b) shows the design of the cell used to study penetration of inorganic materials, specifically penetrant chemicals which might become contaminated by laboratory or other extraneous contaminants. As is noted in Figure 9(b), the bottom half of the cell is open and a flexible polyethylene container was sealed under the butyl specimen. With this design, the penetrant emerging from the underside of the butyl specimen was automatically sealed off from the walls of the cells and from other sources of contamination. Subsequent operations consisted of merely cutting the plastic bag and exposing the underside of the butyl specimen.

Recovery of penetrant was achieved either by continuously purging the underside of the test specimen with a flow of hydrocarbon free air or by vigorously rubbing and washing the penetrant free. Penetration by simple organic liquids was determined by sweeping the underside with a flow of 2 lpm of processed air cleaned by passage through a carbon bed followed in line by a bed of heated hopcalite. The resultant hydrocarbon content of the sweep air was maintained at <2.0 ppm.



a. Arrangement for Swatch and Penetration Test with Gases



b. Arrangement for Penetration Test in Contaminant Free Environment

FIGURE 9 - MODIFICATIONS OF PENETRATION CELL FOR GASES AND CONTAMINANT FREE SERVICE

Penetration by metal salts solution was determined by vigorously swabbing the underside of the butyl using 10 cc of water and processed glass wool or a flat Teflon rod.

3.3.2 Typical Test Operation

Butyl rubber test specimens consisted of circular discs measuring 9.76 cm (3.84 in.) diameter overall.

For simple organic liquids, the test specimen was installed in the cell as shown in Figure 7. The completed cell was then placed on the four unit module as shown in Figure 8. A vacuum pump aspirated a flow of 2 lpm of clean air through the lower portion of the cell. A reading of the Magnehelic gage insured that a significant negative pressure relative to the atmosphere was not developing. Normal vacuum developed as a result of aspirating the air flow was limited to <0.13 cm (<0.05 in.) water. At selected intervals, the clean air sweep passing through the chemically free cell was diverted to the analyzer. The cell was considered clean when the analyzer showed no difference in quality between the clean air at its source and the clean air which passed through the cell. At this point, the test was initiated by the addition of 10 cc of chemical to the upper portion of the cell. The cell was continuously purged by the 2 lpm flow until the test was terminated (20 hrs). As a minimum, analysis of the quantity of penetrant in the purge air was performed at the end of 3 hours and again at 20 hours.

Penetration by inorganic salt solutions was studied in the open-end cell unit shown in Figure 9(b). A quantity of 10 cc of a saturated water solution of the salt was applied to the top of the butyl specimen. No air purge was used and the test apparatus was stored at room temperature for 20 hours. After this period of time, the chemical was removed, the cell was inverted and the underside of the butyl was exposed and subsequently scrubbed/washed with 10 cc of distilled water. Collected washings were subsequently analyzed for selected metal ions in components.

Methods of detection are summarized in Section 3.8.

3.4 Single Decon Tests

When a butyl rubber specimen passed the 20-hour penetration test, the effectiveness to decontaminate the chemical from the butyl rubber was examined. The single decon test consisted of removing the butyl rubber from the penetration test apparatus, wiping and rinsing off excess chemical. It was then placed in a wire basket suspended from the lid c, a household blender containing one liter of cold water with a 1% concentration of powder detergent. The blender was operated for one minute.

The specimen was removed and rinsed with cold water and excess water removed with an absorbent paper towel. The specimens were checked for decontamination of the organic or inorganic chemicals as discussed below.

3.4.1 Organic Chemicals

Effectiveness of the single decon test was examined by placing the butyl rubber in the penetration test cell. The chemically exposed side was mounted as the underside where it was swept with clean air flowing into the total hydrocarbon analyzer. Any increase in the total hydrocarbon concentration above the clean air reading indicated residual chemical contamination and ineffective decontamination. Ineffective decontamination then required that a new butyl rubber specimen be given a multiple decontamination test.

3.4.2 <u>Inorganic Chemicals</u>

Effectiveness of the single decon test was examined by placing the butyl rubber specimen on a petrie dish. A 10 cc syringe was filled with distilled water and discharged onto the chemically exposed side of the butyl specimen. A Teflon spatula was scraped over the butyl rubber and distilled water was used to extract any residual chemical remaining. The distilled water/solution was decanted from the petrie dish into a test tube and analysis made of the water in the manner described in 3.8 to determine if the single decon test was effective. Any detection of the test chemical required that a new butyl rubber specimen be given a multiple decontamination test.

3.4.3 Single Decon Instrumentation

Instrumentation and procedures used in the single decontest consisted of those discussed in the penetration test.

3.5 Multi-Decon Test

When the butyl rubber specimens exposed to the chemical for 20 hours failed after a single decon, a multi-decon test was conducted. As discussed previously, this test consisted of exposing a new butyl rubber specimen for three hours followed by a single cold water detergent wash. Examination for residual chemical was made and if chemical was present future R&D alternate decontamination studies were deemed necessary. If no detection was made after three hours, eight 2-hour exposures and decons were performed and after the eighth cycle examination made for presence of residual chemical. Similarly, as after the initial 3-hour exposure, detection of the chemical placed it in the category of future R&D alternate decontamination studies. If the chemical was not detected after the exposure cycles, butyl

rubber was given a pass performance rating. If any contaminant was detected after the initial 3-hour exposure or after the eight 2-hour exposures, the butyl rubber was assigned a fail performance rating (see Appendix C).

3.5.1 Butyl Rubber Test Specimen

Butyl rubber specimens (MIL-C-12189E) were cut identically to those used in the penetration test.

3.5.2 Test Procedure

Apparatus and analytical procedures were the same as for the penetration test. Chemical was placed on the butyl rubber for a three hour period. The butyl rubber was removed from the test cell and then given a single one minute decon wash and dried. It was then examined for any trace of the chemical. If the chemical was detected, it was assigned in future R&D alternate decontamination studies. If no chemical was detected, the butyl rubber was exposed to the chemical for eight 2-hour exposure cycles with a decon wash given between each cycle. After the eighth cycle, the butyl rubber was examined again for any trace of the chemical.

3.6 Concentration vs Time Tests

Butyl rubber failing between three and 20 hours was given the concentration versus time test. Butyl rubber specimens, apparatus and test procedures were the same as followed in the penetration tests with the exception that additional monitoring was done to determine breakthrough time period.

3.7 Seam Test

The purpose of this test was to investigate edge effects and soundness of the seam design used in the HCPCO suit construction.

3.7.1 Seam Test Specimen

Typical seams used in the HCPCO suit were obtained from ILC Dover, Fredrica, Delaware. The test specimen was circular and cut to the size used in the penetration test. It was cut so the seam was centerline in the penetration cell. Seam construction is a lap fold sewn with a 2.54 cm (1 inch) butyl strip glued (Bostic Co. 1177) to both the inner and outer surface of the butyl rubber sections it forms.

3.7.2 Seam Test Chemicals

Seam test chemicals were chosen primarily from those chemicals which had been tested in the penetration/single decontests and found acceptable.

3.7.3 Test Procedure

The seam specimen was placed in the cells used for the penetration test. Because these seamed specimens were thicker than the single layer butyl rubber fabric used in the penetration test, additional takeup to achieve a seal required the use of a silicone rubber gasket rather than a Teflon gasket. Since the silicone rubber gasket is less chemically inert than the Teflon, it was placed under the butyl rubber seamed specimen to prevent it from contacting the chemical. Otherwise, apparatus and procedure remained the same as followed in the penetration test; the specimen being examined after three hours and 20 hour time period.

3.8 Summary of Analytical Methods

Table 13 summarizes the analytical methods used in this program. For practical reasons, the number of different procedures were held to a minimum.

The lower limits of detection listed in Table 13 are a reasonable estimate of the minimum quantity of chemical that could be detected under the test conditions. Invariably, trace background contamination was present in which case the quantity of chemical which could be detected above the background level was approximately twice the suggested or ideal lower limit.

3.8.1 Detection of Organic Vapors

Analysis of organic vapors in the effluent purge gas was performed exclusively by a MSA Hydrocarbon Analyzer. This instrument offers an extended range with the bulk of the work performed on the 0-12 ppm full scale range. The instrument was calibrated with 5.0 ppm methane gas mixture supplied by Matheson, East Rutherford, NJ.

3.8.2 Detection of Metal Salts

A Model 303 Perkin Elmer Atomic Absorption Spectrophotometer was used to analyze metal ions dissolved in the specimen wash liquid.

Penetration and decon effectiveness were determined by using a control or blank and also included the generation of a recovery factor. In short, washing an uncontaminated test specimen was performed in order to determine background contamination that could be present. To a second specimen, a known quantity of a corresponding metal salt in solution was added and allowed to dry for 20 hours. Usually, the added spike totalled 50 µg of the metal ion. The specimen with the added contamination was subsequently washed and the washings processed along with the washings of test material. The amount recovered was usually about 30 to 50%

TABLE 13 - SUMMARY OF ANALYTICAL METHODS

Basic Method Employed	Chemical Detected	Lower Limits of Detection (ppm)*
Hydrocarbon Analyzer	Organic Vapor	0.05
Atomic Adsorption	Metal jons Cd++ Cr(III) Cr(VI) Na+ K+ Pb++ Hg(II) Hg(II) Zn++ Li+ Ni++	0.04 0.15 0.15 0.015 0.05 0.07 11.0 11.0 0.025 0.04 0.15
Specific Ion Elec- trode	Selected non- metal jons NH4 F-	0.2 0.2
Detector Tubes	Selected Gases/Vapors SO ₂ H ₂ S Cl ₂	<2.5 <1-2 <0.5

^{*}Present in the 2 lpm air sweep or in 10 cc of wash liquid.

of the initial added quantity. In all cases, the wash procedure was found effective for metal ion recovery when used at the 50 μg level of contamination.

3.8.3 Detection of Nonmetal Ion

Ammonium and fluoride in wash liquid were determined using specific ion electrodes obtained from Fisher Scientific. Recovery factors and trace contamination levels (blanks) were established as described above.

3.8.4 Use of Detector Tubes

The detector tubes were obtained from MSA. These tubes are certified to offer a minimum of 35% accurcy at the concentrations listed in Table 13. When the tubes are used for qualitative purposes the lower limits of detection can be reduced by at least a factor of three. In the laboratory, the detector tubes were used for both quantitative and qualitative purposes.

4.0 DISCUSSION OF RESULTS

Absence of complete toxiological data was the primary obstacle in establishing Levels of Protection categories for all of the 985 chemicals. A sealed suit requirement was designated for 403 chemicals. There were 327 chemicals in this list of 403 chemicals which were based upon Sax's Toxic Rating Guide and 76 chemicals which were based upon other standards or best judgment. In the total of 985 chemicals reviewed there were 45 chemicals in which no level of protection assignment was made. The breakdown of the Level of Protection results were as follows:

Requirement (Level of Protection)	No. Chemicals
Sealed Suit	403
Nonsealed Suit	436
Fire Suit	15
Goggles, gloves, hard-	
hat and footwear	86
Unassigned	45
Total	985

Assignment of level of protection was given for 940 of the 985 chemicals and 45 chemicals remain for further definition pending additional toxiological information. Assignment of 403 chemicals to the category of sealed suit became the basis of material compatibility testing investigation.

Survey of commercially available protection clothing revealed that a variety of sealed, nonsealed and other protective apparel are available. Being less sophisticated and not subject to special military type specifications, the commercial type protective clothing is much less expensive than the HCPCO sealed suit. Commercially available protective clothing material is primarily elastomeric polyvinyl chloride, polyvinyl alcohol, butyl rubber, neoprene rubber, urethane, polyethylene, etc., some of which are sandwiched over nylon or cotton inner fabrics. Even considering this variety of materials relative to the listing of CHRIS chemicals, lack of chemical resistance data obscured a complete alternate material substitution listing. This also applied to head protection materials such as clear polyvinyl chloride, polycarbonate, cellulose acetate and acrylic most commonly employed. Because of its chemical inertness, Teflon promises the most universal chemical resistance; however, stateof-the-art fabrication has negated its applicability because of its inherent inertness properties. A somewhat clear Teflon material is available and can be considered for a visor although

it does not possess as good visual properties as clear polycarbonate. Teflon yarn is available for sewing seams. Bonding patches over the seams would be required to cover holes left from sewing.

Because of insufficient chemical compatibility data on butyl rubber and polycarbonate used in the HCPCO suit, a testing program was required. In addition, testing was required to further explore penetration and decontamination properties since little or no such data was available for the majority of the chemicals. Consequently, a test program plan had to be devised to establish an overall performance rating since chemical resistance data was insufficient for butyl rubber but judged adequate for polycarbonate. A test program plan was developed in this effort. This plan represents a model of the experimental process one can employ in evaluating the material protective properties against a spilled chemical contaminant. The output of the experimental model served as a guide to ascertain suitability of butyl rubber for use as a sealed suit.

The plan included provisions for dealing with certain chemicals which, because of practical or technical reasons, could not be processed in a straightforward manner. Along these lines certain classes of chemicals were excluded from the experimental portion of the test plan. These classes were: water reactive, solid insolubles and chemical formulations containing one or more components known to be material incompatible.

The threat of chemicals which are both solid and insoluble in water is mainly particulate and a well-made suit would normally offer adequate protection. The performance rating assigned in this case was the generally accepted literature designation. Chemicals which were formulated as mixtures containing a solvent or vehicle known to adversely affect the test materials were failed. This was a frequent occurrence with agricultural type formulations, e.g., chlordane in hexane and the specific ingredient or solvent was suitably identified.

Realistically, a spill of a water reactive compound can be encountered in the field as a mixture ranging in composition from 100% pure compound to 100% reaction products. A closer examination of the chemicals for sealed suit use showed the predominant reaction product to be a strong mineral acid, notably HCl. Selected studies showed HCl solution at concentrations of even 10% caused serious damage to the structural strength of the butyl. In one case, no penetration was observed from the parent or unreacted compound. The addition of water caused a dramatic increase in temperature ($\Delta T = 25\,^{\circ}\text{C}$) and resulted in immediate chemical penetration. As a consequence, all water reactive chemicals yielding strong mineral acids were categorically failed. Water reactive materials yielding less reactive products were assigned the literature designation with this rating limited to the parent or unreacted chemical.

Basically, five different types of experimental tests were conducted in the test plan sequence. The tests and test functions were:

<u>Test Type</u>	<u>Test Function</u>
Swatch	Establish basic compatibility data for chemical where no data was available (applicable to both butyl rubber and polycarbonate).
Penetration	Evaluate butyl rubber as an impermeable protective barrier.
Single Decon	Isolated chemicals in which no decontamination problems exist after 20 hours chemical contact.
Multi-Decon	A less demanding test revealing decontamination effectiveness for a shorter or single mission 2-3 hour exposure period to the chemical.
Concentration vs Time	Establish use limits of butyl rubber relative to service time.

To provide supplementary data, seam testing was conducted to show soundness of the HCPCO suit seams and any limitation of decontaminating presented by edge/adhesive effects.

Appendix C lists a test performance rating of polycarbonate and butyl rubber for 403 chemicals requiring a sealed suit. A summary of test performance rating results were as follows:

<u>Polycarbonate</u>

Pass (20 hr contact) Pass (>3 hr <20 hr) [Pass] (Parent compound, water reactive) Fail (<3 hr contact) [Fail] (Parent compound, water reactive) No data	211 1 29 90 29
No data Total	403

Butyl Rubber

Pass (totally acceptable)	142
Pass (totally acceptable) Pass (>3 hr <20 hr) [Pass] (Parent compound, water	5
reactive)	6
Fail (unacceptable) [Fail] (Parent compound, water	183
[Fail] (Parent compound, water	
reactive)	4
No data	_63
Total	403

Both polycarbonate and butyl rubber offered excellent protection against all metal ions and neutral salts. This fact tends to exaggerate the applicability of these two plastics. Table 14 lists the overall performance of these materials against specific classes of chemicals as categorized by CHRIS. The findings suggest that both polycarbonate and butyl rubber are somewhat weak against the more abundantly produced organic chemicals. On the basis of test results wherein chemical compatibility for both butyl rubber and polycarbonate exists, 130 chemicals can be handled with the HCPCO suit. Acceptability for use mandates that both materials be compatible with the chemical of interest.

It is logical, especially in view of the high failure rate, to examine the viability of the test methods. The impact of specific experimental tests on the number of fail ratings is shown below:

Type of Test	Test Condition	% Failed*	% of All Failures
Penetration	3 hr contact	35*	78.5
Penetration	20 hr contact	9.4	
Single Decon	20 hr contact	33	
Multi Decon	3 hr contact	52	17.5
Multi Decon	8 2-hr contact	25	4

*Based on total number entering test

The bulk of the failures were recorded during the first three hours of the penetration tests, and this suggests an inherent weakness in the butyl. Conducting the test for 20 hours did not result in a significant increase in failure rate. It is interesting to note that one out of three chemicals could not be removed from butyl rubber by a simple water/soap wash when the contact time was 20 hours. Approximately half these chemicals could not be removed when the contact time was reduced to 3 hours. Overall, the various decontamination tests were responsible for only 16 failures or a total of 21.5% of all failure ratings.

TABLE 14 - CHEMICAL RESISTANCE BY CLASS

		Performance Ra	ting
	Chemical Groupings	<u>Polycarbonate</u>	Butyl
1.	Nonoxidizing Mineral Acids	Pass	Fail
2.	Sulfuric Acids Spent 98% or less	Fail	Fail
3.	Nitric Acid	Fail	Fail
4.	Organic Acid	Pass	Pass
5.	Caustics	Fail	Pass
6.	Ammonia	Pass	Pass
7.	Aliphatic Amines	Fail	Fail/Pass
8.	Aromatic Amines	Fail	Fail
9.	Organic Anhydrides	Pass	Pass
10.	Isocyanates	Pass	Pass
11.	Vinyl Acetate	Pass	Fail
12.	Acrylates	Fail/Pass	Fail
13.	Substituted Alkyls	Pass/Fail	Fail
14.	Alkylene Oxide	Pass	Fail
15.	Epichlorohydrin	Pass	Fail
16.	Ketones	Pass	Fail
17.	Aldehydes	Fail/Pass	Fail
18.	Alcohols, Glycols	Pass	Pass
19.	Phenols and Cresols	Pass/Fail	Fail/Pass
20.	Caprolactum Solution	Pass	Pass
21.	Aromatic Hydrocarbons	Fail	Fail
22.	Vinyl Halides	Pass/Fail	Fail
23.	Halogenated Hydrocarbons	Fail/Pass	Fail/Pass
24.	Nitriles	Pass	Fail
25.	Carbon Disulfide	Fail	Fail
26.	Ethers	Fail	Fail
27.	Nitrocompounds	Fail/Pass	Fail

Certainly, the adoption of "detectable quantity" for the above tests impacted directly on the relative number of pass or fail ratings. Undoubtedly detection levels for the tests could be lowered by the use of more sophisticated and involved methods of enrichment and detection which presumably would have resulted in a higher failure rate. This can be countered somewhat by the fact that the final disposition of the chemical either as a penetrant or as a residual following decon is uncertain. Realistically, the Strike Team does not experience a chronic exposure to the same chemical as would be the case for an industrial worker. Overall, the method and procedure used in the experimental program reflects a certain middle-of-the-road approach.

This study provided an improved and protective base for recommending levels of protection and protective equipment needed by Strike Team personnel. Because of the large number of chemicals, the different types of chemicals and the lack of toxicological data, recommendations could not be formulated for all 985 chemicals on the CHRIS list. Additional R&D is required, and specific recommendations are presented in Section 5.0.

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5.0 RECOMMENDATIONS

5.1 General Comments

Butyl rubber and polycarbonate materials are used in the HCPCO sealed suit. A preliminary literature survey on chemical compatibility of these materials indicated a higher degree of compatibility than actually found in the testing programs. This was because the testing program was more extensive in scope and was used to evaluate not only compatibility but also penetration and decontamination. For example, butyl rubber was found to be about 38% compatible from the experimental testing whereas the indicated chemical compatibility from preliminary literature survey indicated much higher chemical compatibility.

Combining compatibility test result data for both butyl rubber and polycarbonate materials jointly further reduces the compatibility of the HCPCO suit ensemble. Mutual combination of compatibility data for these materials indicates the suit to be satisfactory for only 130 (32%) of the 403 chemicals requiring use of a sealed suit.

Alternate candidate materials would be expected to have a lower acceptability rate than what might be expected from literature and manufacturers data, also. However, alternate material investigation is viewed as necessary because of the limitations of butyl rubber and polycarbonate used in the HCPCO suit.

5.2 Future R&D Effort (Task VI)

Future R&D efforts are recommended as follows:

- 1. Alternative material investigation for butyl rubber and polycarbonate used in the HCPCO suit.
- 2. Additional testing of those chemicals which were not evaluated because of water reactivity, insolubility, unavailability coupled with the absence of pertinent literature data.
- 3. Determination of levels of protection for those chemicals with insufficient toxicological data. Sealed suit level of protection assignment will require additional butyl rubber and polycarbonate testing.
- 4. Assessment of alternate decon agents and methods.

5.2.1 Alternative Material Investigation

Appendix A lists some alternative materials for cases where butyl rubber and polycarbonate were not compatible or when data did not exist. This data served only as a quideline to establish compatible or incompatible status; the more rigorous testing conducted in Task V (Material Compatibility Testing) shoved that an unacceptable status could develop even when compatibility was indicated. Table 15 was prepared to list those chemicals and alternative materials that can be considered for future R&D investigation. Available literature was used to select alternative materials. Thus, it is a preliminary guideline for materials which must be further verified by testing. Table 15 also shows the need for alternative materials because of the weakness of butyl rubber and polycarbonate when their performance is compared with the variety of chemical groups involved. There are cases in which the chemical or chemical group in which a material choice could not be determined. In some cases where chemical compatibility data indicates good or excellent resistance, disappointing results were obtained in the more rigorous tests conducted in this program.

5.2.2 Additional Testing of Butyl Rubber and Polycarbonate

There were instances where no compatibility data was generated relative to the testing program plan. This was because of insoluble solids, lack of availability and insufficient information to relate it with a similar chemical reactivity group (see Table 16).

On a practical basis, solid chemicals would be encountered as dusts or general particulate matter. The insoluble nature of these chemicals suggests a certain chemical inertness although this argument is not sufficiently convincing to suggest that no attack (solid-solid reaction) with the protective ensemble can occur.

Some decisions on compatibility were based on the byproduct materials, primarily acids formed by the reaction of the
parent product with water. In some cases, the data indicated that
butyl rubber and/or polycarbonate were compatible with the parent
compound but not with the reaction products. Additional testing
of water reactive compounds and reaction products is recommended.

5.2.3 Levels of Protection

Table 7 lists 44 chemicals for which assignment of levels or protection data were not available. It is possible that additional toxicological information may be forthcoming from chemical manufacturers and other sources. In general, however, toxicological studies are long term activities requiring many years of study before

TABLE 15 - POTENTIAL ALTERNATIVE MATERIALS FOR INVESTIGATION AS PER CHRIS CHEMICAL REACTIVITY GROUPING

handard Burnelland	Available	Data	Ho	od Mater	aT .		Alternative Mater	Sult Hateri	· · · · · · · · · · · · · · · · · · ·
hemical Groupings	PoTycarbona	Buty!	Acetate	Tiny	Others	VIRVI	Polyethylene	Reoprene	Uthers
 Mon-Oxidizing Mineral Acids Mydrochloric, Hydrofluoric Ac 	fd Pass	Feff	NA.	₹ NA		x			Chlorinated Polyethylan
Spent & 98% or less	Fatl	FAIT	٠.	1 x	1	l x	l	١.	}
. Mitric Acid	Fail	Fa11]	x		x]	X	}
. Organic Acid	1	''		} ^	"	^	1 "		}
Acetic	Pass	Pass	NA.	NA.	į MA	NA	MA	NA.	NA.
Acry'ic Acid Oxalic Acid	Pass Pass	(rass)	MA MA	NA NA	NA NA	x	- · · · · · · · · · · · · · · · · · · ·	*-	Chlorinated Polyethyler
. Caustics	}	(,		} ~	\ '''' \	^	1		Chlorinated Polyathyler
Potassium & Sodium Hydroxic	Fail	Puss		x	1 1	MA	KA	MA]
- Ammonia	(Pass)	Pass	. #A	NA.	NA	NA.	MA	NA.	HA.
Aliphatic Amines Butyl Amine	F.4,			1	1 1				
Cyclohexylamine	Fait Fait	Fail		}	1 1	••			Chlorinated Polyethyler
Dibutylamine Di-n-propylamine	Fail Esti	Fa11]	1 1	••		••	Chlorinated Polyethyler
Ethyl Amine	Fail Fail	Fail		l	!!		Į į		
Propylamine Tetraethylenepentamine	Fail Fail	Fail		}	1 1		}		}
Triethylamine	Fail	Pass Fail		1	1 . 1		ļ		
Triethylenetetramine	Fall	Pass		{	1 1				
Arometic Amines Aniline	Fail	Fa11		}					
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	1	1 1			Polypro-		!!!		
Organic Anhydrides	1	1 1		1	pyrene		1		
Organic Anhydrides Acetic Anhydride	Pass	Pass	NA.	NA.	NA	NA	NA I	NA.	M
. Isocyanates				1			ļ .		
Diphenyl Methane Disocyanate	Pess	Pass	NA.	MA	i iA	MA	· M	MA	NA
• Vinyl Acetate Vinyl Acetate, Inhibited	(Pass)	(Fa11)	NA	KA	l M		1 1		063
Acrylates	} (,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	_ { `` ` ```}	nun ,	~	{ ~ {		{ {		Chlorinated Polyethylen
Ethyl Acrylate (Inhibited)	(Fa11)	Fa11		[!!		}		Chlorinated Polyethylen
2-Ethylhexvl Acrylate (Inhibit Ethyl Hethacrylate (Inhibited)	ed) (Fa11) (Fa11)	Fail Pass	K4	NA.	NA	MA	l na l	NA.	NA.
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) (Fa11)	Fatt			} {		}	- {	
Acrylonitrile (Inhibited)	(Pac	Fa11	NA	NA.	NA		} <u> </u>		Chlominated Delumber
Allyl Alcohol Allyl Chloride	(Pe (Fari	Fa11	NA	NA	NA NA		{		Chlorinated Polyethylen Chlorinated Polyethylen
. Alkylene Oxide	(54) (Fa11			} }		}	1	• •
Butylene Oxide	(Pass)	Fall	NA	NA.	l MA] [1	
. <u>Epichlorohy</u> dr <u>in</u>	} ```				1 "		1	1	
Epichlorohydrin	(Pass)	Fad1	MA	MA	NA		l x l	x l	••
. Ketones Mesityl Oxide	(Pass)						1		
.marty. Oxide	(7452)	Fail	NA	NA	NA				Ethylene Propylene Chlorinated Polyethylene
- Aldehydes	1	1 1			1 1		1	1	anton history to the cult tell
Aceta I dehyde	(Fa11)	fa11	••		Ethyl		1		
Acrolein	(Pass)	Fet1	NA.	NA	Cellulos			X	Sflicone
Butraldehyde Decaldehyde	Fail Fail	Fail			{		1		
Ethy I hexa I dehyde	Fail	Fail (i		[]	x	x		••
Formaldehyde Glutraldehyde Solution	Pass Pass	Fail Fail	NA NA	NA NA	NA NA	X		{	Mylon, Viton
Alcohols, Glycols	1	1		, man	} ~ }		}	j	
Cyclohexanol	Pass	Pass	MA	NA	MA	NA	NA	. MA	KA
. Phenois & Cresols Carbolic Oll	40						! }	ł	
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Caprolactum Solution	(Pass)	(Pass)	NA	NA	M	NA .	l na l	MA {	NA.
Aromatic Hydrocarbons		1			Acetal		· · · · · · · · · · · · · · · · · · ·	~ }	na.
Benzene Cumene	Fa11	Fa11			olyviny1				Viton
. <u>Vinyi Halides</u>	1	1 1	}		Acetate	1	}	}	
Vinvi Chloride	(Pass)	(Fa11)	NA	MA	NA.	i	1	1	
Vinylidene Chloride (Inhibited	Fa11	Fail					ļ		
Halogenated Hydrocarbons Chloroform	Faft	1	- 1]]	}	
Dichlorobenzene	Pali	Fail				ţ	İ	ļ	Mylon Yiton
1-1, Dichlorpethane Dichlorpethyl Ether	Fail	Fall	. }			}	1	}	***************************************
Ethyl Chloride	}	(Pass)	,]	}	}	}	••
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Methyl Bromide	Fall	(Fa17)	1	1		i	j	j	
1,1,2,2-Tetrachloroethene	(Fail)	Fail	{	- 1	1	{	1	Į.	
Mitriles Acetonitrile	(Pass)	6> 20 hr	NA			1	-	}	
Adiponitrile	(Pass)	Fa11	MA	NA NA	NA NA	}	1	ļ	
Carbon Disulfide		1 1	j		th 1 orine tad	į	1	į	Mylon
Carbon Disulfide	Fail	Fa11	•• }		Polyether	į	1	1	nyron Viton
Ethers Tetrahydrofuran	(Fa11)	Fafi	{	ł		ł	1		
Ni trocompounds	(((((((((((((((((((((Tall)	1	}	1	· ·	}	}	
	1 /5-/11	1 5000	- 1	Ì	1	- 1	1	1	
(mono-) Mitrobenzene) or 2-Mitropropane	(Fail) (Pass)	F#17							

TABLE 16 - SEALED SUIT CHEMICALS FOR FUTURE R&D EVALUATION

BDE Bisphenol A Diglycidyl Ether Calcium Arsenite Insoluble X X X X X X X X X	CHRIS			Performance D	ata Sought
CAS O-Chloronitrobenzene CHOC Chromyl Chloride COU Dry Di-(p-Chlorobenzoyl) Peroxide DTD Diutyl Phenol Diuron DNZ DNZ DNZ DNZ DNZ DNZ DNZ DNZ ESF Endosulfane ETI Etylene Imine FFX Ferric Fluoride Lead Stearate LSA LEAd Stearate LSA LEAd Stearate LSA LEAd Stearate MRT MRY		Chemical Name	Untested Status		
CAS O-Chloronitrobenzene CHOC Chromyl Chloride COU Dry Di-(p-Chlorobenzoyl) Peroxide DTD Diutyl Phenol Diuron DNZ DNZ DNZ DNZ DNZ DNZ DNZ DNZ ESF Endosulfane ETI Etylene Imine FFX Ferric Fluoride Lead Stearate LSA LEAd Stearate LSA LEAd Stearate LSA LEAd Stearate MRT MRY					
OFT Chloronitrobenzene Chromyl Chloride Chromyl Chr					X
CMC Chromyl Chloride Coumaphos Unicolophos Di-(p-Chlorobenzoyl) Peroxide Unavailable X X X Di-(p-Chlorobenzoyl) Peroxide Unavailable X X X Dinsoluble DPT Dibutyl Phenol Diuron Insoluble X X X Dinsoluble X X X Dinitrobenzene Insoluble X X X Dinitrobenzene Insoluble X X X ESF Endosulfane X X X ESF Endosulfane Insoluble X X X X X ESF Endosulfane X X X X X X X X X X X X X X X X X X X					X
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DZP Di-(p-Chlorobenzoyl) Peroxide Unavailable X X X DIU Diuron Diutyl Phenol Insoluble X X X X X DINITY Phenol Diuron Insoluble X X X X X X X X X X X X X X X X X X X					-
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EST Endosulfane Insoluble X					\
EST Endosulfane Insoluble X				^	\
EST Endosulfane Insoluble X				, .	^
ETI Ethylene Imine FFX Ferric Fluoride FSA Fluosulfonic Acid KPE Kepone LSA Lead Stearate LSA Lead Stearate MHH Maleic Hydrazide MRT Mercuric Thiocyanate MRK Methylisopropenyl Ketone METHylcyclopentadienyl Manganese MRX Mirex MRX Mirex NRD Naled NRLD Naled NRLD Nickel Hydroxide NIL Nitralin NAA Nitrotriacetic Acid, Salts NAA Nitrotriacetic Acid, Salts NAA (Anitroaniline OXA Oxalic Acid PPP Phosphorous Pentasulfide PCB Polychlorinated Biphenyl BO Potassium Binoxalate SML Sodium Methylate SML Sodium Methylate SML Sodium Methylate SMR Strychnine TED Tetramethyldithiopyrophosphate TED Tetramethyldithiopyrophosphate TED Tetramethyldithiopyrophosphate TED Tetramethyldithiopyrophosphate TCA 2,4,5-Trichlorophenoxyacetic Acid TOT Toluene 2,4-Diisocyanate URS Uranium Peroxide UNAN Uranium Nitrate Analytical X X X X X X X X X X X X X X X X X X X					\
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OXA Oxalic Acid					X
PPP Phosphorous Pentasulfide POB Polychlorinated Biphenyl Unavailable SML Sodium Methylate Water React X SML Sodium Methylate Water React X STR Strychnine Insoluble X X X STR Strychnine Insoluble X X X X X X X X X X X X X X X X X X X	NAL	4-Nitroaniline	Insoluble	X	
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URP Uranium Peroxide Insoluble X X UAN Uranium Nitrate Analytical X X URS Uranium Sulfate Analytical X X VSF Varadyl Sulfate Analytical - X	TP0	Tris(Azidinyl)Phosphine Oxide	Unavailable	X	X
URP Uranium Peroxide Insoluble X X UAN Uranium Nitrate Analytical X X URS Uranium Sulfate Analytical X X VSF Varadyl Sulfate Analytical X	TDT	Toluene 2,4-Diisocyanate	Slow Water	-	X
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VSF Vanadyl Sulfate Analytical - X		Uranium Sulfate			X
	VSF	Vanadyl Sulfate		-	X
VCM Vinyl Chloride Unavailable - X	VCM		Unavailable	-	X X
					X
ZCN Zinc Cyanide Insoluble X X	ZCN				X
ZCT Zîrconium Tetrachloride Water React. X -	ZCT	Zirconium Tetrachloride	Water React.	X	-

recommended exposure levels are established. Therefore, until the data becomes available, fully encapsulating suits should be used for spills of these 44 chemicals. This dictates the need for additional materials compatibility studies for these chemicals.

5.2.4 Alternative Decon Agents

The decon method employed in this work was a cold water, detergent solution wash. Chemicals ineffectively decontaminated are listed in Table 11. Alternate decon materials and methods suggested for future R&D investigation are:

- 1. Use of hot water and strong detergent solution wash cycle.
- 2. Investigate longer and multiple wash cycle times.
- 3. Use of decon agents other than the detergent:
 - (a) Acid type chemicals; use 5% basic solution wash cycle.
 - (b) Basic type chemical; use 5% weak acid wash cycle.
- 4. Specialized techniques such as steaming, vacuuming, autoclaving, extended airing, etc.

An acceptable decon agent is one which does not in itself present a toxic hazard or else additional deconning would be necessary. An acceptable decon agent and method must be inexpensive and implemented without logistic problems. An examination of the general class of chemicals found to require additional decon studies shows they are closely related to chemical classes which were ultimately found to cause a breakthrough such as aldehydes, amines, aromatic types, etc. This can be interpreted that these chemicals have penetrated somewhat into the matrix of the butyl. Depth of penetration apparently is not severe; however, experience suggests that these chemicals would eventually defy rather mild methods of deconning.

Depending upon the chemical and its penetration into the matrix of the fabric, there is the possibility that deconning may be obtained after an extended time in the open air.

No edge effects resulted with the circular butyl speciments decontaminated because the periphery of the specimen was outside the chemically exposed surface. In conducting seam tests, edge effects were present and this presents another matter. Some chemicals which were effectively decontaminated from the butyl

rubber with no edge effects but which were found to fail with butyl rubber specimens having seamed edge/adhesive present are:

- 1. ACY Acetone Cyanohydrin
- 2. CHN Cyclohexanol
- 3. NTT p-Nitrotoluene
- 4. CBN 4-ChlorobutryInitrile

Although these seamed specimens passed the penetration test, edge effects can present other decon considerations to reckon with. This suggests that perhaps future testing be more expediently conducted including seam portions of the fabric.

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APPENDIX A

LEVELS OF PROTECTION

(References listed by number in the tables are included in the Bibliography;

Legend:

- CHRIS Code CN - Chemical Name

- State S

- Water Reactive WR

- Flammable

- Vapor Pressure (mm Hg @ 37.8°C)
- Time Weighted Average ۷P

STEL - Short Term Exposure Limit

- Inhalation

- Skin Absorption SA

IR - Irritant

- Particulate Filter PF CC - Chemical Canister

Air SupplySealed Suit AS SS NS - Nonsealed Suit

- Butyl В

- Polycarbonate

DNA - Data Not Available

- TLV or Hazard Rating Shown in ()

Parentheses Based on Decomposition

Products

INTRODUCTION

Strike Team personnel could be exposed to hazardous chemicals in liquid, solid and/or vapor states during cleanup of spills. Protective clothing and respiratory protection could be required, and the degree of protection will depend upon the chemical involved and the nature of that chemical. Since Strike Team members could be in direct contact with hazardous liquids, solids and/or vapors during initial entry into a spill area, a conservative or "safe side" approach was used in recommending levels of protection. As cleanup progresses and the degree of hazard is better characterized, the On-Site Coordinator may decide that a lesser level of protection is acceptable.

Attempts were made to assign levels of protection for each of the 985 chemicals listed in CHRIS. A toxic hazard rating developed by Sax (Sax, 1975) was used to estimate the acute exposure effects due to inhalation, skin absorption or irritation (Table A-1). These estimates were used in recommending the type of respiratory protection and clothing required for the various chemicals (Table A-2). The compatibility of butyl rubber and polycarbonate, the materials used in the Hazardous Chemical Protective Clothing Outfit (HCPCO), with the chemicals were determined from a literature search.

This Appendix contains information which will be useful for decision making by the On-Site Coordinator. A few examples are given to assist the user in interpreting the information in the Appendix.

Acetone is a liquid (S) at normal temperatures and is not water reactive (WR). It is flammable (F) and has a moderate vapor pressure (VP) of 37 mm Hg. The permissible 8 hr exposure level (TWA) is 1000 ppm, and the short term exposure level (STEL) is 1250 ppm. It poses a moderate inhalation (IN) and skin absorption (SA) hazard but is not particularly irritating (IR). A chemical cartridge respirator (CC) or an air supply (AS) is recommended, but a sealed suit is not required (NS). Butyl rubber (B) is compatible with acetone, but polycarbonate (P) may have a limited lifetime before clouding occurs.

Anhydrous ammonia is a gas (S) at normal temperatures and is water reactive (WR). It is flammable (F) and exhibits an extremely high vapor pressure (VP). The permissible 8 hr exposure level (TWA) is 25 ppm and the short term exposure

TABLE A-1 - TOXIC HAZARD RATINGS ACCORDING TO SAX

Rating	Effect	Comments
0	None	(a) No harm under any conditions;
		(b) Harmful only under unusual conditions or overwhelming dosage.
1	Slight	Causes readily reversible changes which disappear after end of exposure.
2	Moderate	May involve both inneversible and reversible changes not severe enough to cause death or permanent injury.
3	High	May cause death or permanent injury after very short exposure to small quantities.
U	Unknown	No information on humans considered valid by authors.

TABLE A-2 - GUIDELINES FOR SELECTION OF PERSONAL PROTECTION EQUIPMENT

	Clothing	None required "	Nonsealed Nonsealed	 	Nonsealed Nonsealed	FES ¹ FES ¹
	Respiratory	None required "	Particulate 	Particulate &		Air supply ³
Response	unds Clothing	None required "	Nonsealed Nonsealed	!!!	Nonsealed Nonsealed	FES ¹ FES ¹
	eas & Liquids Respiratory	None required "	Chemical filter ²	Chemical filter	- Acadas	Air supply ³
9. 7. 7.	Entry	Inhalation Skin Absorption Irritation	Inhalation Skin Absorption Irritation	Inhalation	Skin Absorption Irritation	Inhalation Skin Absorption Irritation
4 4 4 6	Rating	0	-	5		m
				Δ_Λ		

¹Fully encapsulating sealed suit indicates need for air supply 2Contaminants in gaseous form at STP indicates need for air supply 3Carcinogens indicate need for air supply limit (STEL) is 35 ppm. It is a severe inhalation hazard (IN) and irritant (IR). An air supply or self-contained breathing apparatus (AS) and a sealed suit (SS) are recommended. Both butyl rubber and polycarbonate are compatible with the chemical.

Ammonia bisulfite is a solid (S) at normal temperatures and is water reactive (WR). No data were available on the flammability or vapor pressure of the chemical. The permissible 8 hr exposure level (TWA) is 5 ppm based on the chemical-water reaction product of SO₂. Sulfur dioxide is a severe inhalation (IN) hazard and irritant (IR). An air supply or self-contained breathing apparatus (AS) and a sealed suit (SS) are recommended. No data were available on the compatibility of butyl rubber and polycarbonate with sodium bisulfite, however these materials are compatible with SO₂.

It was, of course, impossible to assign levels of protection for all 985 of the chemicals on the CHRIS list. Additional research is being conducted to determine the hazards posed by the various chemicals, the required levels of protection and the type of materials to be used in spill clean up. As this information becomes available, it is anticipated that the information contained herein will be updated. Meanwhile, On-Site Coordinators and Strike Team personnel should find this information valuable and should become acquainted with the usage of this information.

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Comparison Com						,	1 <u>.</u> (e	æ .	Š	CKIC		=	E S C		3	_		
Example District Example Exa	ن ت	×.	S		<u> </u>	<u></u> -	2	S)*	347 N	\$ 5	ع ام			3 0		+		
Second Column Second Colum			Ì	-				1		5	=	<u>.</u>	+-	+	┿	Т		
Example Designation Control	2	Benzyl Chloride	<u> </u>			22	- 11511		~ ~	<u>س</u>	۳,					2 3	Yes	Peacts with water to form HCl
Exposition Control C		Benzyl a-Butyl Pathalate Benzyl m-Butyl Pathalate				×0.25	DIA NO		, m		, m					₹ ¥	2	No toxicity _sta found
Fig. 10 Fig.						1					1	- -	+	+	-	-	-	
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Execution Control Co	, <u></u>	Beryllium Fluoride			<u> </u>	<0.25 D	.002*(Be)		<u>e</u>	1	<u> </u>	<u> </u>				<u>e</u>	es	loxicity rating as beryillum
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Bery	2 2 2	Beryllium, Metallic				< 0.25 0	200	0.025	Œ	٠	Œ	•				Yes	7es	rating as
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Brownies S			+	Ť	1]	3	1	1,	4	+	+	<u> </u> ;	- 3	3
Browling Porticition 1	BAC	Boron Tribromide				-0.25 -2.25	<u>.</u>	3.0	 5:E	<u>3</u> ,	25	٠.				Tes	Ke S	rating
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	į	Section 2	_	_		262			-		~	,				Yes	- A	
Secondarian Commission Co	5			+	+	;	;	.	•	1	+	+		+	+	-		
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Butation inhibited G No Ves > 760 1000 1250 2 - 2 - X X - Yes Ves Yes Actaon 1.3-Butanediol - X X - Yes Yes Yes - - Yes Yes Yes - - - - Yes Yes Yes - - - X X - - Yes Yes Yes -	2	פרעכות	رم د		<u> </u>	< 0.25	š		~ ~		'					<u>-</u>	2	-
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1,4-Butenedio 1 No Yes 24 150 200 2 1 1 1 1 1 1 1 1 1	3					<u> </u>	 §		·	,	-					<u> </u>	<u>!</u>	
No. Ves. St.	2						SK4		3	>	>					7es	Yes	
Sec-Butyl Accidite 1 No Yes 51 200 250 2 1 -	2	n-Butyl Acetate	_	-	Ş	22	951	200	7	-	-	١,				ž.	₽.	
Iso-Butyl Acrylate	BIA	Sec-Buly Acetate tert-Buty Acetate			S :	<u>.</u>	88	550	~ ~			•				Yes	(PVC)	
n-Butyl Acryla, n-Butyl Acryla, n-Butyl Acryla, n-Butyl Acryla, n-Butyl Acryla, n-Butyl Alcohol	\$ 3	Iso-Butyl Acrylate			S S	3	N W	ney		. ,	(2)					Yes	Yes	Toxicity as BTC
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terr-Butyl Alctiol 1 No Yes 2 2 1 X X X Yes Yes <th>33</th> <td>sec-Buty Alcohol</td> <td></td> <td></td> <td></td> <td></td> <td>3 5</td> <td></td> <td>۰,۰</td> <td>·</td> <td>-</td> <td></td> <td></td> <td></td> <td></td> <td>S S</td> <td><u> </u></td> <td></td>	33	sec-Buty Alcohol					3 5		۰,۰	·	-					S S	<u> </u>	
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tert-Butyl Hydroperoxide 1 No. Yes ~ 30 DNA - - - X - Yes <	<u>.</u>			-	- ;		•			-		 .						
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n-Butyl Methacrylate I i No Yes 10 DMA 2 - 2 - X X X - X Yes in-Bytyl Amine		n-Buty) Mercaptan		- ·•			9.5		7	,	7	,				2		[c] mondan
	5	n-Butyl Meinecrylate	 			2 8	A PER		~ ~	٠,٠	~ ~	1	_			Yes	e e	

		ROTES		Asphyriant			Toxic rating as chlorosilanes,	174 do 1861	Toxic rating as aldehydes	Toxic rating as aldehydes		Toxic rating as cadmium compound Suspected carcinogen	Toxic rating as cadmium compound Susperted carcinomen		* * * * * * * * * * * * * * * * * * * *	Toxic rating as cadmium compound		Recognized carcinogen. Toxic	rating as arsenic compound Toxic rating as arsenic cmpd.	Toxic rating as calcium cmpd.	Toxic rating as calcium cmpd.	Toxic reting as chromium cmpd. Recognized carcinogen	rating as	rating as	ioxic rating as calcium cupu.	rati. d.	Toxic : as nitrates			Tox: cing & TLV as PH3	Pyrophoric Gives off flammable vapors when basted		Eye, skin irritant [9] Poisonous via skin	
MATERIAL Compati-	BILITY [10	ط	2 :	2 ×	Š	Yes	£	£	£	2	Yes	Yes .	les	Yes	Yes	, es		Yes Yes		Yes	Yes	ž Š	Yes	<u>ن</u> ور	ž ž	Yes	S &	i s		<u> </u>		, se	š š	Yes
MATE CONTRACT		В	Yes	<u>د</u> و	ž.	ž	Yes	Yes.	Yes	Yes.	Yes Yes	Ze Z	Se.	Yes	Yes	Ş		Yes		Yes	, es	. ž	, es	es .	ž š	Yes	e s	i S	,	ē š		Yes	e e	Yes
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	5	!	sec-Butylamine	Buty lene	1,2-Butylene Oxide	0-Sert-Rutvlohonol	Butylti ichlorosilane	1,4-Butynediol	iso-Butylvaldehyde	n-Butraldehyde	n-ButyFic Acid	Cadmium Acetate	Cadmium Bromide		Cadmium Fluoborate Cadmium Nitrate	Cadmium Oxide		Cadalum Sulfate	Calcium Arsenite	Calcium Carbide	Caldium Chlorate	Calcium Chromate	Calcium Cyanide	Calcium Pydenxide	Calcium Hypochlorite	Calcium, Metallic	Calcium Oxide	Calcium Peroxide	Calcium Phosohate	Calcium Phosphide	Calcium Resinate Camphene	Camphor (0il)	Captan	Carbacy
	ن		BTL	A	2	0.10	BCS	ATO -	8	BTR	8	\$5	8	9	25	Ē	}	88	CAS	800	<u>ဗ</u>	18 18	5	3	5 5	3	<u> </u>	දිදි	+	 58	# E	_	Sign	

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		NOTES	Toxic rating as phenols			By product wood turpentine [2]	12]	Toxic rating as chlorine, fluorine, fluorides	Highly toxic Powerful lachrymator TLV & toxic rating as HCl	Toxic rating & TLV as analine Organic cyanide Suspected carcinogen	Irritant Toxic rating as chloroethylether Highly toxic [8] Toxic rating as phenol	Yes Lachrymator & irritant No No (No)(PV¢) As sulfuric acid [10]	As chromic compound, recognized carcinogen	Montoxic[5];moderately irritant organic ecid[8] As cobalt compounds As bromide compounds	As cobalt compounds As fluorides	As formic acid As nitrates
RIAL ATI-	BILITY [10]	Ь	Yes	(ACE) Yes	(ACE) e 1121	Yes No Yes[12]	to Yes[Yes Yes Yes		Yes No	Yes Wo	Yes	Yes N o (%) (PV (10)	Yes	Yes	Yes	Yes
MATERIAL COMPATI-	BILI	В		Tes Yes	No[14](ACE)	Yes Yes	Yes Yes Yes	S.	No Yes Yes	% % % % % % % % % % % % % % % % % % %	No Yes	Yes No Mo(PVC) [10] No[11]	Yes	Yes	Yes	Yes
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) AIL	MG/	¥.	(5)	20 00	2	DNA 5/20*	2* DNA 0.027	0.1	DWA 0.345* (5)	(2) 75 DNA 25	5 DNA 1 [2] DNA	0.1 25 (5.0)[9] G:4	(0.05)* (0.05)* (0.05)* (0.05)*	(0.05)* DNA 0.1*(Co)	0.1*(50)	
	<u>\$</u>		1.7	> /60 > 760 > 760	2/0	5.1 20.1 <0.25	<0.25 <<0.25 <<0.25 > 760	> 760	< 0.9 0.25 30	0.26 26 1.7 346	16 180 0.66	1.6.	< 0.25	32 < 0.25 < 0.25	< 0.25	< 0.25
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	5		Carbolic 0il	Carbon Monoxide	Sarbon Tetrachloride	Carene Catechol Caustic Potash (solution)	Caustic Soda (solution) Charcoal Chlordane Chlorine	Chlorine Trifluoride	4-Chloro-o Toluidine Chloroacetophenone Chloroacetyl Chloride	p-Chloroaniline Chlorobenzene 4-Chlorobutyronitrile Chlorofo:m	Chiorohydrin (crude) Chloromethyl Methyl Ether o-Chloronitrobenzene p-Chlorophenol	Chleropicrin Chloroprene Chlorosulfonic Acid p-Chlorotoluene	Chromic Acetate Garonic Anhydride Chromic Sulfate Chromous Chloride	Chromyl Chloride Citric Acid Cobalt Acetatc(ous) Cobalt Bromide(ous)	Cobalt Chloride(ous) Cobalt Fluoride(ous)	Cobalt Formate(ous) Cobalt Mitrate(ous)
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		NOTES	Primarily a respiratory threat	Téxicity based on ether content As copper cmpds;TMA as 0,2* as	As arsenic compounds, recog-	As bromides & copper compounds As bromides & copper compounds	copper compounds cyanides fluoride :s fluoroborate copper "x_pounds	TLV as copper cmpd, additive to	As copper cmpd,toxic_rating bTLV	As copper cmpd, toxic rating As copper 6 oxalate compounds As copper compounds As copper compounds	As ammonia; TLV as copper As copper compounds, toxic rating 8 TLV Nontoxic [6] Cholinesterace inhibitor [5]	Recognized carcinogen of skin;	experimental carcinogen of lung	Strong irritant to tissues, toxic [5] As cyanides,toxic rating & TLV	As cyanides As cyanides As hydrochloric acid	Suspected Carcinogen
_	<u></u>		Prime	Texic As co	As all	As b	AS S S S S S S S S S S S S S S S S S S	TLV		\$ \$ \$ \$ 8 \$ 2 2 2	As contraction (Cho)	Reco	expe	Stro toxi	As As	Susp
MATERIAL	WANI- BILITY (10)	۵	Yes	ž š	Yes	Yes	Yes Yes Yes	Yes	Yes	Yes Yes	Yes	Yes	¥ Ye	Yes Yes	Yes Yes Yes Yes[12]	Yes
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	¥	_	, O,	22	ş	222	2222	운	₹2 1	2222	22 22	윤	윤 , 오	22, 2	2 % es	2
	S	-	۷	N L	8	SSS	NN 1N	\ <u>\</u>	~~~	nnnn	NN JN	_	۔۔۔۔۔۔۔۔۔۔۔۔۔۔۔۔۔۔۔۔۔۔۔۔۔۔۔۔۔۔۔۔۔۔۔۔۔۔	v	L/6 L/6	1/8
	Š		Cobait Sulfamate(ous) Cobait Sulfate(ous)	Collodion Copper Acetic(ic)	Cupper Acetoarsenite(ic)	Copper Arsenite(ic) Copper Browide(ic) Copper Browide(ous)	Copper Chloride(ic) Copper Cyanide(ous) Capper fluoborate(ic) Copper formate(ic)		Copper Iodide(ous) Copper Lactate(ic) Copper Mapthenate(ic)	Copper Nitrate(ic) Copper Oxalate(ic) Copper Subacetate(ic) Copper Sulfate(ic)	Copper Sulfate(ic) Amnonated Copper Tartrate(ic) Corn Syrup Coumaphos	Cresote, Coal Tar	Cresols Cresyl Glycidyl Ether Crotonaldehyde	Cumene Cumene Hydroperoxide Cupriethylenediamine Solution Cyanoacetic Acid	Cyanogen Cyanogen Browlde Cyanogen Chloride Cyclohexane	Cyclohexanol
_			SS	000	-	200 800 800 800 800	3333		\$ 5 5 5 \$ 2 5 5	S C C C C C C C C C C C C C C C C C C C	\$5 53 \$5 38	133				CHM Cyc

1	CN S NR F YP MS/P	£	g.	95	7	NS (S		ILV (PPH OR MS/H3)*	SAX N	SAX-TUXIC FEZARU PATING IN SA	2	- 8	LEMELS OF PROTECTION RESPIRATORY CLOTHING	15 E	CLOTHING	<u>_</u> 2	MATERIAL COPPATI- BILITY [10]	- L C	
1		200		5	+-				<u></u>	5 '	<u>د</u> (=	3	2	3	2	+		WIES
(3) (3) (3) (-1) (2) (2) (3) (3) (3) (3) (3) (4) (4) (4) (4) (4) (4) (4) (4) (4) (4	Cyclopendate L No Yes 520 10 Cyclopendate L No Yes 520 0MA Cyclopropane G No Yes 720 0MA Cyclopropane L No Yes 73.5 0MA	No Yes 22 No Yes 520 No Yes 720 No Yes 3.5	Yes 520 Yes 520 Yes 720	3.5 3.5 3.5		00 00 00 00 00 00 00 00 00 00 00 00 00			~~@	m	%		· ×××	***	× 1 1 1	1888			, toxic rating
1 - 2	2,4-D Esters L No - <0.25 DNA	No - <0.25	- <0.25	<0.25	+-	PKA										1		Moderate	
(3) (3) x	Dalapon - L No - DNA DNO DNO - DNA	No - No (0.25		<0.25		¥ 8			7		2	, ×	×	×ı	1 1	××			erbicide
(3) (3) (4) (5) (5) (7) (8) (8) (8) (8) (1) (9) (1) (1) (1) (1) (1) (1) (1) (1) (1) (1	No Yes <0.25	No Yes <0.25	Yes <0.25	<0.25		_				,	·	: ×		ı					In water shalation E impostium
(3) (1) (2) (2) (3) (4) (4) (4) (4) (4) (4) (4) (4) (4) (4	Decaborane S Yes Yes 0.18 0.3*	Yes Yes 0.18	Yes 0.18	0.18		0.3		•6.0	3	'	<u>e</u>	-		×	×	+	+	+-	ng as boron hydride
(3) (3) (3)	Decalparination	No Yes 0.27 No Yes 1.7 No Yes 2.7	Yes 0.27 Yes 1.7 Yes 2.7	0.27		DRA DRA			∙	·ŝ·	~	111	×IX	×××	· × ·				lty,skin absorp. Narcutic. Toxicity Axene
(3) (3) (3) (3) (4) (4) (4) (4) (4) (4) (4) (4) (4) (4	Decyl Alcohol	No Yes 0.02	Yes 0.02	0.05		¥			>	2	3		•		ŀ	-	+	Ť	eriments indicate
(3) (2)	n-Decy1benzene L No Yes ~0.02 DNA Dumeton t. No Yes <0.25 0.0087	No Yes ~0.02 No Yes <0.25	Yes ~0.02 Yes <0.25	\$0.05 \$0.25		DNA 0.0087			- 6	⊅ (£	۰ د	ı	,	×	×				ly. M as marathion
(2) - (2) -	L No No <0.25	No <0.25	No <0.25	<0.25		SHA DIE			_		3	ı	'		. 1				ace inhibitor.
(2) - (1) - x x - x no	DI-{2-Ethylhexyl} Phosphoric Acid L No Yes 0.25 DNA DI-{p-Chlorobenzoyl} Peroxide L No Yes 0.25 DNA	No Yes 0.25 No Yes 0.25	Yes 0.25 Yes 0.25	0.25	<u> </u>	28.28						٠.	1.,	· ×	××	┼	+	+	ity [8]
3 - 3 -	Di-n-Butyl Ether L No Yes 0.25 DWA L No Yes 13.5 DWA	No Yes 0.25 No Yes 13.5	Yes 0.25 Yes 13.5	0.25		DNA			<u> </u>		33	1 1	1 34	××	1 1		22	Organic pe Toxic rati	oxide [8] 19 as anyl alcohol 19 as phthalic acid
2 - 2 - x x - x ves ves ves (2) (2) (2) x x x - x x - ves ves (3) (3) x x x - x ves ves (45) (2) (2) x x x - x x - x ves (45) (3) (3) x x x - x ves (45) (3) (3) x x x - x x - x ves (45) (3) (3) x x x - x x - x ves ves (45) (3) (3) x x x - x x ves ves ves ves ves 2 - 1 - x x x - x ves ves ves 2 - 2 - 1 - x x x - x ves ves ves 2 - 2 - 1 - x x x - x ves ves ves 2 - 2 - 1 - x x x - x ves ves ves	Di-n-Butyl Ketone	No Yes . No Yes 5.8	Yes 5.8 Yes 5.8	5.8 5.8	5.8	DE BEE	,		m =	, 3	ma	1	× 1 1	1 ××	1 ××	+		+	ketone Dressure irritatinu
(2) (2) (2) X X - Yes Yes Yes (2) (3) (3) X X - X Yes Mo(AGE) (3) (3) (3) X X - X Yes Mo(AGE) 2 - 2 X X X - X - X Yes	Diacetone Alcohol L No Yes 3.3 50	No Yes 3.3	Yes 3.3	3.3		95			2	, ,	~	1	×	×	,				Je Je
(2) (2) (2) X X X X Yes Mo(AE) (3) (3) (3) X X X - X MO(AE) 2 - 2 X X X MO(AE) 2 - 2 X X X MO(AE) 1 - 1 X MO(AE) 1 - 1 X MO(AE) 2 - 2 X MO(AE) 2 - 2 X MO(AE) 3 X MO(AE) 4 MO(AE) 5 - 1 - X MO(AE) 5 - 1 - X MO(AE) 6 - 1 - X MO(AE) 7 MO(AE) 7 MO(AE)	Diazinon L No Yes <0.25 0.1*	No Yes <0.25	Yes <0.25	<0.25	L	0.14		0.3*	<u>e</u>	Ξ		-	-	×	×	1.	┿	+	g as parathion.
(3) (3) (3)	Underly Peroxide S No Yes ~0.003 *	No Yes ~0.003	Yes ~0.003	-6.00 -0.01		0.403			~(<u>S</u>)	(2)	%	× ·	××	·×					ace inhibitor g as butyl alcohol
2 - 2 X - 1	S No Yes <	No Yes <0.25	Yes <0.25	<0.25		(5)		(01)	Ξ	Ĉ	(3)	_	•	×	×	,	<u> </u>		g & TLV as phenol
2 2 X X I X X I X X X X X X X X X X X X	Ofchlobenti S No - DNA B"	No - DHA	PAR	¥ 8	¥ 8	<u> </u>						,			<u> </u>			Probably m	derately toxic, pest control [5]
2 2 3 3 X X Wes Yes Yes Yes Yes Yes Yes Yes Yes Yes Y	No - 0NA - 7es < 0.25	No - DNA - Tes < 0.25	- ONA Yes <0.25	ONA < 0.25	DNA 0.25			-	2	,	8	< × :	· × ·	ı ı×		- ××		Fungicide	halation & ingestign.
2 - 1 - X - X Ves Yes Yes Yes No 2 2 2 2 1 - X X - X Wolf4]	0-Dichlorobenzene L No Yes 3.2 50 p-Dichlorobenzene S No Yes 4.8 75 Dichlorobutene L No Yes 7.4 DNA	Mo Yes 3.2 Mo Yes 4.8 Mo Yes 7.4	Yes 3.2 Yes 4.8 Yes 7.4	3.2	ļ	05 % 25			2-3	113	m-3	1 24 1	1	× 1 ×	× 1.×	+	_	- -	eves skin causes
2 2 2 1 - X X - X Woll4]	Dichlorodifluoromethane 6 No No >760 1000	No No >760	No >760	>760	92	0001		1250	_	,	•		×]
3 3 3	1,1-Dichlorgethane	No Yes	Yes		CKA	Y NO			2		-	-	×	*	+		E	Moderately	toxic [5]
	Dichloroethylether 1.2 5 1.2 5 1.2-01thloroethylene 1.2 5 1.2 5 1.2-01thloroethylene 1.3 200 1.3 200 1.3 5 1	No Yes 432	Yes 1.2	1.2	432	 800 800 800 800 800 800 800 800 80		280	600	600	е-c		(×	**			<u> </u>		

			-	-	} 			ļ ;							Æ	MATERIAL	
٠,	3	·	9		9	TLV (PPM OR	¥ .	-XX-V	SAX-TOXIC		E G	RESPIRATORY CLOTHING	MOJEC	THIN'S	8 2	COMPANIA- Bill ITY Cool	
ر 	5	,	 É		 =	Z Z	SIE	2	55	~	H-	υ υ	SA	€2	6 0	Ь	NOTES
ਕ੍ਰੇ	2,4-Bichlarophenol	S	2	Yes	0.35	DNA		(2)	(3)	(3)		× -	*	,	ş Ke	Yes	Toxic rating as cnlorinated phenols
288	2,4-Dichlorophenoxyacetic Acid 1,2-Dichloropropane Dichloropropene	رد د 	222	Yes <	<0.25 93 026	10* 75 DNA	110	282	101	~~~	× 1 1	× ·×	1×1	×ı×	Yes Yes	₹ ₹ S	
25.55	Dichlorovos Dicyclopentadiene Dieldrin Diethanolamine	1001	2222	Yes (6	6.5 <0.25 <0.01	0.1 75 0.25* DNA	0.3	e	e 1		1 1 1	111×	×××:	1118	Yes	Yes Yes No	Highly toxic [8] Very toxic [8] Low toxicity
	Diethyl Carbonate Diethyl Fhthalate Diethylamine Diethylbenzene		2222	Yes Yes Yes	21 <0.25 39 2.6	25 25 26 27 28	*01	22 - 2	1118	2000	1 1 1 1	××××	××××	***	Yes Yes	Yes No No	
20 X 20	Diethylene Glycol Diethylene Glycol Dimethyl Ether Diethylene Glycol Monobutyl Ether Diethylene Glycol Monobutyl Ether Acetate		2222	Yes Yes Yes	0.01 6.0 0.17 0.02	DNA DNA DNA		~ମୃତ୍ରନ୍ତ		-888	1 1 1 4	××××	××××	××××	Yes Yes	Yes No	As glycol As glycol As glycol
06F 06M 0ET 0EZ	Diethylene Glycol Monoethyl Ether Diethylene Glycol Monomethyl Ether Diethylenetriamine Diethyl Zinc		No No Yes	Yes (0.6 0.7 1.1 36.1	DNIA DNIA		(2) (2) (3)	2	(E) (E)	1 1 1	×××	x - x - x - Fire	e Suit	4es Ses	No (ACE) [12]	As glycol As glycol Decomposes violently in water, ignites spontaneously in air
OFE DFA DHP DIK	1,1 Difluoroethane Difluorophosphoric Acid Anhydrous Diehepty!phthalate Disobutyl Ketone	9	No No No	Yes >:	>760 58 <0.25 5.0	200 DNA DNA 25		(3) - 2		1 (2) 2	1 1 1 1	×IIX	×× ×	× ·××	Yes Yes	Yes No	Toxic rating as fluorides Toxic rating as DID
2 4 6 6 6 6 6 6	Ofisobutylcarbinol Disobutylene Disodecylphthalate Disopropanolamine	1/S	S & S	Yes Yes Yes	3 <0.25 <0.02	DNA DNA DNA			1 ;	88		× × 1×	× 111	* ***	Yes Yes	Yes Yes	Animal experiments indicate low toxicity Marcotic in high concentrations Low toxicity [5]
DIN	Diisopropylamine Diisopropylbenzene Hydroperoxide Dimethyl Ether Dimethyl Sulfate	9-	2 - 2 ×	Yes Yes Yes	136 <0.25 >760 1.6	5.0 DNA 0.1		325	2326	03-E		×ı×ı	××××	× 1 × 1	Yes Yes	de Yes	Strong oxidizer,proably toxic
DAC DAC	Dimethyl Sulfide Dimethyl Sulfoxide Dimethyl Terephthalate Dimethylacetamide	1 181	9 9 9 9	Yes >77	760 1.5 0.01 4.2	DIG BRA	15	3 - 6	3 2 €	5 % C	1 1×1		× 11×	××	Yes Yes	Yes Yes	Probably highly toxic, see sulfides as irritant Dust not harmful
¥ 9 5 5	Dimethylamine Dimethyldichlorosilane Dimethylformamide Dimethylhexane Dihydroperoxide, Wet		No No No	Yes >7 Yes 2 Yes 0 Yes <0	760 240 0.8 0.25	DNA DNA DNA	50	2(3)2	110	2 <u>0</u> 2	1 1 1 1	× (×)	****	× 1× 1	Yes Yes Yes	222	Toxic rating as chlorosilanes Organic peroxides, irritant to skin å tissue å eyes

		r	-	-	ř					-				į	-	MATERIA	
						TLV (PPH OR	8	Š	SAX-TOXIC		S	ELS OF	LENELS OF PROTECTION	Æ)		COPPATI-	
J	3	s	¥	ш.	9	MG/M3)*	3)•	HAZAR	HAZARD RATING		8	ESPIRATORY		CLOTHING	Н	ILITY (10	
,		_				F.	STE	Z.	S.	æ	kt.	ម	SS S	8	æ	۵.	NOTES
*	1,1 Dimethylhydrazine	٠.	, se	, kes	√5 0#4	0.5	1.0	3	٤.	3		. ,	× 1	* 1	, es	2	Experimental carcinogen 10 ppm in food (0 in milk)[5]
	Dimethylpolysfloxane				, y	¥							Fire	a Suit	يد		irritating to eyes Pyrophoric
35	Unweinglance 2,4-Dinitroaniline	J.S.	No.	iğ.	<0.25	A A		2	•	7	_	_		-	-	-	
ZMG	Dinitrobenzene				60.1 0.25 0.25	15/1*	6.0	ომ	 ო@			, ,		 	Yes	Yes	Slight fire hazard As dimitrobenzeme,toxic rating
ž Š	Dintrocresols. 2.4-Dintrophenol	S	22	ž ž	<0.01 <0.25 0.25	D.2* [9] DNA		r (E)		- -	1 1				Yes		As 2-3 dinitrophenal
										-	-		-	-	-	-+	
58	2,4-Dinitrotoluene Dioctyl Adibate	ب ہر	No ON	Yes	<0.01 . < 0.25	.5* [9] DNA		м	m				~ ·	. × × .	Yes Yes	ž Š	Low toxicity[5], irritating
88	Diociyl Phthalate Diociyl Sodium Sulfosuccinate		20	ž 2	<0.25 <0.25	. NA		1		o-		**	<u> </u>	· · ×	Yes	es No	Low Vapor
25	1,4-Dioxane Dipentene		ž ož	ž š	5.7	05 S		m	2 .	22		, ×,	× 1	***	2 ¥ €	Yes	Vapor toxic by inhalation[8]:
¥ }	Diphenylether			<u>د</u> و		[6] •0[-	m	m	1		٠ ،	· ×	· ·		Š	low toxicity [5]
§	Utpneny Lamine	7	_1	+							-	+	-	-{,	-	-	
0 K 9 3	Diphenyldichlorosilane Diphenylmethine Disocyanate Dipropylene Glycol	אראר	Slow No or	ves ves res	6.05.09 0.05.09 0.54.09	.02 [5] .08 [5] .08		m =m	ı ====================================	m ⊃m		× :	×× 1×	1 1 1 1 K < 1 ×	4 × 4 × 8 × 8 × 8 × 8 × 8 × 8 × 8 × 8 ×	s s x	Toxic.strong irritant [5] "-derately toxic by ingestion[5] H:gnly toxic contact herbicide & desiccant; see peraquat
ij	Distillates, Flashed Feed Stocks	-	2	, es	93	500/30 min				-	}- ···	-	-	<u></u>	2		Mixed hydrocarbon-low toxicity
888 818	Distillates, Straight Run Disulfton		22	. Yes	8 8								××	× .	2		Same as DFF Insecticide, cholinesterace in- hibitor [6]
DIU	Diuron	S	No			10*		Þ	3		-	-	×	<u>'</u>			Shin Jerifehir hydrolytento di-
ğ	Dodecano:	S	№	Yes :	0.03	DWA		5	 D	 >>	- ×		,	_		ž.	Low toxicity[5]; animal experiments indicate low toxicity[8]
8	Dodecene	<u> </u>	₽	Yes	0.56 200 [9]	[6] 00		5	· ->	 >		×		× 1			Probably irritant & narcotic in high conc. [5]
88	1-Dodecene Dodecki Benzene Sulfmir Arid	N ا	 2 2	Yes	0.56	DNA					 .×	× 1		× ×	ě	ق ق	Same as above [5] Biodeg.detergent;strong org.acid
88	Dodecyl Benzene Sulfonic Acid, Sodium Salt Dodecyl Selfene Distribution Salt				: 0.25	DNA			- ·		 ×						
28	Dodecyl Sulfate, Magnesium Salt Dodecyl Sulfate, Sodium Salt	٦٧		. ,	<0.25 <0.25	DNA DNA						 ··					Irritates eyes & skin
28	Dodecyl Sulfate, Triethanolamine Salt Dodecylbenzene	-	- K	ves[5]	< 0.25 < 0.25	DNA		Э	>			+ -	-	, ×			Irritates eyes & skin Moderate y toxic by ingestion [5], see benzene [5]
8§	Dodecylbenzene Sulfonic Acid, Calcium Salt Dodecylbenzene Sulfonic Acid, Isopropylamine Salt	- v		. ke	0.251	DNA				······································					×		Similar to DSA
855.E8	Dodecylbenzene Sulfonic Acid, Triethanolamine Sult Dodecyltrichlorosilane Doutherm Dursban	۱ د د د	S S	res .	< 0.25 < 0.03 ~ 0.07	DNA DNA DNA DNA		G G	(3)	<u>e</u>		× 1 1 1		1 × 1 ×		s Yes	Based on triethanolamine As silanes;irritant, toxic mtl. Liquid heating transfer smdia As parathion; cholinesterace inhibitor
		1				 								ļ		-	

		MOTES	insecticide;	nsecticide		As parathions; cholines terace	Low toxicity[5];most glycols	animal experiments	Burns eyes, irritates skim, surfactant	Irritates skinå eyes, surfactant Burns eyes, irritates skin	eyes, irritates skin	Burns eyes, irritates skin			toxicity[5];no general state	Irritating to eyes & mucous mem-	יור ווו אואו רשעריני		[5]; haraful if	Moderately toxic [5] Readily polymerizes; mod toxic £ irritating in animal expt.[5]	Varcotic high conc. [8] Corrosive, highly toxic[8],	eyes	[5] ing & toxic,	pyrophoric, fire suit Pyrophoric,	
			Highly toxic insecticide;	As aldrins; insecticide	Mot harmful	As parathions	Low toxicity	Mod. toxic-in	Burns eyes, i	Irritates ski Burns eyes, 1	Burns eyes.	Burns eyes,			Low toxicity	Irritating to	[12]	[12]	Low toxicity Low toxicity	Moderately to Readily polyr	Narcotic high conc. [8] Corrosive, highly toxic	Burns skin & eyes	Powerful odor [5] Very irritating & toxic.	Pyrophoric,	
MIERIAL Oppați-	BILITY (10)	Ь		že s	Yes	Yes	Yes		<u> </u>	Yes	žě	<u>د</u> نه	9 :	2	ž ž	£	No(ACE)	#0 #0(ACE) #0	ž ę	řěs	Yes	Yes	₹	2	5 e 8
£ 8	富	B		Yes	Yes	Yes	Yes		ě	Yes	Yes	s ș	<u> </u>	<u> </u>	ž š	Yes	Yes	Tes Tes	ž š	Yes Yes	Yes	Yes	Yes Yes	Ĕ	5 % S
	COUNTING	æ		1 1			×		×	××	×	٠,	·×		××		×	× 1 × ×	××	×ı	<u>.</u>		- =	<u> </u>	
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PE PE	<u> </u>	₩.	*	××	. ,	××	1		'		<u>.</u>		< × :	-	× 1		×	***	· ×	××	*	**	××	×	H××:
7	₹	α	<u> </u>	-	<u>'</u>	× 1	*	_	× 	××	×	,	<×	'	××			× 1 ××	××	× ·	'	; ×	×		H H
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SAX-TOXIC	HYZARO RATING	55	ļ.	<u> </u>	_	<u>. E</u>	'	4	>	===	-	·		m 	- '	5		1101	<u> </u>	<u>'</u>			' '	· ·	N 101
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IIV (pess no	MS/H3)*	STEL		9													1250	900 120		2.0					125
) ALL	3	¥.	0.1*[5]	0,1*[9]	DHA	0.4.	V CMA	DNA	ž	¥8	¥.	DICA	20(z]	\$2	1000 2	DKA	1000	DKA 400 100	DNA	0.5 DWA	DRA	700 100	DNA	¥ o	00 VM
	3			<0.03	; ,	^ 760	<0.1	DMA	<0.25	<0.25	<0.25	<0.25	ري و	*	3.5	33	>760	7560 7560 413	20.7	>760	×760 5.1	5.50 5.50	3.6	1.5	8E 223 7660
	u.			Yes		Yes	Yes	Yes[5]	Yes	ž	, es	Yes	2 E	Yes	ž ž	Yes	Yes	7	Yes Yes	res	ž ž	Yes	Yes Yes	Yes	Yes
	¥	i	2	2 5	2 .	22	皇	Ş	₽	. 2	2	2	2 2	2	22	S.	Yes	Yes Yes Yes	22	Yes	, es	Yes	¥es ¥es	Yes	8 % B
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					Epicaloranyaria Epoxied Vegetable 011s		lycel	robyrao	Dodecanol	Ethnaylated Honylphenol	Etherylated Tetradecano)	Tridecanol	ite acetate	-	00	±	39	Chloroacetate Chloroformate Ether Formate	ano)	ptan :rylate	Ethyl Hitrite Ethyl Phosphanothioic Dichloride, Anhydrous	Ethyl Phosphorodichloridate Ethyl Silicate	2-Ethyl-3-Propylacrolein Ethyl Alumium Dichloride	Ethyl Aluminum Sesquichloride Ethylamine	osilane
	5		Endosu) fane				Etheny Triglycol	Feboradiland	Etherylated Dodecanol	_	<u> </u>	Ethacylated Tridecanol	Ethyl Acetoacetate	EUM) Acryl	Ethyl Alcohol Ethyl Mutamol	Ethyl Butyrate	Ethni Chloride	1111	2-Ethyl Hexanol Ethyl Lactate			Ethyl Phospi Ethyl Sillica			Ethylhenzene Ethyldichlorosilane Ethylene
1	Ų	,	2	8	2 S	68	E16	1	3	35	ē	£ :	5 3	3	35		ą	355 5	83		65	23	53	35	ESE

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STE	9			}	9		IIV (ILV (PPM OR	\$5 PK	SAX-TOXIC	٤	770	LEVELS OF PROTECTION	F P801	ECTION	و	MATERIA COMPATI- RII ITY	
(3) (3) (3) X X Tress Vess Vess	¥	<u> </u>	<u> </u>	<u> </u> 		*		SE SE	3	इ	2 2	出	2	क्ष	S	2 9	8	NOTES
35	Ethylene Cyanohydrin L Yes Yes 0.15 DNA Ethylene Othora/de L No No 2C 0.385 Ethylene Othora/de L No Yes 139 5 Ethylene Glycol L No Yes 0.25 100	Yes Yes 0.15 No No 28 No Yes 139 No Yes 0.25	Yes 0.15 No 20 Yes 139 Yes 0.25	0. 15 26 139 0.25		0.3 100	.88		(E) m m	<u> </u>	<u>@</u> ~~		×	××× i	×××			Toxic rating as CM Ingestion hazard, toxic in
35		No Yes 1.0	Yes 1.0	1.0	\dashv	VAC			=	5	5	1.	×	1.	1.		+	particulate form [8]
35	ther (No Yes 9.7 Ether (No Yes 119	No Yes 9.7 No Yes 119	Yes 9.7	9.7		D CM			5	>	5		1 1	××				skin & eyes Fire risk[5],not skin irritan
36	L No Yes 1.9	L No Yes 1.9	Yes 1.9	1.9	_	ន			•		,	-	×	×	•			Toxicity-glycol;mod. toxic[5]
35 1 2 1 1 1 2 1 1 1 2 1 1 1 2 1 1 1 2 1 1 1 2 1 1 1 2 1	fonobuty! Ether Acetate 1.	1. No Yes 0.37	Yes 0.37	0.37		DWA 200			1 1	٠ ،	-		× ×	, >	, ,			<pre>Low toxicity[5], irritates skin & eyes</pre>
75	23 23	L No Yes 51	Yes 51 Yes 23	23 - 33		38,88		35	-5	145	>		۲. ۱	< 1 ×				<pre>(Low toxicity [5] Toxic by inhalation & ingestion [5]</pre>
3 3 3 4 7 7 7 7 7 7 7 7 7	Ethylene Oxide G No Yes >760 50 Ethylenediamine L No Yes 29 10	No Yes >760 No Yes 29	Yes >760 Yes 29	>760 29	29	8.2		75	3.5	1 %	m m			**	× ·			(3
(3) 1 (3) -	Ethylenediamine Tetracetic Acid S No No <0.25 DNA Ethyleneimine Stracetic Acid C No Yes 362 5/30 min	No No <0.25 No Yes 362	No <0.25 Yes 362	<0.25 362		DNA 5/30 mi	=			ı m		×ı		·×	. ×			
3 - 3 - 1	Ethylhexaldehyde L No Yes 5.7 DWA 2-Ehylhexyl Acrylate L No Yes 0.43 DMA Ethylhexyl Tallate L No Yes DMA DMA Ethyliderenorbornene L Nc Yes 11.3 0.2[5]	NO YES 5.7 NO YES 0.43 NO YES DWA NC YES 11.3 0	Yes 5.7 Yes 0.43 Yes DWA Yes 11.3 0	5.7 0.43 DNA 11.3 0	-	DNA DNA DNA 0.2[5]			<u>ee</u>	,	<u>ම</u> ෙ	1 1 1 1	1 1 1 1	×× :×	×× ×			Toxic rating as aldehydes Toxic rating as acrylic acid Not harmful Toxic [5]
(3) 1* X X - 1 Yes	Ethylphenyl Dichlorosilane L Yes Yes 0.57 DNA Ethyltrichlorosilane L Yes Yes 61 DNA Ferric Ammonium Citrate S No Yes <0.25 DNA Ferric Ammonium Oxalate S No No <0.25 l* (Fe)	Yes Yes 0.57 Yes Yes 61 No Yes <0.25 No No <0.25	Yes 0.57 Yes 61 Yes <0.25 No <0.25	0.57 61 60.25 60.25	 	DNA DNA DNA 1* (Fe)	† 		mm >		mm 3	× .		** : *	××××	 	 	Low toxicity Moderately toxic [5], burns skin
(3) - (3)	Ferric Chloride S Yes No <0.25 1* (Fe)	Yes No <0.25	No <0.25	<0.25	┼	1* (Fe)			١,	<u> </u>	=		,	×	×	 -	 	Water reacts to HCl, strong
1	Ferric Fluoride S No - 2.5 (F)	- O2		2.5 (F	2.5 (F	2.5 (F	_		3	ı	(3)	,	•	~	×			Toxic strong irritant; toxic
	Ferric Glycerophosphate S No No <0.25 1* (Fe) Ferric Mitrate <0.25 1* (Fe)	No No <0.25 1* No Yes <0.25 1*	No <0.25 1* Yes <0.25 1*	<0.25 1*	<u>*</u>			 	-	ì	~	××		, ,				Used in pharmaceuticals [5]
2 2 - 1 X X X - Yes Yes 3 X X - Yes Yes 1 X X X Yes Yes 1 X X X Yes Yes 3 1 X X Yes Yes Y Yes Yes	Sulfate S No No	No No <0.25 1* No No <0.25 1*	No <0.25 1* No <0.25 1*	<0.25 1* <0.25 1*)* (Fe) 1* (Fe)			,	,	-	××	, ,	, ,	1 1			Based on ferrous sulfate and
2 - 1	Ferrous Chloride S No No <0.25 1* (Fe) Ferrous Fluoborate L DMA 0.098	No No <0.25	No <0.25	<0.25 DNA		1* (Fe) 0.098				,	-	× 1	1 1	ı×	ı ×			Poisonous if inhaled, irritant to skin & tissue
2 - 1	Ferrous Okalate S No No <0.25 50 (Co)	No No <0.25	No <0.25	<0.25		(03) 05				·	е			×	*	,	-	Evolves CO on heating; toxic
- 3 -	No <0.25 No >750	No No <0.25 Yes No >750	No <0.25 No >750	<0.25 >750		DetA		.	266	, ,	- ~ .	× 1	× 1	, ×,	, ×,		es Yes	low toxicity
- 3 X X - No Yes Yes X X - Yes Yes X Yes Yes	L res no runes	res no runes	NO Futures	runes		¥	_ [, -		-	- †	- †	×	7	Ţ		ruming inquid, nigh toxicity
	Filososifonto Acid L Yes No 6.0 DNA Formal dehyde Solution L No Yes 4.6 2 Formic Acid L No Yes 77 5 Fumeric Acid Scid Scid Scid Scid Scid Scid Scid S	Yes No 6.0 No Yes 4.6 No Yes 77 No Yes <0.1	No 6.0 Yes 4.6 Yes 77 Yes <0.1	4.6		0.45 0.45 0.45 0.45			mma 1	1111	m m m -	×	× .	*** !	**!!		es Yes	Avoid skin contact [5]

			_					· 							≖	MAIFFIA	
						ILV (6	TLV (PPM OR	Ř	SAX-TOXIC		Ē	LEVELS OF PROTECTION	PROTE	S) []		COMPATI-	
J	3	S	¥	u.	چ	MG/A	MG/M3)*	16298	HAZARO RATING	و	8	S TOR	9	COLLING	— <u> </u>	BILITY [10]	
						¥	SIEL	NI N	झ	IR F	出	α	St	S3 ₹	-		NOTES
FFA	Furfural Furfury! Alcoho!		22	Yes	5.1 0.16	2 22	50	- 2	2	2 -		××	× 1		X Yes	s Yes s Yes	Dangerous to eyes [8] Toxic by inhalation [5], but
G.A	Gallic Acid	S	ş	Yes	<0.11	DMA		ı	,	,	<u>~</u>		· 		x Yes	s (PVC)	Mild irritant
_ 	Gas Oi): Cracked	7	No	Yes	Propane	500/30min		(2)		Ξ	-	×	×		_		Toxic rating as kerosene
S. F.	Gas Blending Stocks: Alkalates	7	22		อัจ	500/30min 500/30min		€8 ~	1 1	<u>=</u>		××				-	rating as ruting as
33	Gasoline, Automotive (4.23 6)8/Gal)		2 2		흔	500/30min 500/30min		22	<u>ee</u>	<u> </u>		××	××		22 		Toxic rating as propane Toxic rating as butane
	SESULTIFE, AVIACION 4.00 G TO/GAL)			-								_					
S	Gasoline: Casinghead	٦.	2	1	Butane	500/30min		=	Ξ	3	. ,	××	×	-	o¥ ×	L	Irritates skin & eyes Toxic rating as Dentane
3 S	Gasoline: Polymer Gasoline: Straight Run		€ €	S S	Pentane	Pentane500/30min		(2)	ΞΞ			·×			X No[11]	1] yes[12]	
GTA	ulutaraldehyde, Solution	J	N.	2	0.3	0.25*		(3)	ε	(3)		,	×	<u>.</u> ×	: :	<u> </u>	Toxic rating as aldehydes
33	Glycerine Glycidyl Methacralate		22	Yes Yes	6.1 <0.25	SS DNA			1	,		× 1	ı ×	, ×	X Yes - Yes	s Yes	Low toxicity [5] Burns skin & eyes, glycidyl
GOS HTC	Glyoxal, 40% Solution Heptachlor	78	Yes	Yes No	~300 <0.25	DWA 0.5*		2	1	8		× 1	××	1 ×	- X -	s Yes	
T#	Heptane	-	£	Yes	88	400	929			-	-	-	×	-	X MC/ACE	CE [Yes[12]	Marcol
HTM HTE HCC	Heptanol I-Heptene Hexachlorocyclopentadiene		222	Yes Yes	1.2 103 0.26	DNA 0.1	0.3	3-6	3 1-	⊃ 1 ©		×× :	11×	111	X X X	Yes -	Low toxicity[5],not harmful Asphyxiant,effects as argon[8] Toxic rating as CD4 toxic[5]
HSS	Hexadecyl Sulfate, Sodium Salt	-			0.26	Cala				-	-	×	 	-	×	-	Similar to surfactants, cosmetics
芸芸書	Hexadecytrimethylammonlum Chloride n-Hexaldehyde Hexamethylenedlamine	, L	. 22	Yes	<0.25 √10 17	DNA DNA		(3)	- 1	(3)	111	1 1 1	×××	×××	- Yes[1]		Severe eye 6 skin danage on cont Toxic rating as aldehydes Poisonous if swallowed, strong skin irritant or skin exposed(5)
HMI	Hexmethyleneimine Hexamethylenetetramine Hexane	-0-	222	Yes	1.0 <0.25 258	AN 2005	125	21-	3 11	30-		×	×× :	×× 1	X Yes	S Yes	
HXM	Nexano]		3	Yes	39	¥		,	1		1	,	×		X Yes		Toxicity malcohols, Low
芸芸芸芸	l-Hexene Hexylene Glycol Hydrazine Hydrochloric Acid		5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5	Yes Yes Mo	310 0.3 29 362	500 5.1		33-12	1 1 67 1	3355		×× . 1	×ı××	1188	X No Yes[1] - Yes[1]	rs Yes 11 Yes[12] 11 PVC[11]	Poisonous through skin
HFA	Hydrafluaric Acid		Yes	9 €		3		3		E	 -	-	_	×	- No(PVC)	VC PVC[T1]	
₹ 5₹	Hydrogen Bromide Hydrogen Chloride Hydrogen Cyanide	့	Yes Yes	& ₹ ₹	×760 ×760 ×760	~ s 2	51	m m m	1 1 11	m m %	1 1 1		×××	×××	es[1] - res[1] - res[1]	PVC[1]	
FF	Hydrogen Fluoride	כי	, es	2	>760	3		m		m	-	-	 ×	×	- 46 PVC	,Ο.,	
\$ 5 5	Mydrogen Peroxide Hydrogen Sulfide	ب وي د.	22	Yes Yes	>760	-2	2 15	ოო	1 1	m m	1 1	, ,	·		- res[]]	Yes Yes	
HXX	. Hydrogen, Liquified	17/6	Ş	Yes	>760	DAY	•	-	0	0		×	×		Yes		-

) A E) V (1994 95	CAY	CAY_TOY!C		4	NEI C	TUBE	L LOS	<u> </u>	MATERIAL COMPATI-	
CN S INC		¥		L.	9	ME/i	MS/M3)*	HZAR.	HAZARU RATING	2		IRATOR	RESPIRATORY CLOTHIII	CLOTHING	وا	BILITY (10)	
						TWA.	SIEL	×	Ħ	Œ.	胀	ខ	₹	8	€	ВР	NOTES
Hydroquinone S No	<u> </u>	윤		Yes	0.11	2*		9	,	7		,	×	1	× ×	No[11] PVC[11] PVC[11]	
2-Hydroxylamine Hydroxylamine S No Hydroxylamine S No Hydroxylamine Culfate		. 22	-	Yes	21.0 610	DNA		(3)	1 1	(3)	ı×ı	1×1	×·×	× ı×	- 	CENS (PVC)	Burns skin & eyes PVC[18] (PVC)[18] Toxic rating as sulfuric acid
1	+			Yes	°20	DNA							×	×			Acrylates-generally high toxicity Burns skin & eyes
Hydroxypropyl Methacrylate L No Isoamyl Alcohol Isobutane G No		222		Yes Yes Yes	61 8.3 760	100 A	125	۳-	10	2 0	111	×ı×	·× ·	1 1 1	** '	Yes Yes Yes Yes	Low toxicity[5],burns eyes
Isobuty] Acetate L Yes Isobuty] Alcohol L No Isobutylamine L No Isobutylene L -		S S S .		Yes Yes Yes	35 266 744 760	050 50 1000	187 75	2221		2261		×× : :	××××		×× · ·	Yes No Yes Yes Yes Yes Yes Yes	Asphyxiant, nontoxic [5]
Isobutyric Acid L No	+-	₽	•	Yes	3.6	DHA		ε	ε	2	<u> </u>		+	 	×	Yes Yes	Toxic rating inhal Skin as butric acid
		2,2		Yes Yes	6.9 70.4	DNA DNA		ලලල	333	333		111	×××	***	111	Yes Ro	Toxic rating as cyanide Toxic rating as aldehydes Toxic rating as aldehydes
Isodecj Acrylate (Inhibited) L No Isos yl Alcohol	┼~~	22		Yes	10.25 <0.25	¥ ¥		э	э	3	, ,	, ×	× ,	×	ı×	Yes Yes	Same as HAI Alcohols generally low toxicity
1		£		Yes	362	DIE		ε	•	ε	,	× ,	,		* *	9 ×	Toxic rating as hexane; low toxicity as hexane; low
Isopentane No	+	2 222		Tes Tes	1.03	(C) 5 DNA		≘≈-	211	38-	×	< × 1 1	1 × 1	 	+		Low toxicity[5]; see pentane
		2		Š	>760	VIII O		2	•	~	,	×	×			2	
Isopropy) Acetate Isopropy] Alcohol Isopropy] Ether Isopropy] Mercaptan		222 ·		Yes Yes Yes	103 93 242 7150	250 250 DNA	310 500 310	222E	111	66	,	×× I I	××××	IIXX	××ıı	Yes No Yes Yes Yes Yes	Toxic rating as mercaptan
Isopropyl Percarbonate S No	-	2	+	Yes	¥	DHA		5	-					 	-	res res	Legal name Diisopropylperoxy dicarbonate;unstable & stable force Flamable & corrosite fine
Isopropylatine Isovaleraidehyde Jet Fuel:JP-1 (Kerosene)		22 <u>2</u>		Yes Yes Yes	>760 111 5	5.0 DNA 200[9]	10	(3)	(<u>i</u>)	£33°		- ×	×××	×× I	- X	Yes No Yes Ho to(PVC)Yes[12]	
Jet Fuel: JP-3		2	$\overline{}$	Yes	589	[6]002		(2)	,	Ξ		·—	*	·	¥.	No(Bunah Yes PV())]	As kerosene toxicity
Jet Fuel: JP-6 (Kerosene, Heavy) L No Kerone		222		Yes .	1 4 5	200[9] 200[9] DW		222	' ' '	EEE			×××	111			As kerosene As kerosene As kerosene
Kerusene , L No	_	윤	-	Z S	5	[6]002		2	'	-		×	×	,	يو ×	lo(PVC Yes[12] [12]	
Lattic Acid Latex, Liquid Synthetic Lauroyl Peroxide S No		22		Yes Yes	~0.1 < 0.25 < 0.03	DNA DNA		116	1 1 1	226		~~ ·		11 8	**	Yes Yes	Could contain ammonia [5]
			1	1		7		-		+		-	.	-		٦	

) A F	OV (NOW YOU	SAV.	CAY_TOV1C		3	2000	1000	1 1		MTERIAL	
ر 	3	S	≆	ш.	<u>\$</u> 	79	-v (7rm cm	HAZADI	JAN-TUNIC	ـــنــ <u>ب</u>	100	AFCOIDATION CIMILIN			7	LUTTALIT	
_)	: 		:	AN AN	STEL	×	55	Œ	1	<u> </u>	S		#		MUTEC
¥3	Lauryl Mercaptan Lead Acetate	28	22	¥ es	~0.03 <0.25	DNA 0.15*(Pb)		33	l i	(3)			××	+	××	Yes	As mercaptans Suspected carcinogen as lead[8]:
ני א	Lead Arsenate Lead Chloride	y v	2 2	2 2	<0.25	0.15*(Pb)		(3)	, c	(3)	1 1	, ,		* *	Yes	Yes	absorbed via skin,inhalation[5] As arsenic_compounds;suspected carcinogen[8]
LFB	 -	٦.	. 2	1 2	49.25	DNA 0.15*(Pb)		<u>ee</u>	,	+		١.,	+	××	, Kes	, s	As fluoride, as fluoborate As fluorides,toxic rating,strong
CAT	Lead Iodide Lead Nitrate	SS	22	Yes	<0.25 <0.25	[9] 0.15*(Pb)		(3)	<u>6</u> 6		+ 1		**	**	Yes		irritant [5] As lead cmods; toxic rating As nitrate(2-inhalation)as lead
35	Lead Stearate Lead Sulfate	SS	22	¥ es		DKA 0.15*(Pb)		ලල	29	<u>ee</u>	 	1	××	× ·	es.	1 (PYC)	(3-inhalation) As lead cmpd;absorbed by skin As lead compds; toxic rating
55	Lead Tetrancetate Lead Thiocynate	ss	, 2	ž š	< 0.25 < 0.25	0.15*(Pb) 0.15*(Pb)		<u>ee</u>	<u> </u>	<u> </u>		1.1	××	·×	Yes	E \$3	As lead cmpd;absorbed by skin[5] As lead cmpd; toxic rating,de-
공급	Lead Thiosulfate Lead Tungstate Linear Alcohols (12-15 Carbons) Liquefied Natural Gas	s s s	2555	Tes Yes	< 0.03 > 760	0.15*(Pb) 0.15*(Pb) DNA DNA		EE=-	<u>6</u> 6>0	99-0		1 1 × 1	×× ×	***	Yes	Yes	comp. not water " " " " " " " " " " " " " " " " " " "
25 ₹8	Liquefied Petroleum Gas Litharge Lithium Aluminum Hydride Lithium Bichromate	NSSE	Mo Yes	Yes No Yes	> 760 < 0.25	1000 0.2* DNA DNA		-888	00.	-888		× 1 × 1	1×1×	1 × × 1	Yes	Yes	Asphyxiant Toxic rating as lead compound Toxic rating as lithium compound Carcingen [8], toxic rating as chromium compound
<u> </u>	Lithium Chromate	S	£			DAG		(3)	1.	(E)	+	1.		ŀ	\downarrow		Carcinogen. Toxic rating as
물론	Lithium Hydride Lithium Metal Magnesium	NNN	Yes Yes Yes	Yes Yes	< 0.25 < 0.25	0.0705 DNA DNA		222	111	388	***	***	111	***	Yes	Yes	chromium compound Toxic rating as lithium compound Low toxicity Toxic rating as magnesium cmpd.
3 F	Magnesium Perchlorate Malathicn	E S	2 2	Yes	< 0.25 < 0.25	DNA 0.90077		2	,	2 × -	-	× 1	1.2	× 1	Yes Yes	Yes	Cholinesterace inhibitor, in-
11	Naleic Acid Maleic Anhydride	SS	No Yes	Yes	< 0.25 0.6	DNM. 0.25/1*		3.2	1.1	н м			**	11	Yes	Yes Yes	hal. & skin absorp. [5]
15	Maleic Hydrazide Mercuric Acetate	SS	2 %	Yes	2 0.1	DNA 0.05*(Hg)		33	63	<u>:</u>	-	-	**	1 1	řě	Yes	Toxic rating as hydrazine Toxic rating, organic Hg cmpd.
	Mercuric Amonium Chloride Mercuric Chloride	SS	22	€ £	CMA 0.00047	0.05*(Hg) 0.05*(Hg)		<u>ම</u> ෙ	1 1	<u></u>		+ 1	×× ××	1.1	Yes	Yes	Skin absorp. [5]
ZEFS ZEFS	Mercuric Cyanide Hercuric Iodide Hercuric Nitrate Mercuric Oxide	2222	2222	£ 2 5 2	ONA < 0.25 ONA < 0.25	0.05*(Hg) 0.05*(Hg) 0.05*(Hg) 0.05*(Hg)		666 6	(E)	<u></u>	-	***	××××	1111	Yes Yes	Yes Yes Yes	Toxic rating as cyanide Toxic rating-inorg, Hg. cmpd.
MRS	Mercuric Sulfate	5	Yes			0.05*(Hg)	0.15*	(3)	-	(3)	-	×	>-:	1.	(PVC		Toxic rating -Inorg. Hg. cmpd.
A A B	Mercuric Sulfide Mercuric Thiocyanate Mercurus Chlorida	SS	22:	Kes		0.05*(Hg) 0.05*(Hg)	0.15*	33	(S -	<u>: -</u>		××	××			[12] Yes	Toxic rating-org. Hg compounds
		^	2	2	0.25	0.05*(Hg)		(E)	-	(3)		*	_		Yes	Yes	Toxic rating-inorg. Hg cmcds.

		L						-							\vdash	MIERIA	
<u> </u>	5	v	9	u	9	<u> </u>	1LV (PPH OR	¥ 24.	SAX-ICKIC	¥		TESTINATION OF THE PROPERTY OF			<u>.</u>	RILITY COLD	
,		·	í	•	:	F	STE	*	স	≅	#:	ន	8	8	2 9	8 -	SILON
ŧ	Mercurous Mitrate	s	8	ë	DIG.	0.05*(Hg)		<u>e</u>	<u> ·</u>	<u>e</u>		1.	 	<u> </u>	<u>ş</u>	res Tes	Toxic rating-inorg. Hg capds.
효율력	Mercury Mesityl Oxide Methallyl Chloride		888	2 % %	0.002 220 220	0.05*(Hg) 25 DNA		(3)		6,4		118	***	**!	×	25	Toxicity; frr. inhal, skin [5]
ĒØ	Methane Methane Arsonic Acid, Sodium Salt	0.0	₽2	\$ 6 \$	7 160 DBG	DRA		-Ξ	01	οE	1 ×	ļ	× ,	1	x Yes	ž ž	Toxic rating as methane arsonic
53	Methoxychlor Methyl Acetylene, Propadiene Mixture	~ <u>°</u>	22	Š.Š.	<0.25 >760	1000	-		•		×ı	11	114		X X X		acid Less toxic than DOT [5] Animal test indicate minimal toxicity (Ahal, toxic (5)
Ē	Methy! Acetate		ş	řě	36	200	250	2	2	E	1.	×	-	[12119 (117) NOT 121	Asphyxlant
##	Methyl Alcohol Methyl Anyl Acetate Methyl Anyl Alcohol		222	ž š ž	237 1.0 14.0	200 284A 25	250 #0	~~~	0 I M			××I	***	118	X X Y	2 2 2	Poisonous through skin [5] Poisonous through skin [5]
	Methyl Browlde	9/3	2	Yes	> 760	15		~	Ŀ	m	1.	1.	×	×	2,	25	
FĐĐ	Methyl Chloroformate Methyl Chloroformate Methyldtchlorosilane	دوق	\$ 2 2 2 2	Yes Yes	220 220 750	100 DMA 5(HC1)	521	mmm		~~~		, , ,	***	188	Ze Yes	- S 8	·
東京産業	Methy Ethyl Katone Methyl Format Methyl Formate Methyl Mydrazine	دددد	8888	2222	266 266 267 268 268	200 1000[9] 100 0.2 [9]	300	(33.22	· · · (£)	- 126		× 1 1 1	***		N X X	2 2	Skin.eyes.mose.throat irritant As hydrazine
高高素	Nethyl Isabutyl Carbinol Methyl Isabutyl Ketone Methyl Isapropenyl Ketone (Inhibited)		22.	555	2 8 6 2	25 200 100[9]	125	m(ĝ)⊃	m 13	F(8)3	1	 	***	H 1 H	2.2	2 2	Irritates eyes,skin;prolomged
¥	Methyl Mercaptan	%	ş	ş	2 760	0.5		(2)		(3)	٠,	*	*	-	X Yes		CONTACT DISTERS SKIN
Į,	Methyl Methacrylate Hethyl Parathion	70	2 S	ž ž	¢ 0.01	0.2*	0.6	-ŝ	' ≅	=		× '	ı ×	1, 71	2 2	2 <u>2</u> 2	As parathion cholinesterace in-
¥	Methyl Vinyl Ketone Methyl-a-Buryl Ketone	9 5 -	2 2	ž ž	153	100691		m	•	~	, ,	, ,	× ×	× 1	2 2		Skin irritant & lachymator.
3535	Methyl Acrylate Methylamine n-Methylamiline Methylcyclopentadieneyl Manganese Iricarbonyl	د د ی د	5555	ទីទីទី '	10000	2 [9] 0.1 [9]	0.3	m@r	m 1 m	m(f)		1.,	***	ннян	222	252	As monomethylamine Poisonous 1f swallowed or skin exposed
	Nethylethylpyridine Nethylethylpyridine Nethylphosphonulnioic Dichloride, Anhydrous I-Nethylpyrrolidone		\$ 5'5	ž ž ž	2.28 0.9 0.9	BA A A B		⇒ £ =	(2)	. (3 c	1	× 1 1	1 8 8	IMM	2 3 °	10 165 100 163 100 163 100 163	Eyes & stin irritant & narcotic in high concentrations As pyridine Highly Loxic irritant.corrosive Wery corrosive(lite amonia).
8 8 8	Methylstyrene, Alpha Methyltrichlorosilane Hineral Spirits	7	3 5 3	Yes Yes	294	5 (HC1)		26,2	131	~ <u>@</u> -		× · ×	***	1 1 7	X Yes	Yes Yes Yes Yes Ho Ho (PVC)	1
M.X.	M(rex	S	2		-1	780						- :	. *	· .		(c) (e)	n'nhly toxic stomech insect.[8]

																MATERIA		
		,		,	9	TLV (PPM OR	TH OR	\$	SAX-TOXIC			LEVELS OF PROTECTION	F 28	ECTION	Ş	COMPATI		
ں	S	S	¥	<u>.</u>	€	MG/MJ)*	(5)*	3	F.C. S. S.		2	5	-		اج	BILLIY [10		C.L.C.
						≨	ZIE.	=	55	¥	£	3	2	2	2	n	~	MIES
₽ 2	Molybdic Trioxide	v	- 2	2		5*(M) [9]		,	,	_	×	'	۰			Yes	خ وغ	
24	Monochloroacetic Acid	νc	22	Yes	0.9	DMA 1000	-	2 2	1 3	~ ¬	ж і	××	· ×	• •	××			Refriserant; liquid can cause
	Police in the contract of the							,		^	٠		×		<u></u>	Z	ŝ	frostbite [8]
5	MOTOD CHICAGO		2 2	2 3		, Lejs		, =	=	• =	+	;	+	+	\dagger	+		Tay ic by toha & inges irri-
Ě	Monotsopropanolamine		2	6				,		-		;	 ,	····				tates stomach tissue [5]
## ##	Morpholine Motor Fuel Antiknock Cmpds.(Pb Alkyls)		22	e e	0.67	0.1(Pb)	⊋,	γ <u>(</u> E)	7	7	. ,	~ ,		· ×	- /-	e se	No (PVC)	As lead cmpds; Pb alkyls ab-
N.A.R.	Zu trans	,	2	Ş	<0.25	্বি		'	,	,	,	٠	×	×	-	Yes		sorbed by skin [5] Recog. carcinoden:narcotic at
3	Na Care	,	2	2						1		+	1	-	+			high concestin & mycous irritant
MC D	Naled Naphtha: Coal Tar	ر د د	22	Yes	6.7	100 100		⊃ m	3 M	- ~		, ,	× ×	× ×	, <u>z</u>			Cholinesterace inhibitors [5]
				,		ž		,	•				<u> </u>	·····	<u> </u>	Hitrile.	Meopres	ne)
NSS N	Naphtha: Solvent Naphtha: Stoddard Solvent		2	ž š	6.7	300		2 0	1	7 -	1 1	· ×	<×	« 1	- - -		<u> </u>	-
¥	Naphtha: YM P (75 Naphtha)	_	2	, es	6.7	500/30 min		2	 -	2	-	×	×	ļ	-	2	 -	
N.	Naphthalene, Molten	S	Yes		و د،	10/80*		~?	7	7	× ·	*	1		4	16 [[[]		Standard conditions
S	Naphthylamine • Naphthylamine	n vn	2 2	Yes	-0.01	DNA		3,6	ı m	2	< 1	< 1		×	· ·		Yes	Recognized carcinogen
XHX	Neohexane	-	1	Yes.[5]	~700	DNA		3	=	∍	1.	-	×	 			1	Low toxicity; narcotix in high
MKA	Nickel Acetate	_	9	2	25	14(Ni)/6]		(3)	'	Ξ				1			res	As 31 cmpd; recog.carcinogen
8 88	Nickel Ammonium Sulfate Nickel Brownide	S	22	22	<0.25 <0.25			<u> </u>	, ,	ΞΞ	, ,	1 1	××	1 1	××	Yes	Yes	As Mi compounds
MKC	Nickel Carbonyl	_	운	Yes	634			3		+	1.	+		+	+	+	Yes	
ZZ.	Nickel Chloride	s	£	2	< 0.25			<u>e</u>	,	Ê	,	•	×		×	Yes	Yes	As N. cmpd,toxic rating, carcinogen
ŽË	Nickel Cyanide Nickel Flucborate	ν	운,	2 ,	SK A	DHA DHA		ලල	€ .	<u> </u>	, ,	1 1	××	××	1 1	Yes	, kes	ss cyanisus;recog.carcinogen ioxic rating as flucrides
N.	Nickle Formate	N	2	1.	DWA	0.12[9]		(3)		ε	1.	 -	×	*	1.	, es	Yes	Toute rating as mickel cmpd,
WKH.	Nickel Hydroxide	s	2	,]*(Ni)[5]		3	,	Ê		,	×:	×				j Na :
E S	Nickel Nitrate Nickel Sulfate	v v	<u>2</u> %	Kes Kes	× 0.25 × 0.25	[6](IN)*!		 ©©	1 1	ĒĒ	' '	1 1	>: ><			Yes	řeš řeš	
NCS	Nicotine Nicotine Sulfate	100	22:	ž č	~ 0.25 < 0.25	0.069[9]		(3)	F(C)	(2)	.,		×××	×××		Yes	Yes	As nicotine, toxic rating
J ¥C	Nitrain Nitric Acid	<u>م</u>	ze ze	, ,	62.0		4	6	,	е.	'',	, ,	< ×	4 24		No Yes (?	(PVC)	skin,eye irritant
XTX	Nitric Oxide	9	Yes	2,	760	25	35	-	-	67	-	-	× >	× ×	1.		+-	Confirmed carcinomen in rate
٤	Nitriotriacetic Actually Saits	^	2	<u>g</u>	0.0	<u> </u>						1	4,	c :				and mice [5]
ž ž	Z-Nitroaniline 4-Nitroaniline	v v	22	s s	. 0.25 . 0.03	1/6*	2	<u> </u>	m (f)	<u> </u>	1 1	1 1	~ ×	××) (ži.	s S	As p-N:troaniline,toxic rating
MTB.	Nitrobenzene	7	£	ies	0.7		2	6	3	Э	-	 	~	-		+	No (ACE)	
M X	Nitroethane Nitrogen Tetroxide	1/6	¥es Yes	Yes	38 > 760	6 6 8	150	(3,2		(3)	1 1	»: ı	××	1 ×	× i	74.5 (c. Yes	es	As nitric oxide toxic rating
XX.	N: zn. Liquefied		, se	2	° 760	9×10 [5]						-	×		,		×es	Asphixiant, component of air
												: 						

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COPATI- BILITY (10)	\vdash	Ves Ves Nontoxic[5];not harmful [5] No(PVC) = [12] Irritates eyes \$ skin [9]	•	No loxic rating as kerosene	Tritates eyes 4 skin [9] Yes Yes Not harmful No Tritates eyes 4 skin [9] No Toxic rating as kerosene	Yes Yes Exposure data not available No Burns skin & eyes [9] Yes Yes Exposure data not available Yes Carcinogen sv.pect[8], exposure data not available [9]	No Irritates skin & eyes [9] No Irritates skin & eyes [9] Low toxicity [5] Exposure data not available [5,8,9]	No Irritates skin & eyes, chlori-	No Irritates eyes, nose, throat, skin[4] Yes Yes Data not available [5,8)	No.	(mitrie) (mitrie) (see Yes Nontoxic [5] (ves Yes Toxic rating as oleic acid (yes Yes Toxic rating as oleic acid[8], (low toxicity [5])	HO(PVC) (PVC)	Yes	Decomposes on standing to	Yes Yes diveryues No Yes Yes	Yes Yes Linear alcohol; low toxicity[5] Yes Yes Muisance dust[5]); not harmful[9] No Marcotic in high concentration No Asphyxiant	Yes Yes Restable limit explodes with
CLOTHING	Si	××	×	×	× 1 × ×	 	×××	×	× ×	<u>×</u>	***	1	1 1 ×	×	1 1 1	× 1 1×	
	8	<u></u>	<u>.</u>	4	p 1 2 1			ŀ	<u> </u>	ŀ	111	×	× 1 1	<u> </u>	×××	1111	××
LEVELS OF PROTECTION RESPIRATORY CLOTHIN	C AS		<u>'</u>	×		-	1 1 1	<u>'</u>	- '	×		×	× 1 1	×	**×	× · ××	× ×
	α	**	×	×	× 1 × ×	 	×××	×	× ×	×		-	1 1 ×	×	111		- -
	<u> </u>	' '	1				1 1 1	<u> </u>		•	1××	<u> </u>	2 3 1 1 X	<u>'</u>	335	ואו	
ر ا	T.	- =	<u> </u>	Ξ	E	3133	- c	-			-88	-		-		3 . 3 -	-
SAX-TOXIC HAZARO RATING	<i>S</i> 5	12		-	13 1	3123	3 '	-	<u> </u>	<u>'</u>		3		<u> </u>	.,,,,,		+
~ ₹		, =	-	[2]	(2)	3 : 33	3	=		2		-	e 2	2	<u> </u>	3 1 6	
12.V (PPM OR MG/M3)*	SIE												2			750	
TLV O	¥	DNA DNA	DNA	200	DNA DNA 200	DNA DNA DNA	DNA DNA DNA	ONA	0.43! DNA DNA	CMA	DNA DNA A	*	1* DNA 5	DNA	0.00077	10* DNA 600 DNA	A.A.
9	:	DWA 5.3	5.)	5.1				5.1		5.1	60.25 60.25 60.25	<u>-</u>	~0.005 >760 22	23	0.001 ~150 <0.25	<0.25 <0.25 >760 >760	36.6
L-	•	Yes	Yes	Yes	Yes Yes Yes Yes	Yes Yes Yes	Yes Yes Yes	Yes	Yes Yes Yes	Yes	Yes No No	是	Yes No Yes	Yes	Xo Yes	Yes Yes Yes	Yes
£	É	2.8	윥	No	2222	2222	2222	S.	222	2	222	, kes	No No Yes	ક્	No Yes No	2888	No.
v.	,			1		ب بد د. ب	~ ~ ~ ~]_		_	N L N	-	N L L	ררטר	
2	5	Oil, Misc: Linseed Oil, Misc: Lubricating	Oil, Misc: Mineral	Oil, Misc: Mineral Seal	0il, Misc: Motor 0il, Misc: Neatsfoot 0il, Misc: Penetrating 0il, Misc: Range	Oil, Misc: Resin Oil, Misc: Road Oil, Misc: Rosin Oil, Misc: Sperm	011, Misc: Spindle 311, Misc: Spray 611, Misc: Tall 011, Misc: Tanners	0il, Misc: Transformer	Oil, Misc: Turbine Oil, Misc: Clarified Oil: Crude	0il: Diesel	Oleic Acid Oleic Acid, Potassium Salt Oleic Acid, Sodium Salt	01eum	Oxalic Acid Oxygen, Liquid P. raformaldehyde	Paraldehyde	Parathion Pentaborane Pentachlorophenol	Pentadecanol Pentaerythritol Pentane I-Pentene	Peracetic Acid
٠	 د	S as a	₹	500		OSRO OSRW OSP	050 057 011 018	150	018 0CF 01L	90S	OLA OAC OAC	5	OXA OXY PFA	F0.	PT0 PCP	PET	PAA

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		,	!			TLV (PPH OR	æ. Æ.	Š	SEX-TOXIC		LEVELS OF PR	SIGN TO TO TO TO TO TO TO TO TO TO TO TO TO		TECTION INC.	8		
ပ	3	s	¥		}	MG/H2)*	\$	3	HEZEND PALING	-			3 ⁸		ā		
						*	STEL	*	Ħ	IR	PF 0	CC AS	8	2 2	-		MOJES
ME.	Petroleum Maphtha	-1	9€	Yes	12	500/30 mie		2	_	2	× -	× 		×	() () () ()	Yes[12	
	Dheool			,	•	******	2	~	,			_		×	Že.	Yes	
₹ 2	Phenyldichloroarsine, Liquid	د ه	Š	<u>0</u> <u>0</u>	0.23	- M	:	m		<u>.</u>		× ×	× >	_	es,	윤	
₹	Phenylhydrazine Hydrochloride	s	2	₽	< 0.25	¥		n	າ					•	<u></u>	€	
£	Phosdrin	-	윤	\vdash		0.01	0.03	(3)	(3)	L	-		× ;	-	,	3	As parathion[8]
¥.	Phosgene	<u>ور</u>	Yes		2,760			mr	,	m e	. >		_	. >	ě ě	1 es	
PAC	Phosphoric Acid Phosphorous, Black	٧,	2 2	Že Š	C 7.0.	16757 DNA		,	 I						! 	<u> </u>	Does not spontaneously ignite,
•		,	!					_									stable in air
8	Phosohorous Oxychloride	-	Yes	╈	< 0.25	0.5		3	1	\bot	+	+	+	+	Yes	sa	As hydrochloric acid [8]
2	Phosphorous Pentasulfide	, v	e .		< 0.25	10(H ₂ S)		m	,			_			Yes	Yes	As history and [9]
88	Phosphorous Tribromide	<u></u>	Yes	2	6.3	¥ c		99		<u> </u>		· ·	< ×		, es	Yes	As hydrochloric acid [8]
£			S		17			3									•
900	Phosphorous Red	,	, vox	χάχ	< 0.25	OW.		(2)		(2)	<u> </u>	<u> </u>	FIRE	re Suit	ق	res	Reacts with water to form
X.		,	3							. '						_	phosphire[5] which ignites
ž	Phosphorous, White	s	₽	Yes	< 0.25	0.1* [9]		m	ı	m	· 	× 		Fire Sulf			fignites air @ 86°F
7	Phthalic Anhydride	5/1	Yes	Les .	۰ 0.01		-	-	'			×		× ×	šě	Yes Yes	
244	Piperazene	S	윤	+	2.9	¥.			. :	-	\dashv	+	+	+		-+	Hanne are act as accountant in
P.B	Polybutene		£	Yes	< 0.25	*		7	 >		<u>. </u>	≺ —		1	<u>.</u>	<u> </u>	as olefins [a]
52	Polychlorinated Biphenyls	3/1	£		0.01	0.5	0.1	(3)	Ξ	(2)	_	. >	× ·	1 34	Yes	Yes	As chlorinated diphenyls[8] Polymer of methylene (4 obenyl-1
<u>E</u>	Folymethylene Polyphenyi Isocyanate		,	es	000.0	. 35/"[9]								•			isocyanate irritant [8]
PPA	Polyphosphoric Acid	١	Yes	윤	< 0.25	DHA		'	•	2	-	×	<u> </u>	×	ĕ	Yes	Phosphorous comp. low toxicity
39d 11d	Polypropylene Polypropylene Glycol	N J	₽ 2	Yes	< 0.25 < 0.25	***		-		_				1 1	ž š	Yes	Nontoxic [5]
													_] 	_	toxic [8]
ā E	Potyssium Glycol Methyl Ether Potyssium	٧.	Yes	Yes Yes	< 0.25 < 0.25	X X		м	,	т		+ × +	, <u>T</u>	re Suit	: Ma(PV(NO(PVC) (PVC)	Low toxicity;high conc.anesthetic Caustic potash,mat'l. compat.
PAS	Potassium Arsenate	S	욡	2	< 0.25	0.5*[9]		9	'	2	<u> </u>	*		~	Yes	Yes	Recognized carcinogen, As
№	Potassium Arsenite	S				***		~						<u>,</u>			arsenic comp. [8]
82	Potassium Binoxalate Potassium Chlorate	S	2 2	2 %	< 0.25 < 0.25	DNA BNA			,		. נסנוא	: ×	×		Yes	Yes	As oxalates [8]
5		,	!	\neg	2	5		-	7	-	_	-	-	-	Se l	Tes	
질문	Potassium Chromate Potassium Dyanide	u u	e so	22	< 0.25 < 0.25]* [9] 5*(CN)[5]		ලිව	. 8	ලිදි		××			Yes	Yes	As chromium compounds [8]
5 0	Potassium Dichloro-s-Triazinetrione	S	₽:		S.	¥.		3	<u> </u>		<u>'</u> ×			~	ie i	, se	Irritates skin & eyes [9]
2	TOTASSION DICHEMACE	^	<u> </u>		6.0	<u> </u>		<u> </u>	,	<u>.</u>					 Les	Yes	Recognized carcinogen, As chromium compound [8]
E	Potassium Hydroxide	S	Yes	+	0 %	DWA		~	1.	+	+	+	+	+	} - -	4	
PII	Potassium Todide	S	2 :		× 0.25	8		2	1	, ,		· ×	< ·		Yes 5	Yes	Low toxicity [8]
2 0	Dotacetim Dermandanate	n v	2 2		× 0.25	Y S		1 6	,			× >		_	Yes	. Kes	
:		 າ	 2			<u> </u>		<u> </u>				_			ves	Tes	
POP	Potassium Peroxide	S	Yes	Yes	< 0.25	ONA		3	1.	3		×	<u> ×</u>	-	Yes	Yes	er wide, peroxi
PRP	Propane	و	<u>.</u>	-		000		-	-						, A	 	inorganic [8] [8] Ashish cont
P	Propiolacetone, Beta		2 2	•	6.2	0.0 [9]	*	· m				· ×	≻		e Ç		Experimental carcinogen [8]
	عد درست محربان عدر		2	- 1	5	¥	- 1	2	-	. 2	γ .	×		×	Yes	₽.	

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	į	v	9	<u></u> . ц	9	5) ATI	1LV (PPM OR WG/43)*	F 78.	SAX-TOXIC HAZARIJ RATING	٠.		RESPIRATORY CLOTHIN	30		1	WM-A11- BILITY[10]	
د	5	,	É	-	=	1	SIE	Z	55	<u>=</u>	<u>#</u>	υ	SE SE	SSI	-	a.	NOTES
PNA	Propionic Acid Propionic Anhydride		22	Yes	9.3	10 0NA			, , ,	~ (E)	26	. [].	, , ,	××;	Yes		Animal exp. indicate low toxicit; As propionic acid [8]
PAT	n-Propyl Acetate	۔	e S	Yes	~	200	250	2				 ,	. ,		<u>.</u>		
₹	n-Propyl Alcohol	٦	Ş.	, es	37	550	250	2	2	-	,	 	V	<u> </u>		$\overline{}$	
£	n-Propyl Mercaptan	1	ON.	Yes	276	DNA		5	3	3	•	,		' 	- Kes	χ e s	Probably toxic[8], offensive smelling liquid
PRA PP1 PBP	n-Propylamine Propylene Propylene Butylene Polymer	ىدوى ب	888	Yes Yes Yes	410 > 760 < 0.25	DNA 4000 DNA		(2)	105	m 0 3	1 1 1	· * *	×××	* ! ! * ! !	Yes	Yes	As olefins [6-8]
PPG PWE POX PTC	Propylene Glycol Propylene Glycol Methyl Ether Propylene Oxide Propylene Tetramer		2222	Yes Yes Yes	0.27 23 > 760 3.8	DNA 100 100 200 [9]	150	1 1 0 3	1115	03		××	× .		Yes Yes Yes	Yes Yes Yes Yes	Nontoxic [5] Irritant & narcotic in high conc. [9]
PRR PR	Propyleneimine (Inhibited) Pyrethrins Pyridine		222	Yes	250 DNA 4.1	2 DN4 5	01	-32	2 - 2	200		1 1×	××ı		res[14]	No (ACE	Experimental carcinogen [8] Household insecticide [5]
PGA	Pyrogallic Acid	v	ę.	Yes	< 0.25	DMA		m	٣	т	,			×	Yes		
N.	Quinoline	-	2	Yes[5]	0.3	DWA		3	a	¬			 ×	×	 		May produce retinitis [8], irritates eyes & skin [9]
RSC SLA SLD	Resorcinol Salicylic Acid Selenium Dioxide	SSS	222	Yes Yes Yes	0.0087 < 0.25 < 0.25	10 DNA 0.2*(Se) [9]	50	(2-3)	2	2 (2)	1×1	×ıı	×·×		Yes	Yes	As selenium compounds [8]
510	Selenium Trioxide	v-	2 × ×	22	< 0.25	0.2*(Se)		(2-3)	١ ،	3)			××	× 1	Yes	Yes	As selenium compounds [8]
SVA	Silver Acetate	ıv (2	2	< 0.25	0.01*(Ag) [9]				-	× ×	1 1	1 1	× ×	, Kes	Yes	Irritates skip & eyes, harmful if swallowed [9] irritates skin & eyes [9]
SVC	Silver Carbonate	2	Q.	ĝ.	< 0.25	•				-	<		-	-	-	_	3
SVF	Silver Fluoride Silver lodate Silver Mitrate	NN I	22	No Yes	< 0.25 < 0.25	:		<u> </u>	=	<u> </u>	- × -	1 1 ,	× 1 ×		Yes Yes	Yes Yes	As fluorides [8] Similar to bromates & chlorates As chlorates[8], irr.skin & eyes[9 As nitrates[8], irr.skin & eyes[9
S S S	Silver Oxide	20	22	2 2	0.25	0.01*(Ag)		(c) -	5 1	·	×	, ,	< 1	· ×	Ç S		As silver compounds [8]
SVS	Silver Sulfate	S	ş	ê.	< 0.25	=			-	,	×	,	,	× -	Yes	Yes	As silver compounds[8], irritates skin å eyes [9]
SAS SAB SAB	Silver Alkyl Sulfates Sodium Alkylbenzene Sulfonates Sodium Amide	s/L S/L S	No No Yes	- No Yes	< 0.25 < 0.25 < 0.25	DNA DNA DNA		(3)	,	<u> </u>	' '	× 1	· ×	× 1	Yes	Yes	Irr. to skin & eyes [9] As NaOH&NH3-reaction w/H20 [8]
SDA	Sodium Arsenate	S	2	No No	< 0.25	0.5*(As)		(2)	'	<u>e</u>		-	×	×	Yes	Yes	As arsenic compounds, recognized carcinogen [8]
SAR SAZ SBF	Sodium Arsenite Sodium Azide Sodium Bifluoride	NNN	222	Yes No S	< 0.25 DNA -	0.1(0.3*) 2.5*(F)[5]		322	,	ලලල	111	111	×××	×××	Yes	Yes	As arsenic cmpd,herbicide[8] As sodium hydroxide [8] As fluorides [3]
SBS	Sodium Bisulfite	S	2	2	< 0.25	DINA		(2)		1.	×	×	-	-	(11)	PVC)	As sulfites [8]
SD8 SBH	Sodium Borate Sodium Borohydride	NN	No	Yes	< 0.25 < 0.25	2*(NaOH)		(S)(S)	52	>>	××	××	1 1	×× 	Yes	Yes	As boron compounds [8] As boron compound with moisture [6] Wach was be prosent
200	Sodium Cacodylate	8	No.	20	< 0.25	DNA DNA					×			×	, kes	, les	As cacodylic acid [8]

PATERIAL PROTECTION CONTRACTOR	RESPIRATORY (CLOTHING BILL., [10]	PF CC AS SS NS B P NOTES	X	X X Yes Yes Irritates skin & eyes [9]	- X X - Yes Yes As chromium compounds, carcino-	X[6] X Yes Yes As Frencyanides[8]avoid acid	- X X - Yes Yes As sodium hydroxide [8]	- X X . Yes Yes As sulfides(skin irritant)irri-	Yes No(PVC)	As hypochlor Toxic & corr.	X X X Yes[11]Yes[12]Toxic rating as nitrates PVC[11]	- X Yes	X . X - Yes Yes Toxic rating as oxalates X X Yes[11] Wild irritant; skin & eyes [9]	- X X - Toxic rating as selenium.recog-	Yes Yes Nontaxic [5] Yes Yes Yes Laxic rating as fluosilicates Yes Yes Yes Yes Kinni Lissue [5] Innstable H2S produced	X X Yes Yes Toxic rating as sulfites X X Yes Yes Toxic rating by inges-	X - X Fire Suit No Reacts with moisture, burns [9]	X - X X - X Yes Yes Slight toxicity[8], nontaxic[5] X X X - As Chromium compounds [3] - X X - Ijoxic by inhalation & ingestion [5], extremely toxic	X X X X X NO X NO X NO X NO X X X X X X	Yes Yes	7711 00	Yes	X	X X - Yes Yes - X X - Yes Yes
-	ING	IR	-8E		(3)	<u>Ω</u>	<u>33</u>	<u>'</u> -	<u>.</u>	(2)	-		<u>: ^</u>	(2)	(3)	-	m 	(E) (E)	2	(3)		m m	 	 -
SAX-TOXIC	HAZARO RATING	ऊ	(3)		<u> </u> ,	1			1	15			1	,	1 1	,				<u> </u>		, ,		
Š	HAZA	÷ .	3932		(3)	-	(3)		2	(S)	(2)	(3)	1	(5-3)	(3)	(2)	m .	(3)	2	(3)		m m	m m m	m m m
oo maa) All	MS/M3)*	TWA STEL	DNA 0.1* 5*(CN)[5]	DNA	DNA	DNA	DNA	DNA	2*/15]min	200(CH30H)	DNA	DNA	DNA	0.2*(Se)[9]	DIVA 2.5* DIVA	DNA	DNA	.1 DNA 0.5*(Cr) 0.15* 0.45*	100 125 DNA	10(H2S) 15(H2S)		5	s- *-	1 + DMA
	\$		<0.25 <0.25 <0.25	<0.25	<0.25	<0.25	<0.25 DNA	<0.25	<0.25	<0.25 <0.25	<0.25	<0.25	<0.25 <0.25		<0.25 ONA <0.25	40.25 40.25	<0.25 <0.25	2*(sn)[5] <0.25 DMA	13 <0.25	<0.25 <0.10		>760	>760 ~22 <0.25	>760 \u22 <0.25 <0.25
	Ц.	;	222	£	Ş.	Q.	No Yes	ş	§.	No Yes	Yes	Yes	₽₽	윤	- So - So - So	S &	Yes No	Yes -	Yes	Yes		No	No Yes No	No No No
	£		No Yes	S.	S.	S.	No Yes	£	Yes	Yes	No No	Ş.	<u> </u>	ν	£ 22	22	Yes No	S S S S	22	8 S		Yes	Yes Yes Yes	Yes Yes Yes
	S		SSS	s	S	S	SS	-	N	S	S	S	s s	s	ᄱᄱ	SS	νν	SSSS	7.0	_		1/9	1/9 1/9	1/9
	S		Sodium Chlorate Sodium Chromate Sodium Cyanide	Sodium Dichloro-s-Triazinetrione	Sodium Dichromate	Sodium Ferrocyanide	Sodium Fluoride Sodium Hydride	Sodium Hydrosulfide Solution	Sodium Hydroxide	Sodium Hypochlorite Sodium Methylate	odium Nitrate	Sodium Nitrite	Sodium Oxalate Sodium Phosphate	Sodium Selenite	Sodium Silicate Sodium Silicofluoride Sodium Sulfide	Sodium Sulfite Sodium Thiocyanate	Sodium, Metallic Sorbitol	Stannous Fluoride Stearic Acid Strontium Chromate Strychnine	tyrene	Sulfolane Sulfur(Liquid) @ 113 to 120°C		iulfur Dioxide iulfur Monochloride	iulfur Dioxide Sulfur Monochloride Sulfuric Acid	Sulfur Dioxida Sulfur Monochloride Sulfuric Acid Sulfuric Acid, Spent
			888	Š	\N	Ŋ	ಜನ	1 3	Š	N N	10	S	SS	١ ٠	•	0,0,	0.0,	10,0,0,0,	10,0			10,0,	10,0,0,	10,0, 0, 0,

		 	<u>-</u>	! !	! 	> L	ON MODEL OF	X	CAY_TOYIC		. <u>.</u>		1000	, a		MATERIAL	
<u>၂</u>	5	S	≆	<u>u</u>	9	1	Mc/u3)*	HAZAD	HAZARN DATING	 پ	18	ECOLOTION CONTRI	5		·		
	•	· 	[—		: 	₹	SE	2 3	5	<u>≅</u>	2 4	5 -	2 8	S S	+	61.11.[10]	T T
187 181	Tannic Acid Tetrabutyl Titanate	\ <u>\</u> \\	2,5	Yes	<0.25 <0.25			25			: ×	₃ _~ ,	-		╁	× ×	DE .
TEC			Z es		18 37.		10 150	3 (, m ~	. 22		< ,×	· ××		7 4 5 6 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5		As titanium cmpd[β],toxic, irritates skin å eyes[9]
N.C.	Tetradecanol I-Tetradecene		22	Yes	<0.25	DNA		> >	23	33		××	+-	: ××	Yes		Irritates skin & eyes [9]
108 TED	Tetradecylbenzene Tetraethyl Dithiopyrophosphate		1 2	Yes	< 0.25 < 0.25	DNA 0.2*	.0.6*	(3)	(3)	1			1 ×	: * :	Yes		onc.
1EL	Tetraethyl Lead	-	2	Yes	82.7	0.007		3	·	+	+	 	×	, ×	100	مدا	
TEP 116	Tetraehty: Pyrophosphate Tetraethylene Glycol		22	No Yes	< 0.25 < 0.25	0.0039 DNA		٣	۳	•			× 1	· ×	7 - 4 - 4 - 4 - 4 - 4 - 4 - 4 - 4 - 4 -	4] [12] 5 Yes	
di i	Tetraethylenepentamine	_	S.	ě	T	DNA						. ,	×				tor required[10] for required[10] foxic, frrit, eyes & skin[5]
=	Tetrafluoroethylene (Inhibited)	.	2	Yes	> 760	DNA				-	-	-	×	×	ě	s Yes	ay be
- E	Tetrahydrofuran Tetrahydronaphthalene Tetramethyl pad		222	, des	1.3	2500	250	624	1 1	6.2	11	·	××	X	Yes		ates skin å e)
100	Thiophospene	-	2	<u>c</u> ,	┥.	0.003		,		-	,	_	×		<u>.</u>		Like tetraethyl iead [8] skin absorption occurs
TER	Thirds Thorace Thorium Nigrate	NNL	S S S	Yes S	1600.250.25	0 . T		€~	÷.	<u>e</u> -	1 ×	ı×	× -	: ×	, es	Yes	hosgene [8]
E	Titanium Tetrachloride		Yes		ر 5	vs.		m	ı	۳			·×	ا ×	<u>2</u>		low chemical toxicity
10	Toluene		유	Yes	51.7	100	150	2	F	-	+	_	+	×	2	101121	
101 1AP	Toluene 2,4-Diisocyanade P-Toluene Sulfanic Acid	-10	Yes	Yes	0.07	0.02 DNA		(3)	1 1	e ĝ				* * *	Yes	VČ[11 Yes	 Toxicity as o-toluene-
	o-Toluidine	7	9	Yes	8	5	10			~			-				_
TRC	Toxaphene Trichlorfun	v v	22	₽,	< 0.25	0.5* DNA		m(E)	r.E	2	-	 		× ×	Yes	Yes	ating a
TCT TCE	Trichloro-s-Triazinetrione Trichloroethane	s J	Yes	# o Yes	< 0.1	350	440	- 2	1 73	- X	x [6]	[6]x x	- ×	* * * · · · · · · · · · · · · · · · · ·	Yes	Yes	highly toxic [5]
TCL	Trickloroethylene	-	2	No[5]	129	00;	150	m		-		+	×	·	- E	No[12]	Causes cancer in mice car-
TCF	Trichlorofluoromethang		9	2	1190	1000		(2)	Ξ					×	=		4 00 000
TPH	Trichlorophenol 2,4,5-Trichlorophenoxy Acetic Acid	SS	5 E	Yes	< 0.25 < 0.01			<u> </u>	(3)					 	¥ es	Yes	as chlor.p
202	Tricresyl Phosphate Tridecanol	. ب. ب	Yes		< 760 < 0.25	6.5		22	1 2	22		××	-	* *	Yes	0 0	
100	1-Tridecene		2 2	Yes Y	< 0.25 0.14	4 8 0		-	>						Yes	Yes	nd ic
TEA	Triethanolamine	-	- 1	. !					}		-	-	_	_	2		.uw toxicity [5], eye i
TAL	friethylaluminum		Yes	Yes	0.1	DNA		1			× ,		Fi	re Suit	Yes t No	<u>«</u>	Low foxicity [5] Ignites spontaneously in air
TEN :	Triethylamine Triethylbenzene	بـ بـ	2 €	Yes	119	25	40	e =		m=			 	 :	Yes	0	irritant
						200		,	,						. Yes	Yes	Animal expers, indicate Ligh toxicity [8]

				-		7 7 1	TIV (note on	CAK	CAY_TOYIC	 	<u> </u>		E COM	ירונוש		MATERIAL MADATI-	
ں	3	S	≆	ш	9		MG/M3)*	F ZW	HAZARD RATING	<u>.</u>	12	PATOR	RESPIRATORY CLOTHII	COTHING	:2	BILITY [10]	
	3	,				1	SIE	×	55	æ	#	B	æ	8	22	В	NOTES
155 TET	Triethylene Glycol Triethylenetetramine	-1 -1	22	Yes	<0.25 <0.25	DRA		3	-	3	, ,	× 1	1 ><	. ×	<i>3</i>	Yes Yes Yes No	Low toxicity [5] Strong irritant to tissue,
TFC	Triflucrochloroethylene Trifluralin	0.0	¥e So	Yes	<760 <0.25	50 0 K		~ >	3 3		· ×	1 1	жı	1 1	**		An asphyriant, moderately toxic[5] Animal experts indicate low
¥.	Tritsobutya.uminum	_	Yes	res	0.4	DNA					1,	1.	×	e v	12.	Yes Yes	
£	Trimethy]amine	. e	Yes	Yes	>760	(5) (HC1)		2		, E	, ,	, ,	××	××) = 	Yes Yes	Toxic by inhal. burns skin [5] Toxic rating as silanes. 5 rong
= 1	i rime cny i cni oros i i ane				26.5	Traus (c)		3			1		: 1	: 1			irritant to tissues [5] Low tuxicity[5],not harmful
8 8	Tripropylene Glycol Tris(Aziridinyl) Phosphine Ox	~ ~	2 .	Ies -	VII. 63	¥ 36				 		1	×	×	+		Irritating skin & eyes, protect skin[9] highly taxic[8]
199	Turpentine Undecano. (1-) 1-Undecene	٠	222	Yes Yes	14 <0.25 1.2	201 28	150	2	2	2	1 1 1	×1×	×1,	1 1 1	× · ×	Yes Yes Yes Yes	Low toxicity[5],eye irrit.[9] Irritating skin & eyes [9]
200	n-Undecylbenzene Uranium Peroxíde	-20	,	Yes	<0.25	DNG 0.2*	0.6*				,	\	×	×	 .	_	Recognized carcinogen,as uranium
33	Uranyl Acetate Uranyl Mitrate	NN	ક ક	Yes	<0.25 <0.25	*\$0 *\$0					1	,	×	×	»» •		Highly toxic [5]
종홍	Uranyl Sulfate Urea	νν	22	Yes No	<0.25 DNA	0.2*(U)						, ,	× ,	× 1		Yes Yes	See uranium oxysulfide Essentially nontoxic[8], not harmful [0]
2	Urea Peroxide	S	%	Yes	ANC	DNA		,	,	2	×	×	,	,	×	Yes Yes	Decomposed by moisture at temp. above 40°C [5]
Ą	Valeraldehyde	-	O _R	Xes	69	5.0		2	•	-	-	~	×	-	~	-	Low toxicity
Y01	Vanadium Oxitrichloride		Yes	2	~40	5 (HC1)		3	•	(3)	•	,	×	×	<u>-</u> -	Yes	Hydrolyzes [5,8], toxic rating
YOA	Vanadium Pentoxide	s	₽	2	< 0.25	DIKA					1	,	×	×	- 1	Yec Yes	Toxic rating as vanadium,toxic y inhalation,variable toxicity
VSF	Vanadyl Sulfate Vinyl Acetate	2	₽.	No Yes	<0.25	DNA 10	20		•	-	1 1	,×	× -	×ι	1 ×		May be toxic & irritant [5]
VCF VEF	ther	9/1	윤윤		>760 >760	O MA		2 2	, ,	2 3	, .	,	×		- 	res res	Carcinogen; no allowable exposure Toxic rating ethers; acute
VF1	Viny) Fluoride (Inhibited) 6 Vinyl Methyl Ether (InhibitedL/G	9/1/6	<u>2</u> 2	Yes (es[5]	×76.) ×760	2.5*[^r](F)		(E) u	13	(E) a	1 1	1 1	××	> 1	1 ×		Toxic rating as fluctides Irritates eyes, nose, throat, liquid causes frostbite [9]
I)	Viny;idenechloride (Inhibited		£	Yes	∿760	01	20	(2)		(2)	- -	-	×	×	,	Yes Mo	Toxic rating as vinyl chloride
YKT	Vinyl Toluene	_	£	Yes	3.5	100	150				1	,	×	,	×	<u>~</u>	Mod. toxic [5],chronic exp.
VTS MCA	Vinyl Trichlorosilane Wax: Carnuba	-1-	Yes	Yes	146	S (HC1)		m0	. 0	m 0	, 1	1 1	× ı	×ı		Yes No	Fusing liquid
XLM XLM	Max: Paraffin Xylene (M-)	1/s	윤 윤	Yes	<0.25 17	2* 100	150	2 0	2	0-	1 1	: ×	· ×	1 1	, ×	Ves Yes o(PVC)No[12] (111 ACE(121	Fumes, carcinogen of stomach
K K	Xylene (0-) Xylene (P-)		₽₽	Yes	2 8	88	150 150	22	22		, ,	××	××	1 1	××		
X	Xylenol Zectran	SS	<u>₹</u> %	že Š	0.47	45 AMO		(£)	E	E		 , ,	× ×		<u> </u>	res Tes	Toxic rating as phenol Insecticide
ZWA	Zinc Acetate	· s	윤		<0.25	PA .					× :		<u> </u>	1		Yes Yes	Toxic rating as Zn cmpd. Low toxicity[8], irrit.eyes, nose, eye
ZAC	Zinc Amenium Chloride	S	S	2	< 0.25	ANG				7	×	-	-	-	×	_	Irrit. skin,eyes,mose,throat[9]

		MOTES	18	Carcinogen, toxic rating as Cr cod	loxic rating as boron empas;	Low toxicity[5], toxic rating as	Montoxic [5], traces added to	Fumes highly toxic. Soln. irrit.	Carcinogen, toxic rating as Cr ced Toxic by Italia indest. [5].	Lube oil additive[5],skin 5	tissue irritant [9] Toxic rating as fluorides [8]	toxic	Zinc compounds for toxicity,	toxic rating as sulfites[8] Nod. toxic[5],toxic rating as	Toxic by ingest.[5],eye,skin incl	Toxic by ingest. Zinc chromate	is carcinogen Toxic rating as fluosilicates	Low toxicity[5],skin tissue irrit Toxic by inhalation [5], skin,	tissue irritant [9]	loxic rating as nitrates, skin, tissue irritast [9]	Mo toxic data [5,8] Toxic & irritant [5], toxic	fly as Zr. eye, skin frrit. [9]	Toxic rating as HCl acid,									
MATERIAL COMPATI-	BILLTY [10]	۵	Yes	.;	î	ç,		Yes						š	25		Š	Š ģ		les	ž.	Yes		 -				 	-			
28	3 🛱	8	Yes	,	<u>.</u>	Š		Yes	Yes					Yes	Yes		Yes	Yes Yes		S	Yes	Yes							-			
 <u>8</u>	CLOTHINS	\$	-	1		×	•		1 1	_	١	· ×	_	×	100	k	1	××	_	<	× 1	×	-									•
**************************************	RESPIRATORY CLOTHII	8	×	× 1	_	<u> </u>	'		× ×	-		× 1	Ļ	<u>'</u>	- 1 1	<u> </u> '		' '	_	'	·×	1	×	 _				 				
LS OF	ATTORN	SA	×_	× 1		<u> </u>			~ ×	 	× ;	-	<u> </u>	<u>'</u>	1×		×	<u>' '</u>	\downarrow			<u>'</u>	×				_	 				
LEVE	RESPIR	ន	-	٠,>		×	•			+		· ×	<u> </u> '	*		<u> </u>	<u>'</u>	<u>'</u>	4	•	1 1	<u>'</u>	1	 _					-	· -		
 	-	R.	_		_	×	<u>~</u>			-			*	×	× 1			× ×	1	₹	× ·	×	_	 -		··	- -	 				
	MING	S S	(2)	(3)	_	= -			<u> </u>	 		<u> </u>			(2)	+	(3)		-		(3)		<u>e</u>	 1			\downarrow	 				
SAX-TOXIC	윤	×	(3)	33		9			(3) (3)			- <u>-</u>	(2)	(2)	(3)	+-	(3)		(6)			_	(3)	 +			+	 	+			
, 	=	-	_		_	1				-		- <u>:</u>				-	ت —		+	- 	ت 		<u></u>	+	·		-	 	-			, ,
TLV (PPM OR	MG/M3)*	뜅																														
77	HG/	¥	0.013	0.1	: ;	Y Y	NA.	<u>:</u>	0.05*	DWA		DMA	VMO	DNA	蠶	DIA	2.5%	(12) +S	(4 (72)	(47) -6	5* (Zr) 2.5*[5](F)	5* (2r) .	(z) [s]*s					 		-	<u>-</u> -	
	9-		<0.25	.0.25		37.5		<0.25	<0.25	<0.25	.0.25	6.63	SE SE	<0.25	29.25 28.25	¥	8	6.23 6.23	2, 0,	 	¥ ¥	20	A A	1			+-	 				
	ш.		₽	22	: ;	2	₽	2	22	Yes	2	2			22	-		운 윤	-	 ≩	٠ <u>٠</u>	윤		 +			†-	 				
	¥		ş	2 2	_	†	₽		₽ ₽	£	2:		₽.	<u> </u>	No.			£ £	┥		2 <u>2</u>	9	Yes	 \dagger			╁-	 		-		
(S		s	S		1	s	S	s s		٠.,	S	S	·,	v.	~	٠,٠	~ v	1	,	n n	v	<u>-</u> -	 +			-	 	-			
5	5		Zinc Arsenate	Zinc Bichromate		Zinc Browide	Zinc Carbonate	Zinc Chloride	Zinc Chromate Zinc Cyanide	Zinc Dialkyld:thiophosphate	Zinc Fluoberate	Zinc Formate	Zinc Hydrosulfite	Zinc Witrate	Zinc Phenolsulfonate Zinc Phoschide	Zinc Potassium Chromatu	Zinc Silicofluoride	Zinc Sultate	Zircosia Kitrata	anpinim millioni7	Zirconium Oxychloride Zirconium Potassium Fluoride	Zircenium Sulfate	Zirconium Tetrachloride									
	ں 		ZAR	78C	-	4	208	זָני זַני	ZCR	d02	ZFB	ZFR	SH2	ZNT	33 5	ZPi	ZSL	107 107	710	17	700 7.00 7.00	702	77									-

APPENDIX B

MANUFACTURERS AND SUPPLIERS OF PROTECTIVE EQUIPMENT

PROTECTIVE CLOTHING HOODS/GOGGLES/FACESHIELDS

Name and Part No.	<u> </u>	1anu fac i	turer	<u>Material</u>
Hood #460-625 (Air Supplied)	Fabohio,	Inc.		Vinyl
Hood #420-620	и и			Viny1
Helmet (Acid Helmet) 15836	Wheeler P	rotective	Apparel, Inc.	Fiberglass with Vinyl window
Hood Three Piece Style F-100-80	Norton Sa	fety Proc	ts Div.	Neoprene/Nylon
Hood, Three Piece Style 98-300	u	и	ti.	Polyurethane/Nylon
Hood, Two Prece Style NSVP-80	11	u	n	Vinyl/Nylon
Hood, Three Piece Style 58-00		ц	и	Vinyl/Nylon 14-15 oz
Goggle, Flexible Frame Style 1022 (Chemical Splash-High Humidity)	н	H	н	
Faceshield, Model 7-10	n	et .	u	Acetate
Chemgard Acid Hood (Air Line Supply Type) Cat. No. 79168	Mine Safet	y Applia	nces Co.	Vinyl/Cotton
Chemgard Acid Hood Cat. No. 38022	u	11	u	Vinyl/Cotton
A-1-A Firefighter Hood Cat. No. 75514	11	+1	11	Glass Fiber
Skullgard Faceshield Fra	ame "	"	u	Cellulose Acetate
Chemical Splash Goggle Cat. No.791079,clear Cat. No.791082,yellow	Mine Safet	y Applia	nces Co.	Polycarbonate Polycarbonate
Wide Vision Goggle Cat.No. 791081	и	H		Polycarbonate
Rubber Frame Goggle Cat. No.791072,clear Cat. No.791072,with breathing vent	ч	" B -	- 2	Polycarbonate Polycarbonate

Goggles, Encon 160 EN Fog Lens-0.06"Poly- carbonate, Pt. No. 01-4000-51	Encon Manu	facturi	ng Co.	Vinyl Vinyl/Neoprene Strap
Regular Lens-0.06" Polycarbonate, Pt. No. 01-400-55	n .	1		Vinyl/Neoprene Strap
Hood, Air Supplied Series 6000	Rich Indus	tries		Vinyl
Acid Hood	Rich Indus	tries		Vinyl
Splashproof Hood #H-516(Cape Style)	Jomac Produ	icts, li	nc.	Vinyl/Fiberglass
Hood, SD-3672 Series	Standard Sa	afety E	quipment Co.	Gralite
Hood, Self Ventilating SD-3884	11	16	п	Gralite
Hood, Shoulder Mounted SD-5424	11	н	и	Gralite
Mounted Hood SD-4196	u	ıt	11	Gralite
Hood, SD-3943 (for Scottoramic Face Mask)	11	Ħ	u	Gralite
Acid Hood with Hard Cap,SD-4435, SD-4432, SD-4107	и	11	ч	Gralite
Airborne Dust Hood SD-3478,SD-5308	u	u	ч	Gralite
Hood, Shoulder Length #186-11	Encon Manuf	acturin	ng Co.	Neoprene/Nomex
Style Hood #5361-11	11	11	и	Neoprene/Nomex
Style Hood/Safety Cap #6410-11	11	11	и	Neoprene/Nomex

PROTECTIVE CLOTHING BOOTS/SHOE COVERS

Name and Part No.	į	<u> 1anufact</u>	urer	Material
"TV" (TRI-VAC) 2TV04010(16") Steel Toe	Boss Manu	ıfacturing	Natural Rubber Neoprene or Vinyl	
Rubber Knee 2TV-4010(16") Steel Toe	Boss Manu	ıfacturing	H u	
Industrial Rubber Hip 2HS-6231 Steel Toe	Boss Manu	ıfacturing	Black Rubber	
Boots 320-840	Fabchio,	Inc.		Vinyl
Overshoe 68974	Wheeler P	rotective	Apparel Inc.	Vinyl
Leggings/Spats Leggings, Style 57-00 (Hip Length)	Norton Sa	fety Produ	ucts Div.	Neoprene/Cotton
Spats, Style 57-01				Neoprene/Cotton
Shoe Cover,Booties Series 1000	Rich Indu	stries		Vinyl
Leggings, Hip Length Cat.No. 37690(29") Cat.No. 37691(30") Cat.No. 37692(31")	Mine Safe	ty Appliar	Neoprene Latex/Cotton	
Spats, #37689	Mine Safe	ty Appliar	ices Co.	Neoprene Latex/Cotton
Seal-Tight Boots (with steel toe, for use with Standard Safety complete en- closure garments)	Standard	Safety Equ	Neoprene	
Gralite Boots SD-3528	и	n	II	Gralite
Gralite Waders SD-4139	II	н	ti	Gralite-20

Gloves/Guantlets/Sleeves

Name and Part No.	Manu	ufacturer		<u>Material</u>
Tyvek Sleeve No.08-004518 (Throw-Away)	Singer Safe	ty Products	•	Tyvek (Polyethylene)
Disposable Poly- ethylene Glove 36-001151(small) 36-001152(medium) 36-001153(large)	п	н	u	Tyvek (Polyethylene)
Black Stanzoils IUN-0054(.030 gage) 14 in. long	Boss Manufac	cturing Co.		Neoprene Coated
Heavy Weight Safety Cuff 1SP-5115	н	н	u	Vinyl Coated
Tough Wear 1UH-0343(.024 gage) 12 3/4 in. long	ti	1	II.	Nitrile Rubber
Interlocking Glove/ Ring Assembly,Acidwean 150000 (Sleevlets,rings,glove	•	tective App	parel, Inc.	Vinyl
Sleeves, 55-00	Norton Safe	ty Products	Div.	Neoprene/cotton
Glove, Super Flexible Knitwrist Cat.No. 3764 12" Gauntlett Cat.No.	12	Appliances	s Co.	Vinyl/Jersey Lining
Glove, Python Knitwrist Cat.No. 3799 Gauntlett(14")Cat.No.3		H	п	Neoprene/Canton Flannel Lining
Glove, Butyl Rubber, Cat.No. 38554(size 9) Cat.No. 38555(size 10) Cat.No. 38556(size 11))	п	П	Butyl

Redmont Gauntlet (Heavy Duty Line)	Edmont-Wilson Div. Bechton, Dickinson & Co.	Neoprene
Grappler, 14" Gauntlet Premium PVC Line #05619	, " "	Vinyl Coated
PVA, 14" Gauntlet Fully Coated, #14306	15 16 11	Polyvinyl Coated Alcohol
Coated Gloves, PVC Heavy Duty #9112,12" Gauntlet #9114,14" Gauntlet #9116,16" Gauntlet #9118,18" Gauntlet	Jomac Products, Inc.	Vinyl
Gloves, SD4313, SD5525	Standard Safety Equipment Co.	Graylite

PROTECTIVE CLOTHING JACKETS/SHOP COATS

Name	Man	<u>ufacture</u> r		Material
TYVEK 40" Shop Coat (Disposable) No-08-004701-Small through No.08-004704-XLarge (Knee Length)	Sinyer Safe	ty Produc	ts	Tyvek (Polyethylene)
Coat, Acidwear 30036	Wheeler Pro	tootive A	pparel, Inc.	Vinyl
Jacket Style F-100-05 Tufflon F-100	Norton Safe	ty Produc	ts Div.	Neoprene/Nylon
Coat, Style F-10G-2310 Tufflon F-100	11	16	H	Neoprene/Nylon
Coat, 49" Style 92-348 Super Rhind-Lite	И	II		Polyurethane/Nylon
Jacket, 30" Sewn on Hoo Style 90-344	d "	tt	н	Polyurethane/Nylon
Hooded Jacket Style NSPV-107	и	II.	ıı	Viry1/Nylon
Coat, 49" Style NSVP-2348-1	11	н	u	Vinyl/Nylon
Laboratory Smock	Rich Indust	ries		Vinyl
Series 3680 Shop Coat/Splash Re- sistant, Heavy Duty, Style 55-415	Edmont-Wilson Div. Becton		on & Co.	
Jacket 32", No. J-340 with zipper, #3 Fabric	Jomac Produc	cts, Inc.		Vinyl/Cotton
Coat, basic, 44" No. C-440, Series 610	N .	: 1	п	Vinyl/Nylon
Fireman's Bunker Coat 60Z, #7500-5	Encon Manuf	acturing	Co.	Neoprene/Nomex
Jacket, Style 5600 #5600-11	u	II II		Neopreme, Polyfoam, Nomex-Cotton Lined

Open Back Jacket SD-4499 (Seal tight sleeves) SD-4178	Standard	Safety	Equipment	Co.	Gralite 20 Hi-Glo
Chemical Jacket(30") SD-4354	μ	n	н		Hi-Glo
Coats/Jackets, Side Closure, SD-3109 SD-4390	11	11	и		Gralite-20 Hi-Glo
Coats/Jackets, Front Closure, SD-4354 (Seal tight sleeves)	tt	н	4		Hi-Glo
P.O.M (Pyrophoric organic materials) Jacket, SD-4364	н	11	44		Gralite-20
Switch Pullers Jacket SD-4300 SD-5532	н	11	u		Gralite-20 Winter Glo-20

Coveralls/Overalls/Pants

Name and Part No.	<u>Ma</u>	inufacturer		<u>Material</u>
Tyvek Coverall (Disposable) No.08-004071-74 small-extra large (full length)	Singer Pro	oducts Co.		Tyvek (Polyethylene)
Overalls (bib) 3PN-0871	Boss Manuf	facturing (Co.	Vinyl/Nylon
Coverall with Sewn in Body Harness 47073E	Wheeler Pr	rotective /	opparel, Inc.	
Overall, Acidwear 48536	11	и	и	Vinyl
Coverall Style F-100- 615, Tufflon F-100	Norton Sai	fety Produc	ts Div.	Neoprene/Nylon
Overall, Bib, Style F-100-12	ı	II	н	Neoprene/Nylon
Overall, Bib, Style 91-325, Super Rhino-Lii	te "	11	II .	Polyurethane/Nylon
Waist Pant, Style 91-36	50 "	•	u	Polyurethane/Nylon
Overall, Style NSPV-12	tt	n	u	Vinyl/Nylon
Coverall, Style 56-15	15	и	n	Vinyl/Nylon 14-15 oz
JFA Heavy Duty Coveral Style No. 55-510		lson ton, Dickir	nson & Do.	Polyolefin
Pants, Bib P-010 #3 Fabric	Jomac Prod	ducts, Inc.		Viny1/Cotton
Coverall, CO-040 #3 Fabric	II	18		Vinyl/Nylon
Coverall, Style 5500 #5500-11	Encon Man	ufacturing	Co.	Neoprene/Polyfoam Nylon/Cotton Lined

Pants, Style 5610 #5610-11	Encon Ma	nufactur	Neoprene/Nylon	
Coveralls SD-3981(seal-tight sleevcs) SD-4178	Standard	Safety	Equipment Co.	Gralite-20 Hi-Glo
Anti-Contamination Coverall,SD-574	•	н	и	Cotton
Air Supplied Coverall 8 Mil, SD-600	"	14	u	Plastic (Vinyl)
Bib Overall SD-3110 SD-4046	u	н	u	Gralite-20 Hi-Glo
Rocket Fuel Handler's Coverall SD-3663	u	и	11	Gralite-20

Aprons

Name and Part No.	Manufacturer	<u>Material</u>
Bib Apron No. 08-004136	Singer Safety Products, Inc.	Tyvek (Polyethylene)
Apron, Frost Clear	Fabohio, Inc.	Vinyl
Apron (Bib Type) 40136	Wheeler Protective Apparel, Inc.	Vinyl
Apron, Style 54-10	Norton Safety Products Div.	Neoprene/Cotton
Apron, Style 54-30	n n n	Neoprene/Nylon
Apron, Style 34-40	n n u	Hycar Rubber/Cotton
Apron, Style 64-30	и и ч	Vinyl/Nylon
Chemgard Plastic Apron Cat. No. 73124 (Med.Weight, 0.008")	Mine Safety Appliances Co.	Vinyl
Synthetic Rubber Coated Apron Cat. No. 37668 (Med. Weight)	. 41 11	Hycar Rubber/Cotton
Synthetic Rubber Coated Apron Cat. No. 31049 (Heavy Weight)	u u u	Buna S Rubber/Cotton
Synthetic Rubber Coated Apron Cat. No. 33441	u u n	Neopreme Rubber/Cotton
Vinyl Coated Apron Cat. No. 37673 (Med. Weight)	u n u	Viny1/Cotton
Apron, Series 1000	Rich Industries	Vinyl
Heavy Duty 20 Mil Vinyl Apron, 45 #73640, Style 54-733	Edmont Wilson Div. Bector, Dickinson & Co.	Vinyl
Reinforced Heavy Duty Apron, 23 Mil, #72261, Style 54-367	u u u	Hycar

Apron, Bib Style AW-3-36, #3 Fabric	Jomac Products,	Inc.	Vinyl
Apron, Bib Style	Standard Safety	Equipment Co.	Hi-Glo
Apron, Split-Type	n u	ıı	Hi-Glo

Suits, Unencapsulating

Name and Part No.	<u>Manufacture</u> r	<u>Material</u>
Fire Chemical Suit (new, no part no.)	Fyrepel Products, Inc.	Butyl with aluminized oversuit
Disposable Suit #237 (one piece)	Fabohio, Inc.	Polyethylene, Span bonded or extended poly thylene
Acid Suit #212 (one piece)	at N	Vinyl
Coverall #47036 (with 15836 Hood)	Wheeler Protective Apparel, Inc.	Vinyl
Two Piece Suit Series 3900	Rich Industries	Vinyl
Acid Suit, Two Piece Series 3675	u 11	Vinyl
Heavy Duty Overali JFA	Edmont-Wilson	Polyolefin
Sun-Ray Suit(Two-Piece) Cat.No. 37107 (small) Cat.No. 37108 (medium) Cat.No. 37109 (large)	Mine Safety Appliances Co.	Viny1/Nylon
Rubber Utility Suit (Two Piece) Cat.No. 32605 (small) Cat.No. 31934 (medium) Cat.No. 32606 (large)	Ik U II	Synthetic Rubber
Rubberized Protection Suit 'One Piece) Cat.N. 33495 (small) Cat.No. 33436 (medium) Cat.No. 33437 (large) Hood for above, Cat.No.	. 33370	Neoprene
POM Suit, #7550	Encon Manufacturing Co.	Aluminized Nomex

Dust Suit SD-2632 SD-4475	Standard	Safety	Equipment	Co.	Cotton Dacron/Rayon
Liquid Oxygen Suit,SD-3779	tt	II	н		Whiteside
Two Piece Disposable Plastic Suit	ıı	16	44		Plastic

Encapsulating Suit (Air Supply/Air Respirator Req'd)

Name and Part No.	<u>Manufacturer</u>	<u>Material</u>
Basic Coverall #CO-910 #2 Chemical Cloth	Jomac Products, Inc.	Vinyl/Nylon
One Piece Suit Zipper, 0.020" Tk. Series 4000	Rich Industries	Vinyl
Rocket Fuel Handler Suit Cat.No. 92682 (Ai. Supp Cat.No. 38590 ir Mask	Mine Safety Appliances Co.	Vinyl
ChemPruf Suit with Hood Cat. No. 75096(small) Cat. No. 75097(medium) Cat. No. 75098(large)	Mine Safety Appliances Co.	Buty1/Nylon
Disposable Suit,Airfed Cat. No. 91783 (large) Cat. No. 91784 (medium)	u . 1 u	Vinyl
Acid King #47836C(with back pouch #47836H(without pouch)	Wheeler Protective Apparel, Inc.	Vinyl/Betacloth
One Piece Air Supplied Suit #13-20(0.020")	Fabohio, Inc.	Vinyl
One Piece Air Supplied Suit #13-10,100Z Nylon	u u	Viny1/Nylon
Two Piece Air Supplied #15-6,0.006" Vinyl #15-12, 0.012" Vinyl #15-10, 100Z Nylon	u i	Vinyl Vinyl Vinyl Vinyl/Nylon
Proximity Suit #FD-7599-V	Encon Manufacturing Co.	Neoprene/Nylon
Standard Acidmaster Suit, SD-3711 SD-5520	Standard Salety Equipment Co.	Gralite-20 (Vinyl) Winter Glo-20

Standard Compressed
Air Suit Standard Safety Equipment Co.
SD-3896
SD-5542(seal-tight bonts)

Standard Accordian
Suit
SD-5196
(Vinyl Overboots)
SD-5256 (Neoprene
Steel Toe Bnots)
SD-5563 (Neoprene
Steel Toe Boots)
Winter Glo-20

Winter Glo-20

Winter Glo-20

APPENDIX C

MATERIAL COMPATIBILITY TEST DATA (Task V)

(References listed by number in the tables are included in the Bibliography)

Legend:

L	-	Liquid
S		Solid and soluble
Ğ		Gas
M I	-	Slightly soluble
I	-	Insoluble
-	-	Dash indicates no testing performed
ВD		Below detection
[]	-	Water reactive, pertains to parent
		compound, literature data
R	-	Performance rating based on literature review
Т	_	Performance rating based on laboratory
•		tests of parent compound or water re-
		action product
T - D	_	Performance rating based on inability
, ,		to decontaminate
ND	_	No data
.,,		110 000

INTRODUCTION

A literature search revealed that at least 403 chemicals on the CHRIS list would require a fully encapsulating suit/and or a supplied air or self-contained breathing apparatus. A further review of the literature was conducted to determine whether butyl rubber and polycarbonate, materials used in the Hazardous Chemical Protective Clothing Outfit (HCPCO), were compatible with those chemicals requiring a fully encapsulating suit.

Data on the compatibility of polycarbonate with the chemicals was accepted without further consideration, and compatibility testing was limited to those chemicals for which no compatibility data was available. In the case of butyl rubber, a rating of incompatible was accepted. Additional compatibility testing was conducted when a compatible rating was listed, due to the uncertainty of the test methods used in the past, or when no data was available.

Butyl rubber and polycarbonate were both subjected to 3 hr swatch immersion tests. If gross degradation was observed after the 3 hrs, the materials were judged to be incompatible with the chemical and no further testing was done. When the materials showed no degradation after 3 hrs, they were reexposed for a total of 20 hrs. If no degradation was observed after 20 hrs exposure, polycarbonate was deemed compatible, and butyl rubber was subjected to penetration testing. Decontamination effectiveness of a detergent-water solution (butyl only) was also evaluated.

This Appendix contains information which will be useful for decision making by the On-Site Coordinator in selecting appropriate protective gear for Strike Team personnel. A few examples are given to assist the user in interpreting the information.

Acetaldehyde is a liquid and is not water reactive. Review of the literature (R) indicates that polycarbonate is not compatible with acetaldehyde. Laboratory tests (T) showed that acetaldehyde penetrates butyl rubber in less than 3 hrs exposure. Neither polycarbonate nor butyl rubber is recommended for use with acetaldehyde.

Acetic anhydride is a liquid and reacts with water to form acetic acid. The literature (R) indicates that polycarbonate is compatible with acetic anhydride. The literature also indicates that polycarbonate is compatible with acetic acid, a water reaction product of acetic anhydride. The survey revealed that butyl rubber

is compatible with acetic acid. Polycarbonate and butyl rubber can be used with the parent material, acetic anhydride, and its water reaction product acetic acid.

Acetonitrile is liquid and is not water reactive. Polycarbonate is compatible with acetonitrile according to the literature (R). Butyl rubber was found, in laboratory tests (T), to resist penetration for 6 hrs, but penetration did occur in less than 20 hrs. Polycarbonate can be used with acetonitrile, but butyl rubber will have limited usage, e.g., a single spill cleanup.

According to the literature (R), polycarbonate is compatible with acrylic acid. Laboratory tests showed that butyl rubber resisted penetration for up to 20 hrs, but decontamination with a water-detergent solution was ineffective (T-D) in removing acrylic acid from butyl rubber. Polycarbonate and butyl rubber can be used as protective gear during clean up of acrylic acid spills, but due to decontamination problems the butyl suit would be limited to single usage.

Appropriate materials of construction for a protective garment could not be assigned for all 403 chemicals. Additional research is being conducted to determine alternate materials which could be used for a protective garment. As this information becomes available, it is anticipated that the information contained herein will be updated. Meanwhile, On-Site Coordinators and Strike Team personnel should fine this information of value in assessing applications of the present HCPCO protective gear.

			WATER		POI YCARR	155	SWATCH TEST	EST					₹	MULTIPLE			
CHR1S CODE	CHEMICAL NAME	or-∢r	-CAE	3	WEIGHT		FH] ~ Z = 5	PHYS. ERTY NGE	Mad de	PENETRATION TEST ppm or mg*	SINGLE DECON TEST		TEST CYCLES	PERFORMANCE RATING	JANCE NG	
		ш		m	(*) hr 20 hr		3 hr 20 hr	r Tensile	()	3 br	20 hr		3 hr	2	POLYCAR- BONATE	BUTYL	
AAD AAC ACA ACY ATN	Acetaldehyde Acetic Acid Acetic Anhydride Acetone Cyanohydrin Acetonitrile		 	22722	1 1 1 1		- MS ACETIC		 	.08 88 88 4.08 88 88	- 80 80 80 0.9	PASS PASS PASS	1111		FAIL(R) PASS(R) [PASS](R) PASS(R) PASS(R)	FAIL(T) PASS(T) [PASS](T) PASS(T) PASS(T) >6<20 hr(T)	
ARIL AARIL ACC ACC ACC ACC	Acetyl Bromide Acetyl Chloride Acrolein Acrylamide Acrylic Acid			>>ZZZ		FORMS	MS HYDE	ORMS HYDROBROMIC SEE HYDROCHLORIC	AC10 AC10	67 1.2 80 80	- ' ' ' 88	PASS FAIL	FAIL	FAIL	[PASS](R) [FAIL](R) PASS(R) PASS(R)	FAIL(T) FAIL(T) FAIL(T) PASS(T) FAIL(T-D)	
A P P P P P P P P P P P P P P P P P P P	Acrylanitrile Adiponitrile Aldrin Allyl Alcohol Allyl Chloride			ZZZZZ	1 1 1 1 1				1 1 1 1 1	5.0			1 (1 1 1		PASS(R) PASS(R) PASS(R) PASS(R) FAIL(R)	FALL(T) FALL(T) PASS(T) FALL(T) FALL(T)	
ATC ACF ACL ALF ALM	Ally) Trichlorosilane Allyl Chloroformate Aluminum Chloride Aluminum Fluoride Aluminum Sulfate	77000		:	0.0	~~~	E HYDRO CTS SLO E HYDRO 0 +0.3	SEE HYDROCHLORIC ACID SEE HYDROCHLORIC ACID 0.0 +0.3 0.0 0.0	ACID 720 to HC ACID 0.0	HCI 83			1 1 1 1		[FA1L](R) [FA1L](R) [PASS](R) PASS(R) PASS(T)	FAIL(T) FAIL(T) FAIL(T) PASS(R) [PASS(R)	
AMA ABF ASU ACH ACH	Ammonia Anhydrous Ammonium Bisulfite Ammonium Bisulfite Ammonium Carbamate Ammonium Chromate	00000	×××××	19990	.2 +0.5 .0 +1.8 .0 +0.1	5 +0.9 8 +0.7 1 0.0	7 +16.1 9 +8.0 7 +0.9 0 +0.5	-8.7 -2.0 -2.0 -2.3	1.6	8 1 1 1 1	8888	FAIL FAIL FAIL PASS PASS		1:1:1	PASS(R) PASS(T) PASS(T) PASS(T) PASS(T)	PASS(T) FAIL(T-0) FAIL(T) PASS(T-0) PASS(T)	
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C			┝╾┵╾	WATER		POL YCARB	-	SAT	SWATCH TEST	EST BUTYL			DESETTORITOR		₹	MULTIPLE DECON		z	
T U U U U U U U U U	RIS	CHEMICAL NAME	ν – «		L	WE15	 =;	HETG	景	20 hr PROPE	PHYS.	T Med	EST or mg*	STNGL		TEST	PERFORMAL RATING	PERFORMANCE RATING	
Amenorius Dichromate S S N + 0.1 +0.1 +0.2 +0.2 +1.7 +2.0 - 80 PASS - Amenorius Dichromate S S N + 0.1 +0.1 +0.2 +0.2 +1.7 +2.0 - 80 PASS - Amenorius Dichromate S S N + 0.1 +0.1 +0.2 +0.2 +1.7 +2.0 - 80 PASS - Amenorius Dichromate S S N + 0.1 +0.1 +0.2 +0.2 +1.7 +2.0 - 80 PASS - Amenorius Dichromate S S N + 0.1 +0.1 +0.2 +0.2 +1.7 +2.0 - 80 PASS - Amenorius Dichromate S S N	Ä		-	<u>د</u> د	·			3	ž.	E S	394			TES		8			
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Name Silicofluoride S S N	æ ×	Ammonium Fluoride Ammonium Oxalate	n vi	n v1	= Z		, ,					٠,	32	PASS			PASS	<u> </u>	
Margorium Suffide	동	Ammenium Silicofluoride	S	S	z		,		•	1	•	ı	•		<u>.</u>	,	PASS(R)	PA6S(R)	_
N-Awy Trichlorosilane	ايرا	American Sulfide	 !	; ! •	Z			,				1	,			1	PASS(R)	PASS	_
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Polytical Mirate	<u> </u>		i — I	, >		(¥)	,	(F)	CHARGE SERVICE	يب	_		12.51	-	8		
Partition Without the beautified S			w	ස්					(2)		3 17	20 hr				POLYCAR-	BUTYL
Pass			1	\dagger	+	3 hr 20 t	ml_		lens 1 le	lear						DOWNIE	
Beryllium Mittate S																	,
Beryllium, Metalic S I N	BEN	Beryllium Nitrate	s	,		· •		SEE N	TRIC ACT		1			,	,	[PASSIR)	
Bisphenol A Diglycidy Ether L N	BEO	Beryllium Oxide	s		z :	,			1		,	,				PASS(R)	PASS(R)
Bisphenol A Diglycidy Ether L N	BE	Beryllium, Metallic	<i>^</i> •	-	Z 2			ı		•		,		•		PASS (K)	PASS(K)
Bisphenol A Diglycidy Ether L - N	BPA	Bisphenol A	n vs			1 1				1		1 1				PASS(R)	PASS(R)
Brown trichloride	1		-			!	i	į .	1 2	!	!					Ç	9
Brown Trichloraide	100	bisphenol A DigiyCldy Ether Roiler Compound ionid	٠.	•	= Z	-		UMAYA	LASE	, 1	, (, ,	, ,	, ,		PASS (R)	PASS (R)
Browning 2 2	Borner Tribromide	1 -	_				DRMS HY	ORUBROMIC	ACID	,		•	,	,	I PASS 1 (R)		
Brownine Pentafluoride	88.	Boron Trichloride	_	_	. >-	<u>'</u>		EE HYD	ROCHLORIO	ACID	1	,		,	,	[FAIL](R)	FAIL (
Brownine Pentafluoride	BRM	Bromine		ī	z	•	1	,	í	1	,	ı	•	,	,	FAIL (R)	
December Trifluoride	BPF	Bromine Pentafluoride		•	<u></u>		<u>.</u>	SEE HYD	ROFLINORIG	ACID		! .	: : :	ı	,	[FAIL](R)	FAIL (T)
Tert-Buty1 Hydroperoxide	BTF	Bromine Trifluoride	_	,	>	'	_	SEE HYD	ROFLUORIC	ACID	ı	,		,		2	FAIL (T)
N-Bury Amine	묾	Tert-Butyl Hydroperoxide	-	-		•	1				8	8	PASS	'		PASS (R)	
Butyl Trichlorosilane	5 C	N-Buty Amine 1 2-Rutylone Ovide	- L	l i		-	1 1	•			36.			, ,	1 1	DACK (B)	FA11 (T)
Butyl Trichlorosilane L - Y	2	1,2-bucy icine online	1		-						3 :	1	i	1		(N) 250 :	
1,4-Butynedio 1,4-Butynedio 1,4-Butynedio 1,4-Butynedio 1,4-Butynedio 1,4-Butynedio 1,7-Butynedio 1,7-	BCS	Butyl Trichlorosilane	_	1	<u>~</u>	•	<u>~</u>	E HYD	ROCHLOR 1¢	ACID	ı	١		,	,	(FA1L] (R)	
N-Butraldehyde	910	1,4-Butynediol	<u>.</u>	S	z	-	ا 	1	-	,	œ .	<u>ස</u>	PASS	·		~	PASS (T)
Cadmium Bromide Cadmium Bromide S S N BD BD PASS - PASS (R) Cadmium Bromide S S N BD PASS - PASS (R) Cadmium Chloride Cadmium Nitrate Cadmium Oxide S I N BD PASS - PASS (R) PASS (R) Cadmium Oxide	BAD To	ISO-Butraldehyde	- L	1		בעבטני	- 1	113			/ ye		•	1			FAIL (1)
Cadmium Bromide S S N - - - BD PASS - - PASS (R) Cadmium Chloride S S N - - - - BD PASS - - PASS (R) Cadmium Fluoborate S S N - <t< td=""><td>S.</td><td>Cadmium Acctate</td><td>1 0</td><td>· v</td><td> : z</td><td>1</td><td>: ,</td><td>-</td><td>. 1</td><td>1</td><td>2 8</td><td>08</td><td>PASS</td><td>, ,</td><td></td><td></td><td>PASS (T)</td></t<>	S.	Cadmium Acctate	1 0	· v	 : z	1	: ,	-	. 1	1	2 8	08	PASS	, ,			PASS (T)
Cadmium Dromine S S N BD PASS - PASS (R) Cadmium Chloride S S N BD PASS - PASS (R) Cadmium Nitrate S I N BD PASS - PASS (R) Cadmium Nitrate S I N BD PASS - PASS (R) Cadmium Nitrate S I N BD PASS (R) PASS (R) Cadmium Oxide	GRA	Control of the contro										ć	40			•	11,330
Cadmium Fluoborate L - N 0.0 C.1 -0.3 -0.3 +2.9 SEB CADMIUM COMPOUNDS - PASS (R) Cadmium Nitrate S I N BD PASS - PASS (R) Cadmium Oxide PASS (R)	2 5	Cadmium Bromage Cadmium Chloride	n v	n v		· ·		1 1			1 (3 &	PASS	1 1	, ,	PASS (R)	PASS (T)
Cadmium Nitrate S S N B PASS R Cadmium Oxide S I N PASS (R)	CFB	Cadmium Fluoborate	, _) 1					+2.9	SEB		Odwoo M	MDS	,	,	PASS (R)	
Cadmium Oxide	S	Cadmium Nitrate	s	S			1	ı	,	,		8	PASS	1	•	PASS (R)	PASS (T)
	Š	Cadmium Oxide	S			' -				,	,	'		1			
	} } }			•			!	:					ı				
			_	_	-			_	_	_	_	_		_	_		

			WATER	-			SHATCH TEST	TEST						THE STATE OF	MULTIPLE		
		<u>_</u>		L	POLYCARB	8			اب		PENET	PENETRATION		<u>ត</u>	DECOM	TEST	-
		2		L				_	3	2			CTMC	_	TEST	PERFORMANCE	HANCE
-		<u>-</u>	0	~	HETCH	_	UF TOTT	_	ZO DE PHYS	3	=	EST	SINGLE			DATING	
2 38 55 5 65 5 65 5 65 5 65 5 65 5 65 5 65	CHEMICAL NAME	< ⊦		∵ ⊢	CHANGE		CHANGE	. ш	CHANGE	<u>-</u> w	8	ppm or mg"	TEST	5 -	CYCLES		<u>?</u>
		. ш			3		2	_	Ξ						١	POLYCAR-	Part Car
		·			3 hr 20 hr		3 hr 20 hr			Tear	3 hr	20 F		3 hr	2 hr	BONATE	2017
			T	1	-		-										
ž	Cadmin Sulfate	v								1	١	S	PACC	,		PASS(R)	PASS(T)
3		S			_	<u>'</u> -			,	•	,	3 '	3 ,	,	,	PASS(R)	PASS(R)
ક	Calcium Arsenite	S	-			-		_	,	1	٠		•	1	,	, <u>.</u>	
3		S	S	_			-	-	ı	1		•	,	1	•	PASS(R)	PASS(R)
3	Calcium Cyanide	s	•	_		-	DECOMPOSE	OSES		-	. :		· ;	1	•	[PASS (R)	
2	Calcium Fluoride	<u>ر</u>				! 			,	1	,	•	ı	1	,	PASS(R)	PASS(R)
55	Captan	· v	_			<u>.</u>		-		,	•			ı	1	PASS (R)	PASS (R)
8	(Carbolic 0il	_	_	_		-	SEE PHI	PHENOL		•	,	•		1	,	PASS(R)	FAIL (T)
88	Carbon Disulfide	_	1	_	FAILED	<u> </u>		_	,	•	•		,	•	,	FAIL(T)	FAIL(T)
CIC	Catechol	s	s	· 		1		_	,	1	8	욻	PASS	ı		PASS(R)	PASS(T)
S	Caustic Potash Solution	_	-	<u> </u>	<u> </u>	<u>'</u>	<u> </u>	-	,		5	S	. SYM		. ,	FAIL (R)	PASS(T)
css	Caustic Soda Solution	_	-	· ·		1				,	8 8	8	PASS	,	,	FAIL (R)	PASS(T)
3	Chlordane (liquid formulation)	_	-	_		SEE		CEROSEKE	- INCOMPATIBE	PATIB	•	'			,	PASS (R)	
3	Chlorine	5	-	_	<u> </u>	<u> </u>		1			22	8	(01500)	ORED)	٠	PASS(R)	FAIL (T)
CŢF	Chlorine Trifluoride	9	-				₩ ₩	YDROFIL	HYDROFLUORIC ACID	CID	١	•	,	1	۱.	[PASSJ(R)	FAIL (T)
65	4-Chloro-0-Toluídine	s	=	_	FALED	9.0+ 0		+3.3	+2.3	+5.7	ł		: •		,	FAIL (T)	FAIL (T)
종	Chloroacetophenone	s	_			•					ı	•	,	,	ı	PASS (R)	PASS (
3	Chloroacetyl Chloride		•	_	_		SEF H	YDROFIL	HYDROFLUORIC ACID	CIO CIO	1.	• ;	,	1	,	[FAIL](R)	FAIL
3 8	P-Chloroantline	· .								, ;	2 2	& 8	PASS	1	•	PASS (R)	PASS (T)
K 93	4-Chiorobutronitrile	_	-	<u> </u>	rA LED	0 +2.5		+5./		+3./	3	22	PASS	1	•	rait (1)	(=) ccw.
CR.	Chloroform		-		=				-	,	•	,	•	,	1	FAIL (T)	FAIL (T)
웅	Chlorohydrin (Crude)	_	-			<u>우</u> 근		<u>.</u>	+8.0	+4.8	8	8	PASS	,	'	PASS (T)	PASS (T)
38	Chloromethyl Methyl Ether		_	_	FALED		FAILED	_	,		ı	1	•	1	•	FAIL (T)	FAIL (T)
2 6	U-Cn loran I I rocenzene D-Ch loronhano	n u	- - 7		_		_			•	٠ ۾	۱ ۾	- DAG			(a) 5590	DACC (T)
5	r-culorophien.	2	_	- 1	<u> </u>	-	_		,		2	3	2	. :	'	LA33 (R)	() ccv
		·	-														
							-		_								
		-	-	_		- .	_	-	-	_	_				_	_	_

		F	WATER				SHATCH TEST	TEST						E	MULTIPLE		
		-		~	POLYCARB	<u>8</u>		BUTYL	7		PENET	PENETRATION	_	<u> </u>	<u> </u>	TEST	_
CHRIS	CHENICAL NAME	^ - ∀ +		ال که س	WEIGHT CHANGE	토병	WE ZGHT CHANGE		20 hr PHYS. PROPERTY CHANGE	HYS.	E edd	TEST ppm or mg*	SINGLE DECON TEST		1141	PERFORMANCE RATING	MG
		- ш	 ප ස්		*		3	_	E		3 hr	20 hr		₹	2 hr	POLYCAR-	BUTYL
		7	1	+	3 hr 20 hr		3 hr 20 hr	발	Tensile	Tear						DOM	
335	Cyclohexanol Cyclohexanone Peroxide Cyclohexynyltrichlorosilane		111	* * * >			- 38	- HYDROC	HYDROCHLORIC ACID	- AC10	88'	88'	PASS PASS		, , ,		PASS (T) PASS (T) FAIL (;)
*	Cyclohexylamine		,	*	•			,		• !	\$		•	. ;		FAIL (R)	FAIL (1)
P NTO	Decaldehyde Demeton (liquid formulation)		1 1	ZZ	1 1		1 1	1 1	1 1		SEE X		NCOMPAT (BLE)	BLE)	' ;	FAIL (R)	FAIL (T)
05.0 02.0 08.4	DI-(2-Ethylhexyl)Phosphoric Acid DI-(P-Chlorobenzoyl)Peroxide DI-N-Butylamine			zzz			5	UNAVALLABLE	ABLE		2.4	a ''	FAIL -	PASS	FAIL -	PASS (K) ND FAIL	
DZM	DI-N-Propylamine Diazinon	700	1	ZZ:	-0.1	9	19+	1,	1 1	١,			1 1		1.1	FAIL (T)	FAIL (T) PASS (R)
988	Blouty (phenoi 0-Dich lorobenzene Dich lorobutene	<u>٠ </u>	- 1 1	* * *	FOGGED		+	- 61+	5.6	· · -	936	-	111	1 1 1		FAIL (T)	FAIL (T)
# & &	Dichloroethyl Ether 2,4-Dichlorophenol 1 2-Dichlorophenol	- ۷۷ ـ	1 25 1	Z Z Z	FOGED		+3.8	+11.e	-2.5	Ţ.,			1 1 1	1 1 1		FAIL (T) PASS (R) FAIL (R)	FAIL (T) PASS (R) FAIL (T)
25	Dichlorus (liquid formulation) Dicyccopentadiene	ארי		zz	FOGED		+42		1 1			1 1	1 1	1 1	. • •	FAIL (T) PASS (R)	FAIL (PASS (
DEO DFA	Dieldrin Difluorophosphoric Acid(anhydrous) Difsprooylbenzene Hydroberoxide	ν <u>-</u> -	<u>- , , </u>	Z > Z	FOGGED		- S8E +24 +	- HYDROF +29	SRE HYDROFLUORIC ACID	ACID	۱ ۱ ۱	111	1 1 1	111	, , ,		PASS FAIL FAIL
ost ost	Dimethyl Sulfate Dimethyl Sulfide	<u> </u>	1.	z z	1 1					1 1	BD 27	68 '	FAIL	PASS	PASS	~~	

		Г	WATER	Н			SWATCH TEST	EST					3	T101 C		
		~		ш	POL YCARB	Ц		BUTYL			DCMCT0ATTOM			DECOM	-	,
CHRIS	CUPLIFOR MALE) }- - (70		WF 1GHT		FIGHT	20 hr PHYS.	PHYS.	T CMC	TEST	SINGLE		TEST	PERFORM	MANCE
CODE	בעבשוראר אאשנ	K +			CHANGE		CHANGE	PROPERTY	:RTY	E dd	ppm or mg*	DECOM	ວ	CYCLES	RAT	RATING
		- ш			æ		£	Criange (X)	2 -			TEST	-	8		
		\dashv	3	<u>m</u>	hr 20 hr		r 20 hr	3 hr 20 hr Tensile	Tear	3 hr	20 hr		3 hr	2 hr	POLYCAR- BONATE	BUTYL
ಕ್ಷ	Chloropicrin			<u> </u>		'		,	1	2	,	1	,	1	2ASS (8,	
858	Chromic Acetate		S	\$	FAILED +0.1 +0.2	FA]	ALED -	- FOR	FORMS ACID	MIXTURES	- S3 -		1 1	1 1	FAT. (T)	FA (L(T)
5	Lhromic Anhydride (wet)	S	<u> </u>	<u>'</u>	•	•	•	•	•	(BLISTERS)	TERS)	, ,			PASS (T)	PASS(T) FAIL(T)
888 888	Chromic Sulfate Chromous Chloride Chromyl Chloride	SS	NN :		0.0 +0.1	1-0-	+1.0	-3.7	-4.9 Arth		21	CETATE	! • •		FAIL (T) PASS (T)	PASS(T) PASS(T)
ĕ ₹	Cobalt Fluoride(ous) Copper Acetoarsenite(ic)	SS	~~	우 !		2 +0.2	+0.7		4. '	· & ·	' <u>8</u> '	PASS			PASS (T)	FAIL(T) PASS(T)
₹	Copper Arsenite(ic) Copper Cyanide Cooper Fluchmate(ic)	SO-	223	<u>. </u>	!	!						• •	j • •	: ' '	PASS (R)	PASS(R)
6 S	Copper Oxalate(ic) Copper Sulfate(ic)Ammoniated	<u> </u>		9 6	-0.0 -0.0 -0.1 -0.2	2 -1.2	-1.1	0.0	0. ÷	3 · E	LUOBORATES	TES -	·	,		PASS(T)
358	Couraphos Creosote, Coal Tar	<u> N - </u>	22		• •	' '	, ,	<u> </u>		' 8	3 1 8	2 : 1			_	PASS(1)
388	Crotonaldehyde Crimena Hudenosovida		**	• •		• •	1 1			9 60	8.9	1 · ·	TAIL .		FAIL (R)	FAIL(T-D) <3 1/4 hr (T
GES	Cupriethylenediamine Solution	-	Z 2	٠ د	, 6	SEE	CUMENE	, ; ,	, ,						PASS (R)	FAIL(T)
¥.5 C.₹	Cyanoacetic Acid Cyanogen	، ، ق تــ ر	Z Z			; '		0.6	5.0	. 8	· &	PASS	1 1	1 ,1	PASS (T) PASS (R)	FAIL(T)
e 5	Cyanogen Bromide Cyanogen Chloride	<u>8 8 9</u>	: z >	1 1				HYDROCHLORIC ACID	- ACID	e ' 2	<u> </u>	PASS	1 1	1 1		PASS(T) PASS(R)
		<u> </u>	<u> </u>	!	1	-	 				!	,	· · · · · · · · · · · · · · · · · · ·	ı	[PRSS](R)	FAIL(1)
_		_					_									

		F	WATER	_		S	SWATCH TEST	EST		_			1	MIN TIPLE		
		<u>ر</u>	ļ	Ш	POLYCARB			BUTYL		PENE	PENETRATION		<u> </u>	DECON	TEST	-
		21	_	L		_		300	2000	1	1001	213012		TEST	SUNTACE DEBENDANCE	KANCE
CHRIS	CHENICAL NAME	- «	0_	4 0	WEIGHT	30	WEIGHT	PROPERTY	ERTY.	- ed	ppm or mg*	DECOM		CYCLES	KATING	NG NG
200		⊢ () (*)	, 	(¥)	CYANGE	39.			TEST		8		
		n n	a d	<u> </u>			2 hal 20 hr	7.000	Topic	3 hr	20 hr		3 hr	2 hr	POLYCAR-	BUTYL
		1	+	1	11 CO 11		2		100							
40	0.00 mm s 4 mm s	_								•					(0)	
3 6	Dimetnyl Acetamide Dimethylhexane Dihydroperoxide Wet	- v	1 2	. ·	0 1 0 2		· ·	 	, ,	<u></u>	· 6	FATI	PASC	PASS	PASS(T)	PASS(T)
3	Dimethyldichlorosilane				-		SEE HYD	HYPROCHLORIC	AC	3 1	3 '		3	3	[FAIL]R)	_
풀	1,1-Dimethyhydrazine		1		· 			,	'	윮	8	FAIL	FAIL	ı	FAIL(R)	u.
DNZ	Dinitrobenzene	S		<u>'</u>	<u>'</u>	<u> </u>	<u> </u>	. !	' .	,	,	' ;	1		£	웃
SAB DARB	Dinitrobenzene. Meta	S	_						,		1	,	,		PASS(R)	PASS(R)
DNC	Dinitrocresols	S	=	_		'			1	١	,	,	•	•	2ASS(R)	PASS(R)
š	2,4-Dinitrophenol	S	I	_	<u>.</u>	,	1	1	,	'	1	,'	,	•	PASS(R)	PASS(R)
110	2,4-Dinitrotoluene	s	_	<u>'</u>	' 	<u> </u>		•	•	<u>'</u>	•	· —	1	,	PASS(R)	PASS(R)
DAM	Diphenylamine	s		<u>'</u>	· 		.'	1	•	,	1	'			PASS(R)	PASS(R)
Odo	Diohenyldichlorosilane	-	: -	<u> </u>	FOGGED	<u>; </u>	ATTACKED		'	1	٠,	. (ı	•	3 hr(T)	FA11 (T)
DPM	Diphenylmethane Diisocyanate	, ن	_	<u>-</u>	<u> </u>		DECO	_	_		·	,	,		[PASS](R)	
010	Diquat (liquid formulation)		,		+0.1 0.1	<u>-</u>			-4.0	8	8	FAIL	PASS	PASS	PASS(T)	<u> </u>
Sid	Disultton (liquid tormulation)				=	+3.9			-5.0		•	1	1		FAIL(T)	FAIL(T)
ora,	Uluron (water suspension)	:	Σ	¥	+0.1+ +1.4	4 - 6	+3.9	-5.7	-3.4	•	ı	•		,	PASS(T)	FAIL(T)
800	Dodecylbenzene		-		0.0 0.0	82	1	•	'		,	<u>'</u>	,	ı	PASS(R)	
010	Dodecyltrichlorosilane				-		띯	HYDROCHLOR IC	d ACID	,	١	,	,	•	[FAIL (R)	FAIL(T)
3	Dursban (liquid formulation)				FOGGED	+33	•		,	١.	,	,	•	•	FAIL(T)	FAIL(T)
2 5	Endosultane	S		<u>.</u>		·	'		•	1	,	,	1		2	2
ă	ENGLIN	<u></u>	<u> </u>	-	. :	1	'			, :		:	,	٠.	PASS(R)	PASS(R)
EPC	Epichlorohydrin		<u> </u>		<u>'</u>	'	•	'	•	8	_	FAIL	FAIL		PASS(R)	FAIL (T-D)
E10	Ethion		_		FOGGED	+20	1	•	•	,	,	•	,	1	FAIL (T)	FAIL(T)
EAC	Ethyl Acrylate	.	-	<u>'</u>		<u> </u>	1			11.3	ı	'	1	•	FAIL(K)	FAIL(T)
	Ethyl Chloroformate		-	_	· -	•		·		54	ı		'		FAIL (R)	FAIL (T)
<u> </u>	Ethyl Methacrylate			<u>_</u>		'	1		r	8 8	1		,		FAIL(R)	FAIL(T)
					<u> </u>					:	_					
			_					_								
		-	_	_	_		_		_	_		_	_		_	

9 4	TEST	PERFORMANCE	KA! ING	POLYCAR- BITTY	4		- 6	=		2	_	FAIL (R) PASST)	DACC (D) CATHT)	26	2	æ	æ.	FAIL](R) PASS(R)	CALL /TY CALLYTY	=	21		_	FAIL](R) FAILT)		PASS (K)	_	- (-		ISS (K) PASS(K)	CATITION CATITI		FAII (P) CATI'T)		(T) FAIL	-			
	_		· -	S	8			<u> </u>		Ξ.	Ξ.	<u>~</u>	_		2	<u>-</u>	Ξ,	<u>E</u>	-		7. 1	Z ,	<u>.</u>	E		Ž (Ξ.	_	? ?	₹ —		- 4	L			:			_
MULTIPLE	DECON	TEST	CYCLES		2 III.	L			,	ı 	١	,		ı 	'	'	,			· _	'	•	1	1		'	1		'	'	_	'. '	1 1	1	1 1				
₹	_		3		3 111				1	1	•	1			,	,	١			I			•	1		1	١	1	ı 	1			. (1					
		SINGLE	TEST					<u> </u>	-		1	PASS		1		1	,		:	٠	1	1	1	,		PASS	ı												
	PENETRATION:	TEST	ppm or mg*	3	20 nr				1	1	1	8			1	1	•		!	1	!	1	,	•	:	3	1	,				1			. 1				
	PENET	Ē	and d		3 nr			1	, ,	/7	1	68	473	2 5 5	771	3	0.3				4:4	3	1	•	6	3	,				1		ACID		; '			_	
		HYS.	RTY GE	_	Tear	- —					ACID	1	'				,		,			_		ACTD		, 22	017		?	,	ACID	9	21 5110		-4.5			_	
EST	3	20 hr PHYS.	PROPERTY	<u> </u>	Tensile				:		HYDROCHLOKIC			-	1			MATER		_			HYDROCHLOKIC	HYDAOCHLURIC	-	COL UNDOUGH DO TO	T WOTH TO	, ,	6.7	. ,	F HYDROCHI DRIC	-8.5	HORICA		-15.0				
SWATCH TEST	ă	ueteur	CHANGE	<u> </u>	hr 20 hr			·			SHE HYDE		1 1			,	, !	MITH						SEE HYDA		de uvn		, 4		, .	E HYDR		YORDE		+6.4				
35	1	š 	i さ 		3	<u> </u>	170	136	7	. '	^	1	į ·		_	1	- 1	ACT 10	+2.	?	_	, '	_	<u>~</u> -	<u>i</u>	. "	٦ 	, 64	-	ı	V	0.0	7.5	<u>;</u>	+3.5		*	_	
OQV ACVO	3	TUSTON	CHANGE	3	hr 20 hr	_	d'str		725	,	'	•	· '				- 1	MILD REACTION	FORGED	}		? 	•	1		1	١	, 5	-	1		7	_	ا	2			_	
100	2	5	# C		3 4		ŭ	26	2		•					t	1	.	8	•		⊃. ⊃	٠	•	ì	•		, c		, ;	•	+0		,	+0.1	: :			
WATER	۲.	⊿ ∀	ں⊢،	>	ш		2	= 2	= =	= :	-	z	-	; 2	E :	2 :	z	>	2	: 2	: ;	E :	_	-	, a	= >	- 3	= 3	2	5	>	2	>	2	z	,		_	
\$		л c				<u> </u>	_	<u>.</u>	<u>.</u>	<u>.</u>	•	<u>'</u>	<u>'</u>			1	5	<u>'</u>	<u> </u>				1	•	1 4	70		_	٠.	<u> </u>	ا	1	'	1	1	,			
	_		≪ ⊢				i Anhudaous a	e Annyarous														•					_									-	_		
			CHEMICAL NAME				pingthioid Digital	Ethyl Phosphorothioic Dichioride Annyarous	Anias photodica for laste	; . <u> </u>	Ethyldichlorosilane	Ethylene Chlorohydrin	Ethylene Cyanobydrin	Dibromi do	o intromine	Etnylene Ulchloride	0xide	mine	Jahvde	2 February Armylato	A) Acidiace	Crny I Idenenor Dornene	ctny ipneny i Dichlorosi iane	Ethyltrichlorosilane	County Assessed to October	nominal Oxalate	101 100 101 100	uol tue	iuobora te	Ya i a re		ic Acid	nic Acid	Formaldehyde Solution	Glutraldehyde Solution				
	_						Seeker) Dhos	FINAL PRO	Ethyl Pho	Ethy! Amine	Ethyldich	Ethylene (Fthylene (Cthulone Dibnomide	iculy lene	trny lene	Ethylene Oxide	Ethyleneimine	Fthy Theyal dehyde	2 F + hv] ho.	C - L City + 11C.	ctny i gen	trny phen	Ethyltric	Econolic Ame	Ferric Ammonium	forms fluoride	Formour 67 mobourto	Formous Calato	rerrous o.	Fluorina	Fluosilicic Acid	Fluosulfonic Acid	Formal deby	Glutralde				
		Q.	C. S.				000	2 6	5	5	S	픕	E	900	2 6	3	ĕ	Eil	FHA	EAT		בים בים בים	3	ETS	CAS	2 5	ניני	۷ <u>و</u>	2 6	5	FXX	FS	FSA	FW	GTA				

		-	WATER	\vdash	POL YCARB	3	SMATCH TEST	EST BUTYL		DENETOATION	41100		₹°	MULTIPLE	1831	.
		v -	N C		10.101.01	1 3	חכוסת	20 hr PHYS.	۲S.	TEST	F 15	SINGLE		TEST	PERFORMANCE	W NCE
CERIS COE	CHEMICAL NAME	∢ ⊢			CHANGE	# 5 T	CHANGE	PROPERTY CHANGE		ppm or mg*	* 600 r	DECOM		CYCLES	<u> </u>	2
			->		3			- ⊦		3 hr	20 hr		3 hr	2 hr	POLYCAR-	BUTYI,
		+	4	3	3 hr 20 hr		3 hr 20 hr	lensile	lear	+					1	
3	0.00 Example 10.00 Example 10.	<u>·</u>		<u>'</u>	<u>.</u>		ı	1		8	83	FAIL	PASS	PASS	PASS(R)	PASS(T)
5 E	Heptachlor Heptachlor			¥ .	- 1 +0.2	2 +0.6	4,4	0.0	+6.1	· &	- 8	PASS	1 1	1 1	PASS(T)	PASS(T)
₹ ₹\$	nexatecyltimethyl mimorium circolor N-Hexanethylenediamine	N L	- N	22	<u> </u>	+50			1 1	- 08	- 08	PASS	1 1	1 1	FAIL(R)	PASS(T)
1	Hovemothy leneinine		+	<u> </u>	-16	FA	EB				•	•		1	FAIL(T)	FAIL(T)
Ī	Hexamethy lenetetramine	· · ·				• •	•	1 1	, ,	6.0	· &	PASS		١ ١	PASS(R)	PASS(T)
을 로	riydrazine Hydrochloric Acid	۔ ب	S	===	0.0	0.0 +0.9	= -	-74	-67	. ' '	1 1		1 1	1 1	PASS(1)	FAIL(T)
HEA	Hydrofluoric Acid	<u>.</u>		-	Ī	1				,		-	:		70/00/0	1
£	Hydrogen Bromide	9	s	>-	FDRMS ACTIVE	ACTIV		,	1	1	1	•	1		PASKE	FAIL(I)
윷	Hydrogen Chloride	<u>ن</u> د	· ·	>>	FDRMS	ACTIVE ACTIVE	E ACID	1 1	, ,			1 1			PASS(R)	E STEEL
Z Y	Hydrogen Cyanide Hydrogen Flinnide	s		- z	FAILED	·		,	1	1	,	'	1	1	FAILT	PASSO
H D				· E	<u> </u>		1	ı	1	'	'	'	. :	1	(where	(N)
S S	Hudmoon Sulfide	9	\ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	=	'	<u>'</u>		,	,	98	38	FAIL	FAIL	'	PESS(R)	FAIUT-D)
¥		ب			0.0			-12.0	-2.5	28	3 8	PASS	1 1	· ·	PASS(T)	PASS T
HAS		v -	S	2 2	0.0	1 +0.2	40.3) O O	0.0.	2 2	88	PASS	ı	'	PASS(T)	PASS(T)
12	Hydroxypropy Acrylate Isobutylamine	4						,	,	480	,	1	ı	1	FAILIN	ראות ו
IBN	Isohutropitrile	1	1-	=	FOGGED	+3.5	5 +5.0	-3.8	+1.2	0.4	!	· ·	•	,	FAIU(T)	FA11(T)
2			1	a : :			7 2 2 2 4	- 2	ı	3.7	(()	+ 1	FAIL(R)	FAIL(T)
<u>a</u>					- 1	7 5	1 1 N	I SUOCII ALDEHTUE		•					PASST	FAILT
IPE	Isodecylate, innibited Isopropylether			: : z		-		١	÷	3.3	1	· ;	*	! ! 	PASSIN	· · · · · · · · · · · · · · · · · · ·
		<u>-</u>	-				(ì	! ,		
	-	_	_	_			-	_		-		_	_	_	-	_

L				WATER	-			SHAT	SWATCH TEST	2						MULTIPLE		
			٧٢	w c	×w4	POLYCARB	82 5	ן נ		BUTYL 20 iir PHYS.	HYS.	PENE	PENETRATION TEST	SINGLE		DECON	TEST PERFORMANCE	T HANCE
		CHEMICAL NAME	≪ ⊦	٠ ـ ١		CHANGE		3	CHANGE	PROPERTY	RTY	E A	ppm or mg*	DECCA		CYCLES	RATING	9
			- W	_ =	:	*	_	ت	3		d			<u>.</u>		80	200	
	 -		1	d	- LI	hr 2	hr 20 hr	3 hr	3 hr 20 hr	Tensile	Tear	- 3 hr	20 hr		3 hr	2 hr	BONATE	BUTYL
<u> </u>			-	<u> </u>								1000				,	DACC (D)	(T)
	001	Isopropy: Mercaptan		1 1				, ,				80	8	FAIL	PASS	PASS	FAIL (R)	PASS (T)
	IV		<u></u>	,		,	,	,	1	,	1	0.8	'	'			FAIL (R)	FAIL (T)
13	. P R	Kepone Lauroy} Peroxide	s s	X 1-1	zz	, ,	1 1	, ,	1 1	1 1	١ ،	1 1	1 1	٠,			ND PASS (K)	ND PASS (R)
-	2	-4-	-	+	+,	+	+-					ATTAPKEN	KED	'			PBSS (R)	EA11 (T)
	E 2	Laury rereaptain	س ب				1 1					-	-			· ·	PASS (R)	PASS(R)
	3 8		٠ ـ	2 1	. >	-0.1	_	-0.1	+1.2	0.0	+	,	'		١	'		PASS(T)
	LAR.		S	-	=	,			•	•	•	,	,	,	•	•	PASS (R)	PASS (R)
	LFR		S	_		1		,	,	•	1	,	1	1	•	1	PASS (R)	PASS (R)
L	1.SA	Lead Stearate	S	-	=	-		١.	'		'	: 1		•	: ,	;	9	9
	E		S	<u> </u>	· >-			5	JRMS A	FORMS ACETIC ACTO	•	ı 	'	'	<u>'</u>	•	[PASS](R)	[PASS](R)
	LBC		y v	S	Z		4.0	9.0		÷ ;	÷.0		B 6	FAIL	1	1 1	PASS (T)	FAIL (T-D)
	Z Z	Lithium thromate Malathion	^ <u>~</u>	0 1	5 Z	<u>-</u> - '	7.2		7.7	2. '	, ,	9	88	FAIL	FAIL	, ,	PASS (R)	FAIL (T-0)
1_	3	Eslair Arid	1	-	12	-	 -	 	,	'	:	. 6	8	PASS	•	'	DASS (R)	PASS (T)
	MLA		, v	, v		,	,	,	,	1	1	8	8	PASS	,	,	[PASS](R)	PASS (T)
	H.		s	X	Z	,	1		•	1	•	,	٠,	• ;	<u> </u>	,	2	2
	¥ ¥	Mercuric Acetate	y v	v 2		. 1	•	•	•	1	1 (1 1	æ '	PASS			PASS (K)	PASS (1)
1	1		, ;	= ;			- 	 :		,			•	:			200	(1)
_	E		S	S	z		•	당-	E MER		.TS	•		1	1	1		PASS (R)
	Z.		S 4	v -	~ 2				SEE MERC	MERCURIC SAL	SALTS		1	'	'		PASSK	PASSK
_	2 1	Mercuric Logide	n 0	- (1				, ,	1 1			, ,	8	PASS			PASS (R)	PASS (T)
	Š		S	-		-	•	<u> </u>	1	,		·	•	•	'	1	PASS (R)	PASS (R)
لـ	_ :		_	-	_	-	<u>-</u> -	_	-				:	<u>.</u>	,	_		

L				MATER	~		YAS	CUATCH TECT	CT.		_					-	
	-		_		┼	POLYCARB	-	6	BUTYL		-	700		를 -	MULTIPLE		
	CHRIS		o ⊢ ·	v 0	L	WEIGHT	<u> </u>	VETGAT	20 hr PHYS	HYS.	T.	TEST	SINGLE		TEST	PERFORMANCE	ANCE
	COOE	CHEMICAL NAME	< ⊢	75		CHANGE (X)		CHANGE (4)	PROPERTY		udd d	ppm or mg*	DECON	5	CYCLES	PATING	¥
			ш	8	->	3		3	3					- ;		POLYCAR-	1000
				\dashv		3 hr 20 hr		3 hr 20 hr	Tensile	Tear	3 hr	20 hr		Ė	Z hr	BONATE	TL ING
	-															,	
	RS	Mercuric Sulfate	S		<u> </u>	+ -	1 +0.5	+2.8	£.5	+7.0	1	80	PASS	'	1	PASS(T)	PASS (T)
	2	Mercuric Sulfide	л v			1 —		1	,	1	٠	•	•		,	PASS(K)	
	¥ 3	Mercurous Chloride	n v		= Z					, ,	, 1		· .	1 1	٠,	DASS(R)	PASS (B)
	N.	Mercurous Nitrate	S		. <u> </u>	<u>'</u>		•	1		•	80	PASS			PASS(R)	
<u> </u>	£	Mercury	<u>'</u>	+-	-	<u> </u>	•	'		,		•	'	! •	, '	PASS(R)	PASS (R)
	MS0	Mesity] Oxide							 '	•	0.5	,	•	,	,	PASS(R)	
	MAA	Methyl Amyl Alcohol	_	1		· ·	-	1	1	•	8	80	FAIL	FAIL	ı	PASS(R)	•
	M18	Methyl Bromide	9	X .		FOGGED		'	-	١	,	•	1	<u>'</u>	1	FAIL(T)	_
	HCH.	Methyl Chloroformate	_	1		FOGGED	=	+13	-12.0	-13.0	1		,	1	,	FA11(T)	FAIL (T)
	MCS	Methyl Dichlorosilane	-	,		'	7	FF HYD	SEE HYDROCHLORIC	ACID	. '	, 1		1	, ,	(FAIL TR)	FAIL (T)
	717	Methyl Hydrazine		,,	z	FAILED	+3.4	+6.7	-4.0		8	80	FAIL	'	,	FAII (T)	
	Z A	Methyl Isobutyl Carbinol		1	Z 2	1	'	, !		,	8	8	FAIL	FAIL	1	PASS(R)	FA1L(1-0)
	<u> </u>	Methylsopropeny: Retone (Indibited) Methyl Parathion	اب د	-			· ·	5 ,	UNAVAILABLE		1 1		' '	' '	, ,	PASS(R)	PASS (P)
			+	- -	+	-						 		-+	į		
	¥	Methyl Vinyl Ketone	<u>.</u>	1	-	FOGGED	+8.5	+8.6	STRONG OD	400R - 17	AILED	,	•	,	,	FAIL(T)	FAIL (T)
	E A	Methylamine (Water solution)		1 0		1 1	_		1	,	 	۱ 6	1242	DACC	, , o V O	CAT (O)	FA1L (T)
	Ā	N-Methylaniline	ب. د	7 1			. ,	. ,	٠,		3 6	3 2	FAIL	FAI	. A35	PASS(R)	FAIU 1-D)
	ECT.	Methylcyclopentadienylmanganese Tri-	_	_			1	5	UNAVAILABLE	,	3 1	\ (,	,	1	9	Ş
	1	carbonyi	-	+	+	!					- ∤		1				•
	0.0	Methylphosponothioic Dichloride Anhydrous	7	-		FAILED	+32	•	,	١	. '	,	1	'		FAIL(T)	FAIL (T)
	A S	1-Methylpyrrolidone		1		FOGGED	45.8	+5.8 +15		, ;	,	1	,	,		FAIL(T)	FAIL (T)
	\ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	Methyltrichlorosilane		1 -	 > 2		···	EE HYD	SEE HYDROCHLORIC	ACID	•	ı	•	1	1	[FAIL IR)	FAIL (T)
		Motor Fuel AntiKnock CMPDS (PB Alkyls)	<u>- ر</u>			SEE TETRA	ETRAETHYL S	TETRO	TETRAMETHYL LEAD	, o	١,				1 1	FAIL(T)	FAIL (T)
			+	-	4		;	-									•
											i						

		<u>-</u>	٩	Ļ	DOLVERDO	ŀ		ة	PILEVI					Ē			
_		S	s		2	2		3	1 6	,	PENET	PENETRATION TECH			DECOM	TEST	ST
	CHEMICAL NAME	- < +	0 _	≪ U⊦	WE I GHT CHANGE	토병	WE I GHT CHANGE		PROPERTY	E L	e do	ppm or mg*	DECOM		CYCLES	RATING	PERFURMANCE RATING
				_	3		3	_	15 (X)	 			3	-	8		
			×i		1 20 1		1	2 5 5 5			3 hr	20 hr		3 hr	2 hr	POLYCAR-	BUTYL
		+	+	1	3 nr 2				lens) le	lear					+	EUMAIE	
	Nakam (Bionish formulation)	_					1		,		80	S C	PASS			DASS(R)	(I)SSFG
	Mabail (Tiquia ioribia)	-		: 2							}	3	3		·		
	Maleu Josephi Con Ton	· -	٠.	_			100	2001100		1	,	,	,	_	_	(T)	FA11 (T
	Naphicha, Coal lar	-		. z	~		7 2	FATER	, ,		,		· •	· ·	· •	FA1: (T)	
	Naphthylamine	S	Σ	. z	-		_	-	,	1	1	,	·	,	,	PASS(R)	
		<u>:</u>	1	+	+	1	,	j k	i !					1	<u> </u>	; - -	
	Nickel (vanide	~					······	,		1	•	1			·	PASS(R)	PASS(R)
	Nickel Fluoborate			=	0.0	-0.4	-0+	+0.9	-7.0	-4.0	1	80	FASS		'	PASS(T)	
		S	- S	=							,	8	PASS	,	1	PASS(R)	
	Nickel Hydroxide	S		z		,		,	,	,	1	•	,	'	,	9	웊
	Nickal Nitrata	-	T-	-	-	+-	 	-				. 2	PASS			PASS (R)	PASS(T)
	Nickel Micrate			. *				-	•		_	8	PASS		_	PASS(R)	
	Nicotine	, <u> </u>	. ,	 : z		,		,		,	0.7	۱ ۱	 -	,	'	FASS(R)	
	Nicotine Sulfate	S	\$	×		<u> </u>		1	,	,	1	'	· 		•	PASS(R)	
	Nitralin	n	_ I	z	<u> </u>		,	,	,	1	,	١	1	1	'	<u> </u>	9
	アルション・アン・アン・アン・アン・アン・アン・アン・アン・アン・アン・アン・アン・アン	-	•	; 2	- 70 - 10 - 10 - 10 - 10 - 10 - 10 - 10 - 1	<u> </u>	FA I B					٠,	. —	1	(FAII (T)	
	Nitric Octob	ے د		: >	,	<u> </u>	Ý	14	STATE ACT		,	,	,		,	[PASS R)	FAIL (T)
	Mitrilotriacetic Acid and Salts	S				-	<u>-</u>			1	1	١	, 		•	2	
	2-Nitroaniline	S	_	. <u>.</u>	,	٠,			•	,	,	,	_	'	, 	PASS(R)	PASS(R)
	4-Nitroaniline	S	_	z	,	,	,	,	1	ı	1	١	,	'	,	2	₽
	Nitrobenzene	+-	-	2	FAILED	1	+4.2 +15	+	-10.2	+6.5	,	 	'	<u> </u>	<u>.</u>	FAIL(T)	FAIL(T)
	Nitrog≥n Tetroxide	9	٠,	>		-	S	EE NT	SEE NITIRIC ACI	_	,	,	,		,	<u>.</u>	_
	Hitrosyl Chloride	9		_	. —	-	SÆE	HYDRO	SEE HYDROCKILORIC	AC:0	,	,				[PASS](R)	
	2-Nitropropane	_;	_	×	,		,		•	•	8	8	FAIL	FAIL	,	PASS(R)	_
	0-Nitrotoluene		_	z	FAILED	_	6	+15.3	•	1	,	1	·	<u> </u>	'	- FAIL(T)	FAIL(T)

T MANCE NG	BUTYL	PASS(T) PASS(T) FAIL(R) FAIL(T) ND	PASS(T)	PASS(R) FAIL(T) PASS(R)	FAIL(T) FAIL(T) FAIL(T) FAIL(T) PASS(T)	PASS(T) FAIL(T) FAIL(T) FAIL(T-0) FAIL(T)	FAIL(T) PASS(R) PASS(R) PASS(R) FAIL(T-0)
TEST PERFORMANCE RATING	POLYCAR- Bonate	PASS(T) FAIL(R) ND FAIL(T) PASS(R)	PASS(R)	PASS(R) PASS(R) PASS(R)	FAIL(R) PASS(R) FAIL(R) [FAIL(R) FAIL(R)	FAIL(T) PASS(R) PASS(R) [FASS(R) FAIL(T)	PASS(R) PASS(R) PASS(R) PASS(R) PASS(R)
MULTIPLE DECON TEST CYCLES	2 hr		·	1 1 1 1	1111	PASS	
₹a- 5	3 hr		1		1 + + 1 +	PASS	
SINGLE DECON TEST		PASS PASS -	PASS		PASS	FAIL	FAIL
ATION ST r mg*	20 hr	80 - -	08	1 1 1	ACKED - BD	a ' ' ' '	UNAVAILABLE UNAVAILABLE - BD
PENETRATION TEST ppm or mg*	3 hr		8	3.6	9 ATTA >0.2	8	UNAN
HYS.	Tear	-4.0	,	1 1 1	- ACIO	C ACTO C ACTO C ACTO LATOE	AC10
1 2 8 3 3 4 4 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5	3 hr 20 hr 3 hr 20 hr Tens 11e	F2.0 0.0	,	, , ,	HYDROCHLORIC	+4.4 -8.1 -8 HYDROCHLORIC ACII HYDROCHLORIC ACII HYDROGEN SULFIDE LED	нтовесн.ов.т.
 	(2) . r 20 hr			1 1 1		+4.4 HYDRO HYDRO HYDRO LED	HYDROC
SHA CH.) 3 hr	+0.2 +0.2	1	1 1 1	- See	+2.1 SE SE SE FA	SE.
POLYCARB WEIGHT CHANGE	(3)	+0.2	,	1 1 1	1111	ED .	1111
POLY	3 hr	+0.1 _ FAIL	'	F 1 1	1111	F0GGED	1111
LOS RATER	~>W	7. Z Z > Z	Z	222	ZZZ>Z	***	>zzz>
	- eci	1115	<u> </u>	X 1 1	11010	10101	11000
N L K L	ш	8		ν —	11818	16101	77888
CHEMICAL NAME		P-Nitrotoluene Nonyl Phenol Oil, Miscellaneous, Croton Oleum	Parathion	Pentachlorophenol Peracetic Acid Perchloric Acid	Perchloromethyl Mercaptan Petrolatum Phenol Phenyldichloroarsine, Liquid Phenylhydrazine Hydrochloride	Phosdrin Phospene Phosphorous Oxychloride Phosphorous Pentasulfide Phosphorous Tribromide	Phosphorous Trichloride Polychlorinated Biphenyls Potassium Binoxalate Potassium Chromate Potassium Cyanide
CHR1S CODE		O O O O O O O O O O O O O O O O O O O	PT6	₽ ₹5	\$ E E S E	044 049 049 049 049	PPT 928 988 989 975 975 975 975 975 975 975 975 975 97

	THANCE	¥		BUTYL	PASS (T.) PASS (T.) PASS (T.) PASS (T.) PASS (T.)	PASS (R) FAIL (T) FAIL (T) FAIL(T) PASS (T)	FAIL (T) PASS (T) PASS (T)	PASS (T) [PASS)(R) PASS(R) PASS (T) PASS (T) PASS (R)	PASS (T) PASS (T) PASS (T) PASS (T) PASS (T) PASS (T)
	PERFOCMANCE	RATING		POLYCAR- BONATE	PASS (R) FAIL (R) PASS (R) PASS (R) [FAIL](R)	PASS (R) PASS (R) FAIL (T) FAIL (R) PASS (R)	[PASSIR) PASS (R) PASS (R)	PASS (R) PASS (R) PASS (R) PASS (R) PASS (R)	PASS (R) PASS (R) FAIL (R) PASS (R) FAIL (R)
MULTIPLE	TEST	CYC!.ES	8	2 hr		: t			
₹ _	•	3	-	3 hr	11111	Fil			
	SINGLE	DECOM	TEST		PASS PASS PASS PASS	- FAIL PASS	PASS	PASS - PASS	PASS PASS PASS PASS
701140	PENEJIKALION TEST	ppm or mg*		20 hr	88881	- 1 08	- B8	08''08'	88'88
l la	7. F.			3 hr	1 1 1 1 1	''' ' & &			
	HYS.	RTY	- 25	Tear	- - - ROX I DE	CAPTAN - -	ACID		105
EST Butyl	20 hr I	PROPERTY	CHANGE (%)	Tensile		ISOPROPYL MERCAPTAN	HYDROCHLORIC	11171	SODJJM HYDROX
SWATCH TEST BUT	1	CHANGE	G	20 hr	- - - POTAS	150PR +17	HYDR		
SWAT	100	ž	æ	3 hr	.	SEE +14 -	SEE	1111	. , ₩ , ,
POLYCARB	10013	CHANGE	(X)	3 hr 20 hr 3 hr 20 hr		FAILED			1111
<u>g</u>				3 4	1111		1 1 1		1 1 1 1 1
 	w e		:	<u>- w</u>	Z X Z Z >	ZZZZZ	> Z Z	Z>ZZZ	ZZ-ZZ
┝┷╾	2 H				1 8888		100	01000	20010
		CHEMICAL NAME			Potassium Dichromate Potassium Mydroxide S Potassium Oxalate Potassium Permanganate S Potassium Peroxide	Propiolactone, Beta N-Propyl Mercaptain N-Propylamine Propyleneimine (inhibited)	Silicone Tetrachloride Silver Fluoride Sodium Arsenate	Sodium Arsenite Sodium Azide Sodium Bifluoride Sodium Chromate Sodium Cyanide	Sodium Dichromate Sodium Fluoride Sodium Hydride Sodium Hydroxulfide Solution Sodium Hydroxide
	CHBIC	2005			PT0 PTH PT5 PTP P00	PLT PMR PRA PII PGA	STC SVF SOA	SAR SAZ SBF SCH SCN	\$ \$ \$ \$ \$ \$

	3	F	VATER	_		SHA	SWATCH TEST	15						MIII TIPI E		
			F	₽	POLYCARE	Ц	ಹ	BUTYL		DENCT	DEMETDATION		3	DECON	TEST	.
		7 -	٠, در				!	20 hr PHYS	HYS.	31	TEST	SINGLE	Ţ	TEST	PERFORMANCE	WANCE
CHRIS	CHEMICAL NAME	< t		(U)	ME IGHI	#3	CHANGE	PROPERTY	RTY	8	pos or sig*	DECON	3	CYCLES	RATING	9
		- 1			Œ	_	E		¥			3	-	8		
		u	<u>ح</u>	_[3 hr 20 hr		3 hr 20 hr	Tens	Tear	3 hr	20 hr		3 iir	2 hr	POLYCAR- BOMATE	BUTYL
		\perp	+	-												
8							SECONDICES	טפעפ				,			[PASS Tox	(OKSSAD)
ţŝ	Sodium freiny late	_				. ,	2	}		· ·	. 08	PASS	1 1		PASS(R)	PASS(T)
SSE		. v	S		+0.2 20.3	3 +0.2	+1.2	-3.0	16.0	,	8	PASS	•	•	PASS(T)	PASS(T)
SFS				- -	<u>'</u>	1	1	' ;	• ;		2	PASS		,	PASS(R)	PASS(T)
SS	Sodium Sulfide				<u>.</u>	43.9	45.5	-12.4	-7.8	'	• 1	•	•	• •	PASS(K)	FAIL(1)
AT.	Stannous Fluoride		~	7	10 1	2 0 3	4	+2 0	+7.3	_	8	PASS	,	•	PASS(T)	PASS(T)
Š	Structium Chromate	· ~		_		-			? .	,	'		_	•	£	9
STR	Strychnine		_	_	-	•		ı	1	,	•		•	1	욮	윤
SXX	Salfur (1:quid) @ 113-120°C	_		÷		FIRE SULT	111	1	'	, ;	•	•	1		윤	2
SFO	Sulfur Dioxide		· •	ا —	•	'	,	,	1	5.5	,	1	,	•	[PASSIR)	FAIL(T)
95	Sulfur Monochloride			Ļ		SE	HYDRO	HYDROCHLORIC	CIO	,	,	,		,	[FAIL](R)	FAIL(T)
SFA	Sulfuric Acid	_	S	_	FALED	F	FAILED			•	,	•	•	,	FAIL(T)	FAIL(T)
3	Sulfuric Acid (spent)	<u> </u>	s	· =						ATTACKED	63		1	,	PASS(R)	FAIL(T)
렸	Sulfury] Chloride			' 		SE		HYDROCHLOR1C	200	ı	1	1	•	ı	FAIL (K)	FAIL(1)
1ES	2,4,5-1 Esters (liquid formulation)	_	-	<u>'</u>	<u>'</u>	N KE	KEROSENE	INCOMPATIBLE	TIBLE	•				•	PASS(K)	LAIL(1)
TEC	Tetrachloroethane			- -	-		1	•		7.8	,	ť		•	FAIL(R)	
TED	Tetraethyl Dithiopyrophosphate	_	1	<u> </u>	<u>'</u>	75	DEC	SLOW DECOMPOSITION	ı 37	,	•	1	ì		[PASS](R)	≅,
펄	Tetraethyl Lead		_	Z	FALED	•	•	1.	1		,		1	•	FAIL(T)	FAIL(R)
LEP LEP	Tetraethyl Pyrcphosphate	~	_		- 1	' '			• ;	• ;		١,		1	PASS(K)	7.05(K)
<u>a</u>	letraethyleneDentamine		1		106350	ç. +	÷.	7.4.2	0.9		8	S S		1	FAIL(1)	PASS(1)
T.	Tetrahydrofuran	-	-		<u>'</u>	'	,			3700	,		• ! ,		FAIL(R)	_
ĭ	Tetramethyl Lead				FOGGED		,		,	•	•		1	•	FAIL(T)	
1 PG	Thiophosgene	_	_	<u> </u>	<u>.</u>	SEC	HYDR	CHLORIC	CID	,			1	,	[PASS](R)	FAIL
E	Titanium Tetrachloride	_		<u></u>	<u>.</u>		HYDR	HYDROCHEORIC !	AC TO	,	ı	1		:		FAIL(I)
ë E	Toluene 2,4-Dilsocyanate		1	<u>'</u>	'			MPOSITION M	「 歪	•	•	•	1		LPASS J. K.)	rail(1)
		\vdash		_	 					ĺ	<u> </u>	1		1		
		-				_	_			_						

		MANCE	9		BUTYL		FAIL(T)	PASS(R)	PASS(T)	FAILLT	PASS(R)	Ş	FAIU(T)	FAIL(T)	PASS(T)	FAIU(T)	(1)	rAIU!)	2	2 9	€ 5	S.	FAIUT	2	윷	FAIL(T)	FAIŲT)	FA11(T)	¥		PASS(R)	PASS(T)			
	TEST	PERFORMANCE	RATING		POLYCAR- BONATE		PASS (T)	~	PASS (T)		PASS (R)	Q	_	PASS (R)	_	_	(=/, 0000	PASS (1)	2 9	2 9	2 9	2	[FAIL YR)	PASS (R)			FAIL (R)	(FAII Ye)	PASS (R)	9	PASS (R)	PASS (T)			_
MULTIPLE	DECOM	TEST	CYCLES	80	2 hr		•	,	•	1	•		•	•	•	•		•	į	•	•		!	1	1	ı	1	: '	•		ŧ		i		
₹2	<u> </u>		5	~	3 hr		-	•	•	•	•	1	1	•	•	1		•	1	•	•			•	,	_	1	1 1	FATI	'	•	١			_
		SINGLE	DECON	3			•	•	PASS	,		•	•	,	PASS	•	:	ı	,	•	,	•		ı	,	,			EATI		,	PASS	1		
	PENETRATION	TEST	ppm or mg*		20 hr		,	,	8	ı	,		,	ı	80	'	j	,	,		,		-	,	ı	,	,		2	<u>'</u>	•	88			_
	PENET	I	u wdd		3 hr		'	•	8	,	,	'	0.7	3.6	80	9.0			,	,	ı	,	-	•	,	8	×100		2	} ,	ı	,			_
		HYS.	RTY	2	Tear		-12		-6.8	,	,		•	•			;	1	•	'	•	•	1 P	<u>'</u>	ABLE	0.0		97.19	ar a		•	-3.7			_
5	BUITE	20 hr PHYS	PROPERTY		(a)		-32.2	•	-6.7	•	1		ı	•				•	AVAILABLE	1	•		MANDOCKI OS.		IMAVA	+0,3		OLOG INTOGRAN		ı (,	-3.7			
SHATCH TEST	3	100	CHAGE		14.16		75.55			=	'		,	ı	۰.			Ξ,	NOT AV	,		•	<u>i</u>		•	-6.1	,	1.				+1.1	!		
SEA		_	苦苦		-		0.0 +4.7		Ŧ	FAI	1		,	1	ı			FA		,	•	,	30		ı	0.0	1	15	2	. 1	ı	+0.2			_
	ARB	į	CHANGE	E	3 hr 20 hr		0.0	,	0.0	Œ	•		,	,	•	ı		0.0	r	,	ı			ı	•	+0+			,	. (÷			
	POLYCARB		3	_	4		0.0		0.0	¥.	,		,	_,			-	0.0	1	,	,	,		,	ı	9		T	1			÷			
3	~ '	NA	٠,		> w		z	*	z	z	×	-	*	. z	: z	Z	i	>	z	-	Z	*	>	Z	2	×	Z	7	- 2	2	2	Z		 	
WATER	,	Λ C	ب د	=	8		s	-	· s	•	X.	-		ı		S		ī	S	-	S	S	1	v	, ,	1	1				, -	· v			_
	v	າ ⊢	« 1	_ `	4	L	S	S	ب ،	_	S			- 1	-	1 (2	1	_	S	S	S	S) (G	0	ت	1		<i>y</i>		S	,	 	_
			CHEMICAL NAME				P-Toluene Sulfonic Acid	Tokanhana	Trichlorfen (liquid formulation)	Trichloroethylene	Trichlorophenol	2 & S_Trichlorophenoxyacetic Acid	Triothylaging	Triothylbonzone	Triothylonotatramico	Trimethylamine (water solution)		Trimethylchlorosilane	Tris(azididinyl)phosphine Oxide	Uranium Peroxide	Uranyl Nitrate	Uranyl Sulfate	Vaniding Outed of London	Vanadyl Sulfate	Vinol Chloride	Vinyl Fluoride (inhibited)	Vinylidenechloride (inhibited)	Vi Trichloweriles	Vinyl Iffeniorositane	Aylenol (2-4,)	Zinc Arconsto	The Sichtonate			
		;	CHRIS	3			TAP	Z Z	38.	12	TPH T	17.0	<u> </u>	11.		ž		THE	2	ag S	CAN	ERS.	101	×	5	¥	Ϋ́	1	23	717	740	78.7			

			ULTED			CHATC	CHATCH TEST						T WE	9 10		
		,	:-	干	POLYCARB	2	BUTYL	71.		PENETRATION	MOLLE		DECON	8	TEST	
		У	s	<u>.</u> w				SA Pr. DUYC	×	TEST	_		Œ	5.1	PERFORM	ANCE
CHDIS			0	<u>.</u>	WE16HT	ME16	는 도	PROPER	<u>.</u>	ppm or mg*	*	NO 30	CYCLES	LES	RATIN	9
3000	CHEMICAL NAME	<u>د</u>		د	CHANGE	CIANGE (x)	 ت	CHANGE	ш	:		TEST	-	80		
		w				<u> </u>		2		4 4	20 hr		3 hr	2 hr	POLYCAR-	BUTYL
				<u>"</u>	3 hr 20 hr 3 hr 20 hr Tensile	3 hr 2	0 hr	ensile	Tear	_1_					BUMIE	
			-			_								,		
		,			-				,	,	0.1	,	,		PASS (R)	FAIL (T)
ZCF	Zinc Chloride	n (<u>-</u>	= =				,	,	,	,	,	,	,	PASS (R)	PASS (R)
ZCR	Zinc Chromate	7.						1	,						9	ş
NOZ	Zinc Cyanide	Λ.	-	_ `				2	· •		:	•	ı	,	PASS (T)	FAIL (T)
ZPF	Zinc Fluoborate	. v	1 0	¥ ¥	+0.1 +0.2	, O	φ 0.00	-3.6	-4.9	,	80	PASS		(PASS (T)	PASS (T)
747	לושכ נוחחנות	•	+	1	+	-:	+	-;	+	 		-	1	!	(1/3340	DACC (T)
5	7 7inc Silicofluoride	S	s	¥ =	0.1 +0.2	-0.1 +0.3	£0.3	+2.5	±8.1						PASS (1)	FA11 (T)
7 7	Zirconium Potassium Fluoride	s	s	<u> </u>	0.0 7.0	9	÷0.3		13.6	'	,	1				FAIL (T)
ZCT	Zirconium Tetrachloride	s	1	<u></u>	,	SEB S		HLGK1C	a						1	
		_	4	4	-	-		- ∤	_	7	T			7		i

LEGEND

NO TINE CONTROLL IN CONTROL

Liquid
Solid and Soluble
Gas
Slightly Soluble
Insoluble
Insoluble
Dash indicates no testing performed
Water reactive, pertains to parent compound, literature data
Performance rating based on literature review
Performance rating based on laboratory tests of parent compound or water reaction product
Performance rating based on inability to decontaminate
No data

C-20

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